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Title: Investigation on the influence of spray-drying technology on the quality of Sicilian Nero d'Avola wines

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Corresponding Author: Dr. andrea salvo,

Corresponding Author's Institution: university of messina

First Author: Giuseppe Avellone

Order of Authors: Giuseppe Avellone; andrea salvo; Rosaria Costa; Emanuele Saija; David Bongiorno; Vita Di Stefano; Giorgio Calabrese; Giacomo Dugo

Abstract: The purpose of the present work was to find a correlation between microencapsulation technology applied to wines and resulting quality of the wine itself in terms of volatile composition and phenolic profile. To this aim, samples of Nero d'Avola wines produced in Sicily (Italy) were investigated in order to: i) elucidate the aromatic composition by means of HS-SPME coupled with GC-MS; ii) assess the polyphenolic content by UHPLC mass spectrometry; iii) compare the results obtained from both the screenings with those relative to the same wines that had previously been subjected to spray-drying. The results showed a marked reductionThe results here obtained evidenced a marked reduction of odour active compounds in microencapsulated wines, after resolubilization in water/ethanol; when considering the total amount of volatiles, a twofold reduction was observed. Conversely, the qualitative analysis of phenolic compounds and anthocyanin-derived pigments showed no influence of the spray-drying process on these functional constituents.



Dr. Andrea Salvo

Dip. Biomorf
(Scienze Biomediche,
Odontoiatriche, e delle Immagini
Morfologiche e Funzionali)
University of Messina
Viale Annunziata
98168 Messina
Tel. +39-090-3503996
e-mail: costar@unime.it

Editorial Office

FOOD CHEMISTRY

Messina, July 14, 2017

Dear Editor,

Please find attached the revised manuscript titled "**Investigation on the influence of spray-drying technology on the quality of Sicilian Nero d'Avola wines**" after second revision.

All the changes advised by editor have been completed. I hope that the manuscript is now suitable for publication in Food Chemistry.

Kind regards,

Dr. Andrea Salvo

A handwritten signature in blue ink that reads "Andrea Salvo".

FINAL EDITORIAL COMMENTS

Line 31:to the same wines that had previously been subjected to spray-drying. The results showed a marked reduction....

Line 53: Spray-drying has been applied in the food industry for many decades,....

Line 61:the advantages derived from spray-drying are numerous:...

Line 73:destined for food consumption.

Line 87: On the other hand, the wine aroma cannot be neglected as a fundamental parameter....

Line 220:the volatile composition was in good agreement....

Line 223:esters, were found to be odour active.

Line 253:fiber, which was more selective towards alcohols.....

Line 262: ...terpenoids, sulfur-containing, and others.....

Line 263: The content of volatiles was lower....

Line 276: ...wines are presented in Figures....

Line 284: The results led to the conclusion that.....did not affect....

Line 321: The results showed a marked reduction....

Table 1: Round values to at most 3 significant figures (e.g. 59.6 rather than 59.592; 1.89 rather than 1.885 etc.

AUTHORS:

All the modifications have been carried out and highlighted in red.

***Highlights (for review)**

1. The aromatic and phenolic fractions of spray dried wines were investigated
2. Wines under investigation were from red grapes “Nero d’Avola”, produced in Sicily.
3. The microencapsulation process affected the volatile composition of wines.

1 **Investigation on the influence of spray-drying technology on the quality of Sicilian Nero**
2 **d'Avola wines**

3

4 Giuseppe Avellone^a, Andrea Salvo^{b,c*}, Rosaria Costa^{b,c*}, Emanuele Saija^{b,c}, David
5 Bongiorno^a, Vita Di Stefano^a, Giorgio Calabrese^d, Giacomo Dugo^{a,b}.

6

7 ^aDipartimento di Scienze e Tecnologie Biologiche Chimiche e Farmaceutiche (STEBICEF),
8 Università di Palermo, Viale delle Scienze, Parco d'Orleans II, 90128 Palermo, Italy

9 ^bDipartimento di Scienze Biomediche, Odontoiatriche, e delle Immagini Morfologiche e
10 Funzionali (Biomorf), University of Messina, Viale Annunziata, 98168 Messina, Italy

11 ^cScience4Life s.r.l., a spin-off of the University of Messina, Messina, Italy

12 ^dDipartimento di Scienze Agrarie e Forestali, Universita' degli Studi di Torino, Via Verdi 8,
13 10124 Torino, Italy

14

15 *Corresponding author: Dipartimento Biomorf c/o Scienze Veterinarie, University of Messina,
16 Viale Annunziata, 98168 Messina (Italy); e-mail address (dr. Andrea Salvo): asalvo@unime.it;

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24 **Abstract:**

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26 technology applied to wines and the resulting quality of the wine itself in terms of volatile
27 composition and phenolic profile. To this aim, samples of Nero d'Avola wines produced in
28 Sicily (Italy) were investigated in order to: i) elucidate the aromatic composition by means of
29 HS-SPME coupled with GC-MS; ii) assess the polyphenolic content by UHPLC mass
30 spectrometry; iii) compare the results obtained from both the screenings with those relative to
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35 qualitative analysis of phenolic compounds and anthocyanin-derived pigments showed no
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45 Keywords: Spray-drying; wine; Nero d'Avola; phenolic compounds; aroma; SPME;
46 anthocyanins.

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49 **1. Introduction**

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51 Spray-drying falls within the group of microencapsulation technologies, including spray-
52 chilling, fluidized-bed coating, extruding, lyophilisation, coacervation, among others (Desai
53 and Park, 2005; Nedovic et al., 2011). **Spray-drying has been applied in the food industry for**
54 **many decades**, basically due to its cheapness, flexibility, robustness, efficiency. Based on a
55 simple definition, “encapsulation” is a technique which entraps particles (usually bioactive
56 compounds) within a wall material, working as a “shell” or “matrix”. The products of such a
57 technological process are microcapsules with diameters comprised in the range of μm – mm,
58 generally containing active ingredients. More specifically, spray-drying is a technological
59 process where a liquid is atomized through a hot gas (air or nitrogen) current, becoming
60 consequently a powder (Gharsallaoui et al., 2007). Mainly because of the water removal from
61 food commodities, **the advantages derived from spray-drying are numerous**: i) reduction of
62 microbiological decay events; ii) instantaneous solubilisation of spray-dried products
63 (improved product’s handling); iii) decrease of transport costs due to consistent volume
64 reduction of powdered products (“green” feature of the technology); iv) protection of the core
65 material against environmental factors (i.e. moisture, light, oxygen); v) improvement of
66 organoleptic properties of a food (e.g. masking bitterness of an ingredient by coating it with a
67 wall material having a pleasant taste); vi) elimination of cross reactions between more
68 ingredients. A key role in a successful spray-drying procedure is played by the wall material
69 chosen as encapsulating agent; the latter must be able to protect the capsule content, to be

70 stable over time and to avoid interaction with the outer environment. Encapsulation
71 technologies are utilized not only in food industry, but also in other fields (e.g. dried
72 detergents reconstituted upon use). A variety of synthetic polymers is available as wall
73 material; however, this list is definitely restricted when the spray-dried product is destined **for**
74 food consumption. Commonly, carbohydrates (starches, syrup solids, maltodextrins, pectins),
75 gums (gum Arabic, mesquite gum) or milk proteins are employed as wall material
76 (Gharsallaoui et al., 2007).

77 Red wines are suitable matrices for spray-drying, since they are mainly constituted of water,
78 ethanol and bioactive molecules, such as polyphenols. The beneficial effects exerted by
79 moderate consumption of red wine are today well recognized (Di Majo et al., 2008; Dugo et
80 al., 2009). “Nero d’Avola” is one of the most valuable grape varieties of the Italian
81 production. The name refers to the municipality of Avola, in Sicily, where this specific variety
82 was originally selected by vine growers. Nero d’Avola grapes grow easily in a dry and hot
83 environment; grapes strongly recall blackberries, both for the look and the taste. In general,
84 the taste of Nero d’Avola wine ranges from a full-bodied and black-fruit note to an elegant and
85 red cherry-like note, with very little or no aging at all (Nesto and Di Savino, 2013). The
86 phenolic composition of Nero d’Avola wines has been extensively investigated over the past
87 decade (Agozzino et al., 2015; La Torre et al., 2005; La Torre et al., 2014; Dugo et al., 2006).
88 On the other hand, **the wine aroma cannot be neglected as a fundamental parameter** for its
89 identification, evaluation and traceability, fact testified by a long list of scientific reports on
90 this topic (Gonzalez-Barreiro et al., 2015). Numerous reports on the aromatic fraction of wines
91 are based on the use of headspace techniques, in particular of solid-phase microextraction
92 (SPME), mainly because of its simplicity, low cost, effectiveness, sensitivity and selectivity

93 (Costa, 2014; Dugo et al., 2014; Panighel and Flamini, 2014). From an overview of literature
94 data, it is evident that the aroma of Nero d'Avola wines has been scarcely investigated
95 (Cincotta et al., 2015; Verzera et al., 2016; Esti and Tamborra, 2006; Papucci et al., 1999).
96 Against this background, in the present work samples of Nero d'Avola wines produced in
97 Sicily (Italy) were investigated in order to: i) elucidate the aromatic composition by means of
98 HS-SPME coupled with GC-MS; ii) assess the polyphenolic content by UHPLC-HESI-
99 Orbitrap mass spectrometry; iii) compare the results obtained from both the screenings with
100 those relative to the same wines, but preliminarily subjected to spray-drying processing. The
101 purpose was basically to determine if and how microencapsulation affects the quality of Nero
102 d'Avola wine as concerns its volatile composition and phenolic profile.

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105 **2. Materials and methods**

106 *2.1. Samples and chemicals*

107 Commercial Nero d'Avola wines (geographical indication) were from the brand "Conte di
108 Matarocco, Terre Siciliane" and produced by Cantine Paolini (Sicily, Italy). Wines were
109 purchased at local retailers.

110 Maltodextrin (dextrose equivalent 13.0-17.0) and 1-Hexan-d₁₃-ol were supplied by Sigma-
111 Aldrich (Milan, Italy). All solvents, unless specified, were from Merck (Darmstadt, Germany).
112 Acetonitrile, acetone and methanol (LC-MS grade) were purchased from Biosolve B.V.
113 (Valkenswaard, The Netherlands). Acetic acid (100% purity) and formic acid (98-100%
114 purity) were from VWR International B.V. (Roden, The Netherlands). PTFE syringe filters
115 (0.45 µm pore size) were supplied by Sigma-Aldrich. Gallic acid, caffeic acid, quercetin, (+)-

116 catechin, epicatechin, cyanidin-3-O-glucoside and myricetin were all purchased from Sigma-
117 Aldrich.

118

119

120 *2.2. Spray-drying*

121 For spray-drying procedures, a Mini Spray Dryer B-290 (Büchi, Cornaredo, Italy), was
122 exploited. A 200 mL aliquot of wine (12% v/v ethanol) was added with 40 g of maltodextrin
123 in a screw capped conical flask, and homogenized for 15 min ca, at room temperature (19-
124 20°C) until complete dissolution. Inlet and outlet temperatures (nitrogen) were 105 and 65°C,
125 respectively; feed flow rate was set at 18% of the maximum tolerated by the instrument.

126 Drying rate was approximately of 5 mL of wine per 1 min. The yield was estimated as 82%
127 w/v ca.

128

129 *2.3. Solid-Phase Microextraction*

130 For SPME extraction, four different fiber coatings were used:

131 divinylbenzene/carboxen/polydimethylsiloxane (DVB/Car/PDMS, 50/30 µm),
132 carbowax/divinylbenzene (CW/DVB, 70 µm), carboxen/polydimethylsiloxane (Car/PDMS, 85
133 µm), polydimethylsiloxane (PDMS, 100 µm). All SPME fibers were provided by Sigma-
134 Aldrich/Supelco (PA, USA).

135 In order to assess the best extraction time for each fiber several preliminary tests at increasing
136 times (5, 10, 15 and 20 min) were evaluated (data not showed), it was determined that 15 min
137 was suitable to obtain equilibrium and to reproduce the extraction procedure.

138 Spray-dried wines were resolubilized by dissolving 2 g of powdered wine in 10 mL of an
139 ethanol/water (12:88, v/v) solution.
140 4 mL of wine, whether untreated or resolubilized, were added with 0.6 g of NaCl and 10 µL of
141 a 200 ppm solution of internal standard (1-hexan-d₁₃-ol) and placed in a 8 mL amber glass
142 headspace vial, with pierceable silicone rubber septum coated with PTFE film. Wine samples
143 were pre-conditioned at 35 °C for 30 min and under agitation (250 rpm); successively
144 extraction took place by fiber exposure at the same temperature and for a 15 min period. Once
145 the extraction was complete, the fiber was withdrawn from the vial and inserted in the hot
146 (250°C) GC injector. A sample of powdered wine was analyzed apart for a rough qualitative
147 screening, by applying the same SPME conditions above reported, with the exception of fiber
148 exposure time, which was prolonged until 72 hours. In order to avoid carry over effects or
149 artifact formation, blank runs were carried out every three analyses, whereas fibers were
150 suitably cleaned up into a GC injector between consecutive analyses.

151

152 *2.4. GC-MS*

153 A Focus GC- DSQ II gas chromatograph/mass spectrometer (Thermo, CA, USA) equipped
154 with a 30 m × 0.25 mm i.d. × 0.25 µm film thickness ZB-WAX column (Phenomenex, CA,
155 USA) was used. The oven temperature program was from 40°C (3 min) at 10°C/min to 250°C,
156 hold 2 min. Gas flow rate (He) was 0.8 mL/min. Injection took place in splitless mode (3 min)
157 and at a temperature of 250°C. Data were acquired in the electron impact (EI) mode with an
158 ionisation energy of 70 eV, using full scan ion monitoring for a mass range 35-400 m/z.

159 Component assignment was based on computer matching with the WILEY 7 and NIST 02
160 mass spectral libraries; and on comparison with data retrieved from literature (Cincotta et al.,
161 2015; Verzera et al., 2016; Esti and Tamborra, 2006; Papucci et al., 1999).

162 The relative amounts of volatiles (semiquantitative analysis) were obtained by multiplying the
163 area ratio of target compound/internal standard by the concentration ($\mu\text{g/L}$) of the internal
164 standard.

165 *2.5. UHPLC-HESI-MS*

166 For identification of polyphenols (phenolic acids and anthocyanin-derived pigments) in wines,
167 UHPLC-HESI-MS methodology was exploited.

168 The UHPLC-HESI-MS system was a UHPLC (Dionex UltiMate®3000 Rapid Separation LC)
169 system by Thermo Fischer Scientific equipped with an autosampler, a diode array detector,
170 and controlled by *Chromeleon* 7.2 software, by Thermo Fisher (Bremen, DE) and Dionex
171 Softron GmbH (Germering, DE). The UHPLC system was coupled to a Orbitrap mass
172 spectrometer instrument (Q Exactive) (Thermo Scientific, Germany), equipped with heated
173 electrospray (HESI) ion source. Different conditions for identification of phenolic compounds
174 and anthocyanins were applied. Electrospray conditions for analysis in negative ion mode of
175 gallic acid, caffeic acid, quercetin, myricetin, catechin, epicatechin included: sheath gas flow
176 rate 35 (arbitrary units); auxiliary gas unit flow rate 4 (arbitrary units); sweep gas flow rate 7
177 (arbitrary units); spray voltage 3.5 kV; S lens RF level 30; capillary temperature 250 °C;
178 auxiliary gas heater temperature 250 °C. UV detection was performed at 254 and 278 nm.
179 Positive electrospray conditions for the determination of anthocyanins were optimized by
180 infusion of a solution of cyanidin-3-O-glucoside, and were the same as above reported for

181 phenolic compounds, with the exception of: sheath gas flow rate 30 (arbitrary units); sweep
182 gas flow rate 0 (arbitrary units); spray voltage 3.2 kV. UV detection was set at 520 nm.
183 In both cases, the UHPLC column was a Phenomenex Luna C18 (2) 50×1mm, 2.5 μ m. The
184 column temperature was set at 25 °C and the injection volume at 1.0 μ L. Mobile phase
185 composition: formic acid/water 2% v/v (eluent A), formic acid/acetonitrile 2% v/v (eluent B),
186 at a flow rate of 50 μ L · min⁻¹. The gradient was: 0 – 1 min, 3% B; 1 - 10 min, linear increase
187 to 15% B; 10 - 22 min, linear increase to 25% B; 22 - 28 min, linear increase to 50% B; 28 –
188 30 min, hold 50% B; 30 – 31 min, linear decrease 3% B; 31 – 33 min, hold 3% B.
189 The MS was operated in electrospray negative and positive mode and the analyses were
190 conducted in two acquisition modes: Full-Scan (positive and negative mode) and SIM. The
191 resolution power in full scan was 70,000 FWHM (at m/z 200) and the scan range was 100-800
192 m/z. Scan rate was 2 scan · s⁻¹ and the automatic gain control (AGC) target was set at 1e 5 ions
193 for a maximum injection time of 200 ms. For targeted SIM analyses, with a 15 s time window,
194 a mass inclusion list containing expected retention times of target phenolic analytes was built
195 and applied. The resolution power was 70,000 FWHM (at m/z 200) and the isolation window
196 was 1.2 m/z.
197 Data were analyzed with Qual Browser Xcalibur 3.0 (Thermo Fisher Scientific) and
198 identification of individual phenolic compounds and anthocyanins was greatly supported by
199 compound's accurate mass and retention time (if a reference standard was available), as
200 reported in **tables S1 and S2**.
201 Prior to injection, samples of untreated wines were filtered through PTFE 0.45 μ m syringe
202 filters. Aliquots of 2 g of spray dried wines were dissolved in 10 mL of a 88:12 (v/v)
203 H₂O/EtOH solution, and filtered as well.

204

205 **3. Results and Discussion**

206 *3.1. Aromatic fraction*

207 **Figures S1 and S2** show the HS-SPME-GC-MS chromatograms of untreated and spray-dried
208 Nero d'Avola wine samples, each extracted with four different fibers. Almost 100 different
209 volatile compounds were determined in total, distributed among the different types of samples
210 investigated. In order to have a comprehensive view of the whole volatile fraction, all the
211 SPME fibers tested were taken in consideration in data handling. **Table 1** reports quantitative
212 results for samples of Nero d'Avola wines, either untreated or resolubilized after spray-drying
213 treatment. As expected, in several cases a specific fiber showed higher selectivity toward an
214 analyte compared to another one. For instance, all the fibers were successful in the extraction
215 of compound nr. 8 (2,3-butanedione), whereas no traces of compound nr. 43 (1-octen-3-ol)
216 could be determined, unless using a CW/DVB fiber. However, more than a half volatiles were
217 likewise isolated by the all four fibers. One of the experiments carried out consisted of the
218 SPME fiber exposure directly to the headspace of spray-dried and powdered wine. Headspace
219 preconditioning (no fiber exposure) lasted 72 hours, at 35°C, followed by 15 min of fiber
220 exposure: around 75 volatiles could be detected in total, at consistently lower amounts
221 compared to spray-dried and resolubilized wines (data not shown). In general, the volatile
222 composition was in good agreement with previous reports on the same type of wine (Verzera
223 et al., 2016). Aroma, both in terms of intensity and complexity, is the most important factor in
224 the evaluation of wine quality. In the last years, numerous studies have unequivocally
225 demonstrated that the aromatic label of a wine derives from a combination of different
226 aromatic notes, more than from a dominant component (San-Juan et al., 2011). In the process
227 of flavour formation, chemical complexity is a key element, being directly correlated to the

variety of odorants, namely different olfactory characters and thresholds. Following an already established procedure, in order to identify among all volatiles those possessing a real olfactory impact, odour activity values were measured for each compound. The odour activity value (OAV) is obtained by dividing the amount of an analyte by its odour threshold. Compounds were considered “olfactorily active” when their OAVs were higher than 0.5. As can be seen from **table 1**, where odour thresholds retrieved from literature have been reported, eight components, mainly belonging to the group of esters, **were found** to be odour active. When considering these specific compounds, the observation of their amount in neat and in spray-dried wines evidenced a dramatic reduction, attributable to the microencapsulation process, in the range of 48.8 – 99.1% (median = 92.8%). In some cases, the reduction reached 100% (i.e. compound nr. 7). For some compounds (e.g. compounds nrs. 28, 55 and 70), detection occurred only, or at higher levels, in spray-dried wines, presumably due to either phenomena of co-elution or displacement effects exerted by multicoated SPME fibers. However, the amounts of volatiles in such cases were quite negligible.

In general, when the total volatile composition was taken into account, PDMS and CW/DVB fibers showed the highest extraction capability (see **figure 1**). Almost superimposable were the extraction performances registered for DVB/Car/PDMS and Car/PDMS fibers. Regardless of the fiber used, the total amount (ppb) of volatiles resulted to be lower for spray-dried (and resolubilized) wine samples; this finding denotes once again a partial loss of components consequent to the spray-drying process. However, this loss of volatiles didn't compromise to a great extent the volatile composition of spray-dried wines, which reported, on average, an amount of compounds lower by a 1.8 factor.

250 Figures 2 and 3 show quantitative results organized by group-type. In order to better
251 understand the different selectivities reported by the SPME fibers, the predominant
252 component, namely ethanol, was subtracted from quantitative plots. In neat wines, alcohols
253 and esters constituted the predominant fractions extracted by all the fibers, with the exception
254 of the CW/DVB fiber, which was more selective towards alcohols (57.6 vs. 31.4%,
255 respectively). DVB/Car/PDMS, Car/PDMS and PDMS fibers led to the isolation of 46% ca. of
256 alcohols and 47-50% ca. of esters. The PDMS fiber was the most selective toward an
257 important group of aroma compounds, i.e. ethyl esters of fatty acids, such as ethyl octanoate,
258 ethyl-9-decanoate and ethyl decanoate, produced enzymatically during yeast fermentation and
259 from ethanolysis of acyl-CoA. These esters usually give mature fruit flavour nuances and are
260 responsible for the fruity and floral sensory properties of wine. Among less represented classes
261 of volatiles, acids were selectively extracted at higher percentages by CW/DVB fiber (8.2%)
262 followed by PDMS (4.8%), Car/PDMS (2.1%) and DVB/Car/PDMS (1.6%). Minor classes of
263 compounds, namely aldehydes, ketones, terpenoids, sulphur-containing, and others, were
264 evenly distributed among the various fibers. The content of volatiles was lower in spray-dried
265 wines, fibers' selectivities described so far showed a similar trend in resolubilized samples.
266 Therefore, even in the extraction of this type of samples, the CW/DVB fiber extracted the
267 highest amount of acids, whereas the other three fiber coatings showed similar selectivity.
268 Precision of SPME-GC-MS method was evaluated by measurement of RSD% relative to three
269 replicates for each sample and preliminary tests to be analyzed: values obtained were in the
270 range 0.5-7.6%, with an average RSD% of 3.4%.
271
272 3.2. Phenolic content

273 A targeted qualitative screening of the polyphenolic fraction was carried out by means of
274 UHPLC-HESI-MS analysis. Measured masses of target analytes have been reported in **tables**
275 **S1** and **S2**.

276 Comparison of the phenolic profiles of untreated and spray-dried wines **are presented** in
277 **figures 4 and 5**. The elution profiles of anthocyanins are shown in **figures S3 and S4**. Selected
278 ion monitoring allowed to achieve the identification of different anthocyanins and phenolic
279 derived products in both the types of samples investigated. As can be seen from figures,
280 delphinidin-3-O-glucoside, cyanidin-3-O-glucoside, petunidin-3-O-glucoside, peonidin-3-O-
281 glucoside, malvidin-3-O-glucoside, delphinidin-(6-acetyl)-3-glucoside, malvidin-(6-acetyl)-3-
282 glucoside and malvidin-(6-coumaroyl)-3-glucoside, caffeic acid, gallic acid, catechin,
283 epicatechin, myricetin and quercetin, were detected in the two types of samples and
284 compared. The results led to the **conclusion** that the microencapsulation process **did not** affect
285 the qualitative composition of the polyphenolic fraction. Although only a rough screening was
286 carried out in this study, by comparing the signal intensities of target analytes, it might be
287 supposed that the amounts of phenolics are quite similar in both neat and spray-dried samples.
288 All the polyphenols determined in this study were previously reported for Nero d'Avola wines
289 (La Torre et al., 2005; La Torre et al., 2014; Dugo et al., 2006).
290 For LC-MS method development, all the parameters were optimized based on repetitive
291 injections of reference standards. Solutions of standards were prepared by taking into account
292 possible matrix effects, therefore by dissolving standards of phenolic compounds into aqueous
293 formic acid (pH 3)/methanol (90:10). Each standard was injected 5 times consecutively, at
294 one concentration level (namely 1 ppm) and repeatability assessed through RSD% (on average
295 $\leq 2.5\%$). Also, the mobile phase composition for LC gradient elution was suitably varied both

296 in terms of type of solvents and their concentration. The mobile phase composition was indeed
297 a crucial parameter affecting not only the chromatographic resolution (retention, selectivity
298 and efficiency), but also the quantitative transfer of target analytes to the mass spectrometer
299 through the ESI interface. In general, all the parameters were tuned in order to get the best
300 signal from the MS detector. For instance, the use of 2% formic acid at the specific gradient
301 program above reported resulted to be fundamental for improving the parameters of efficiency
302 (peak shape) and sensitivity.

303 As shown in **figures 4** and **5**, the elution time of the polyphenols investigated was about 30
304 min. As above mentioned, **table S1** lists nominal and measured masses of target phenolic
305 compounds. The acquisition of mass spectra further confirmed the identification process
306 carried out by standard co-injection. The choice of the ESI interface in negative ionization
307 mode (for phenolic compounds) and in positive ionization mode (for anthocyanin-derived
308 pigments) was dictated by the proven higher ionization efficiency and enhanced signals
309 (Kruve et al., 2014; Di Stefano et al., 2017). All the polyphenols shown in **figures 4** and **5**
310 were determined at levels well above their S/N ratios.

311

312

313 **4. Conclusions**

314 Microencapsulation techniques are becoming widespread in food and beverage industry, in
315 consideration of their numerous advantages, some of them being preservation from microbial
316 and environmental contamination, elimination of interferences, concentration of bioactive
317 ingredients. In this study, red wines from the cultivar Nero d'Avola were subjected to spray-
318 drying technology and successively analyzed by GC/MS and UHPLC-HESI-MS, for the

319 assessment of the volatile and phenolic composition. The purpose of the study was basically to
320 evaluate if the spray-drying process somehow affects the important components of aroma and
321 phenolic and anthocyanin-derived compounds. The results showed a marked reduction of
322 odour active compounds in microencapsulated wines, after resolubilization in water/ethanol;
323 when considering the total amount of volatiles a twofold reduction was observed. Conversely,
324 the qualitative analysis of polyphenols showed no influence of the spray-drying process on
325 these functional constituents, thus confirming the efficiency of microencapsulation in the
326 isolation and concentration of bioactive molecules. The results here presented give a hint for
327 the development of a sustainable wine product, namely a “wine powder”, which could be
328 exported worldwide with a considerable cost reduction due to the elimination of the liquid
329 volume. Prior to selling/consumption, the wine powder can be safely reconstituted as normal
330 wine through the addition of a hydroalcoholic solution. The final product, as shown in this
331 report, might have a slightly poorer aroma, but would certainly remain a wine of acceptable
332 quality.

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- 433 **Figure 1** – Extraction performance of the four SPME fiber coatings expressed as total amount
434 of volatiles (µg/L).

435
436 **Figure 2** – Groups of volatile compounds (ethanol excluded) extracted from untreated wine
437 samples by means of four different SPME fiber coatings.
438
439 **Figure 3** – Groups of volatiles (ethanol excluded) extracted from spray-dried wine by using
440 four different fiber coatings.
441
442 **Figure 4** – UHPLC-HESI-MS(-) SIM chromatogram profiles of phenolic compounds
443 determined in samples of untreated wines.
444
445 **Figure 5** - UHPLC-HESI-MS(-) SIM profiles of phenolic compounds present in resolubilized
446 spray-dried wine samples.
447
448 **Figure S1** – HS-SPME-GC-MS (TIC) fingerprints of untreated wine samples extracted by
449 means of four different fiber coatings.
450
451 **Figure S2** - HS-SPME-GC-MS (TIC) chromatograms of spray-dried wines after
452 resolubilization, extracted by different fibers.
453
454 **Figure S3** – UHPLC-HESI-MS (+) SIM chromatograms of anthocyanins identified in samples
455 of untreated wines.
456
457 **Figure S4** - UHPLC-HESI-MS (+) SIM profiles of anthocyanins identified in resolubilized
458 spray-dried wine samples.
459
460
461

Table 1 – HS-SPME-GC-MS composition of wine samples. Compounds in bold are “odour active” (see text). Values are means of triplicate analyses. n.f. = not found. N/A = not available.

Nr.	Compound	Odour threshold* (µg/L)	DVB/Car/PDMS		CW/DVB		Car/PDMS		PDMS	
			Neat wine (µg/L)	Spray-dried wine (µg/L)						
1	Acetaldehyde	120	0.08	0.03	0.11	0.40	0.21	0.16	0.05	0.07
2	Dimethyl sulfide	1	0.02	n.f.	n.f.	n.f.	0.01	n.f.	n.f.	n.f.
3	Ethyl formate	N/A	0.02	0.02	n.f.	n.f.	0.01	0.07	0.15	0.08
4	Ethyl acetate	7,500	12.2	0.50	4.23	0.23	19.8	1.29	37.7	1.65
5	Ethanol	100,000	59.6	47.6	336.0	219.0	59.3	66.7	210.0	316.0
6	Ethyl propanoate	10	0.32	n.f.	n.f.	n.f.	0.21	n.f.	0.68	n.f.
7	Ethyl isobutyrate	0.1	1.01	n.f.	0.38	n.f.	0.45	n.f.	2.45	n.f.
8	2,3-Butanedione	0.86	0.04	0.04	0.29	0.25	0.12	0.11	0.08	0.17
9	Ethyl butyrate	1,600	0.88	n.f.	0.31	n.f.	0.49	n.f.	2.13	n.f.
10	1-Propanol	50,000	0.25	0.17	0.17	0.13	0.35	0.13	0.31	0.16
11	Succinic acid, butyl propyl ester	N/A	0.04	n.f.	n.f.	n.f.	0.04	n.f.	n.f.	n.f.
12	Ethyl 2-methylbutyrate	18	0.39	n.f.	0.13	n.f.	0.23	n.f.	0.88	n.f.
13	Ethyl isovalerate	3	0.60	0.02	0.15	n.f.	0.28	<0.01	1.36	n.f.
14	2-Methylbutyl acetate	5	n.f.	0.02	n.f.	n.f.	n.f.	0.02	n.f.	n.f.
15	Isobutanol	40,000	1.31	0.04	1.88	0.07	0.79	0.06	4.42	0.22
16	2,2,6-Trimethyl-6-vinyltetrahydropyran	N/A	0.09	n.f.	n.f.	n.f.	0.041	n.f.	n.f.	n.f.
17	Isoamyl acetate	30	3.03	0.08	0.73	0.06	1.60	0.06	5.54	0.34
18	Ethyl valerate	5	0.02	0.11	n.f.	0.17	n.f.	0.02	0.04	0.15
19	1-Butanol	150,000	0.06	n.f.	n.f.	n.f.	0.07	n.f.	0.17	n.f.
20	Sulfur dioxide	N/A	n.f.	n.f.	0.61	2.23	n.f.	n.f.	n.f.	n.f.
21	Limonene	200	n.f.	0.07	n.d.	0.09	n.f.	n.f.	0.13	0.16
22	Isoamyl alcohol	30,000	37.3	2.44	63.5	3.64	26.9	2.24	93.9	5.50
23	Ethyl hexanoate	14	7.36	0.13	2.57	0.15	3.51	0.03	12.2	0.39
24	2,4-Hexadienoic acid, ethyl ester (2E,4E)-	N/A	n.f.	0.01	n.f.	0.06	n.f.	n.f.	n.f.	n.f.

25	(1E,2E)-Dipropenylcyclobutane	N/A	n.f.	n.f.	n.f.	n.f.	0.20	n.f.	n.f.	n.f.
26	Isoamyl butyrate	N/A	n.f.	n.f.	n.f.	n.f.	n.f.	n.f.	0.06	n.f.
27	Hexyl acetate	1,500	0.05	n.f.	0.01	n.f.	n.f.	n.f.	0.07	n.f.
28	Octanal	0.7	n.f.	0.06	0.13	0.05	n.f.	0.07	0.49	0.12
29	Acetoin	800	0.51	n.f.	0.39	0.26	0.48	n.f.	n.f.	n.f.
30	3-Hexenoic acid, ethyl ester	N/A	0.03	n.f.	0.01	n.f.	n.f.	n.f.	0.05	n.f.
31	4-Methyl-1-pentanol	5,000	0.01	n.f.	0.05	n.f.	0.02	n.f.	0.03	n.f.
32	2-Heptanol	300	n.f.	n.f.	0.01	n.f.	0.02	n.f.	n.f.	n.f.
33	3-Methyl-1-pentanol	2.2	0.06	n.f.	0.07	n.f.	0.04	n.f.	0.09	n.f.
34	Ethyl heptanoate	2.2	0.05	n.f.	0.02	n.f.	0.01	n.f.	0.15	n.f.
35	Ethyl lactate	14,000	1.85	0.59	6.17	2.44	1.96	0.91	2.84	0.71
36	1-Hexanol	8,000	1.35	0.15	1.16	0.09	1.52	0.20	1.42	0.11
37	(3E)-Hexen-1-ol	400	n.f.	n.f.	0.04	n.f.	n.f.	0.06	0.02	n.f.
38	(3Z)-Hexen-1-ol	400	n.f.	n.f.	0.15	n.f.	n.f.	n.f.	0.01	n.f.
39	Methyl octanoate	200	0.11	0.02	0.07	n.f.	0.03	n.f.	0.13	n.f.
40	Nonanal	1	0.08	0.20	0.05	0.35	0.02	n.f.	0.35	0.95
41	Carbon disulfide	N/A	0.02	n.f.	n.f.	n.f.	0.02	n.f.	n.f.	n.f.
42	Ethyl octanoate	5	11.06	0.52	10.5	1.29	1.99	0.17	23.4	2.57
43	1-Octen-3-ol	1	n.f.	n.f.	0.02	n.f.	n.f.	n.f.	n.f.	n.f.
44	1-Heptanol	300	0.22	0.03	0.15	0.04	0.12	n.f.	0.20	0.06
45	Isoamyl hexanoate	N/A	n.f.	n.f.	0.02	n.f.	n.f.	n.f.	n.f.	n.f.
46	Furfural	14,100	0.35	0.18	0.10	0.14	0.24	0.15	0.08	n.f.
47	Acetic acid	N/A	n.f.	n.f.	8.31	9.62	1.04	0.31	n.f.	n.f.
48	2-Propyl-1-pentanol	N/A	0.93	0.85	1.51	4.29	0.46	0.71	1.12	0.99
49	3-Ethyl-4-methylpentanol	N/A	0.11	n.f.	0.12	n.f.	0.08	n.f.	0.18	n.f.
50	Ethyl nonanoate	N/A	n.f.	0.09	n.f.	n.f.	n.f.	n.f.	0.26	n.f.
51	2,3-Butanediol	120,000	0.09	n.f.	0.89	1.86	0.07	0.04	0.20	n.f.
52	Linalool	25	0.26	0.07	0.24	0.21	0.15	0.05	0.29	n.f.
53	n-Octyl formate	N/A	0.21	0.02	0.23	0.09	0.08	n.f.	0.29	n.f.

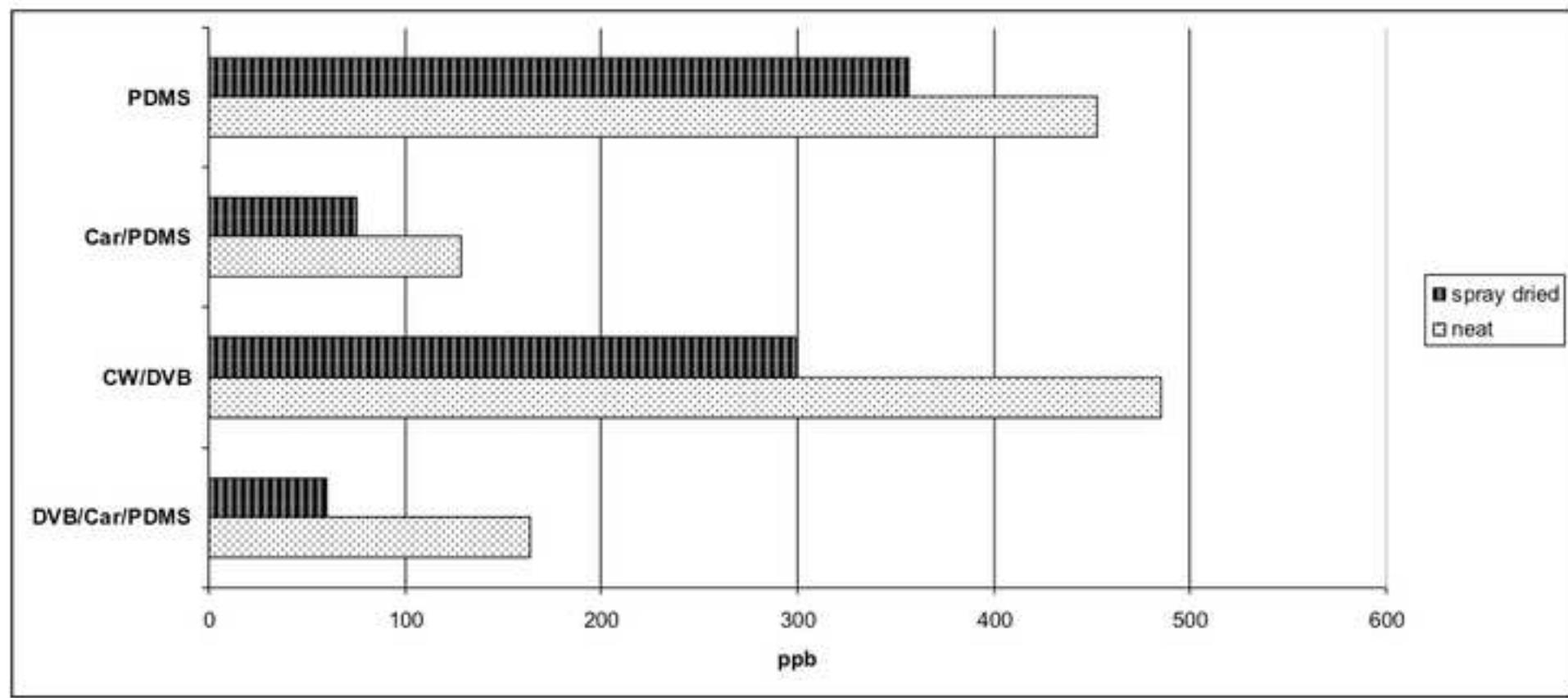
54	Isoamyl lactate	200	0.28	0.02	0.44	n.f.	0.18	0.02	0.42	n.f.
55	β-Ionone	0.09	n.f.	n.f.	0.24	0.60	0.02	0.02	0.03	n.f.
56	Hexadecane	N/A	n.f.	0.03	n.f.	n.f.	0.06	0.02	n.f.	0.47
57	Propylene Glycol	N/A	0.06	n.f.	0.46	0.32	n.f.	n.f.	n.f.	n.f.
58	n-Nonylcyclohexane	N/A	n.f.	0.03	n.f.	0.10	n.f.	n.f.	n.f.	0.29
59	Terpinen-4-ol	110	n.f.	n.f.	n.f.	n.f.	n.f.	0.01	n.f.	n.f.
60	Diethylene Glycol ethyl ether	N/A	n.f.	n.f.	0.13	0.19	n.f.	n.f.	n.f.	n.f.
61	2-Furancarboxylic acid, ethyl ester	N/A	0.10	<0.01	0.11	0.01	0.02	n.f.	0.09	n.f.
62	Ethyl decanoate	200	1.01	0.03	1.52	1.49	0.22	0.04	2.61	2.09
63	Dihydro-2(3H)-furanone	50,000	0.05	0.02	0.29	0.42	0.04	0.05	0.04	0.06
64	Butanoic acid	240	0.29	0.05	0.48	0.25	0.10	0.04	0.42	0.17
65	Furfuryl alcohol	2,000	n.f.	n.f.	n.f.	n.f.	0.05	0.065	n.f.	n.f.
66	Diethyl succinate	200,000	11.2	2.08	17.7	11.2	2.35	0.59	20.2	6.45
67	Ethyl dec-(9E)-enoate	100	n.f.	n.f.	n.f.	n.f.	n.f.	n.f.	0.05	n.f.
68	2-Methylhexanoic acid	N/A	n.f.	0.03	0.69	0.41	0.09	0.04	0.14	n.f.
69	α-Terpineol	330	0.36	0.10	0.33	n.f.	0.09	0.03	0.34	0.07
70	Ethyl decanoate	200	n.f.	0.01	n.f.	0.05	n.f.	n.f.	n.f.	n.f.
71	3-(Methylthio)-1-propanol	1,000	0.06	0.01	0.49	0.25	0.04	0.01	0.15	n.f.
72	Diethyl glutarate	N/A	0.04	n.f.	0.13	0.04	0.01	n.f.	0.05	n.f.
73	Methyl salicylate	100	n.f.	n.f.	0.13	n.f.	n.f.	n.f.	n.f.	n.f.
74	Phenylethyl acetate	250	0.41	0.11	0.49	0.07	0.08	n.f.	0.46	0.27
75	β-Damascenone	0.05	0.22	0.05	0.56	0.29	0.05	0.01	0.53	0.13
76	Ethyl dodecanoate	1,500	n.f.	0.21	n.f.	1.25	n.f.	0.01	n.f.	1.64
77	Hexanoic acid	200,000	0.19	0.04	0.56	0.61	0.11	0.03	0.51	0.17
78	Benzyl alcohol	200,000	0.28	0.09	0.85	1.20	0.09	0.04	0.36	0.20
79	Butanedioic acid, ethyl-3-methylbutyl ester	N/A	0.25	0.02	0.46	0.14	0.04	0.01	0.51	0.11
80	Phenethyl alcohol	14,000	4.37	1.31	10.1	18.1	1.26	0.59	7.76	3.59
81	1-Dodecanol	1,000	0.07	0.66	0.09	6.43	0.05	0.05	0.12	3.79
82	Diethylene glycol	N/A	0.15	0.05	0.07	0.07	0.05	0.02	n.f.	0.04

83	Ethyl tetradecanoate	2,000	n.f.	n.f.	n.f.	0.78	n.f.	n.f.	n.f.	0.61
84	Octanoic acid	500	0.57	0.06	1.19	0.86	0.10	0.02	1.70	0.41
85	4-Hexyl-2,5-dihydro-2,5-dioxo-3-furanacetic acid	N/A	0.08	0.03	0.24	0.34	n.f.	n.f.	0.20	0.33
86	Hexadecanal	4,500	n.f.	n.f.	0.20	n.f.	n.f.	n.f.	n.f.	n.f.
87	1-hexadecanol	N/A	n.f.	n.f.	n.f.	0.44	n.f.	n.f.	n.f.	n.f.
88	4-Ethylphenol	440	n.f.	n.f.	n.f.	n.f.	<0.01	n.f.	n.f.	n.f.
89	Nonanoic acid	3,000	n.f.	n.f.	n.f.	n.f.	0.01	<0.01	0.29	0.14
90	Ethyl palmitate	1,500	0.04	0.06	0.10	1.61	n.f.	n.f.	0.28	1.09
91	Decanoic acid	1,000	0.08	n.f.	0.09	n.f.	0.01	n.f.	0.27	n.f.
92	2,4-di-t-Butylphenol	200	1.58	0.55	3.19	4.31	0.51	0.25	3.55	2.86
93	Dodecanoic acid	10,000	n.f.	n.f.	n.f.	n.f.	n.f.	n.f.	0.24	0.16
94	Tetradecanoic acid	10,000	0.14	0.03	n.f.	n.f.	n.f.	n.f.	n.f.	n.f.
95	Octadecanoic acid	20,000	n.f.	n.f.	n.f.	n.f.	n.f.	n.f.	2.60	0.32
96	9-Octadecenoic acid	N/A	0.29	n.f.	0.71	0.05	n.f.	0.02	5.27	0.95
TOTAL			164.0	59.8	485.0	299.0	129.0	75.6	453.0	357.0

*Values retrieved from references Tao & Zhang, 2010; Verzera et al., 2016; L.J. van Gemert, 2011.

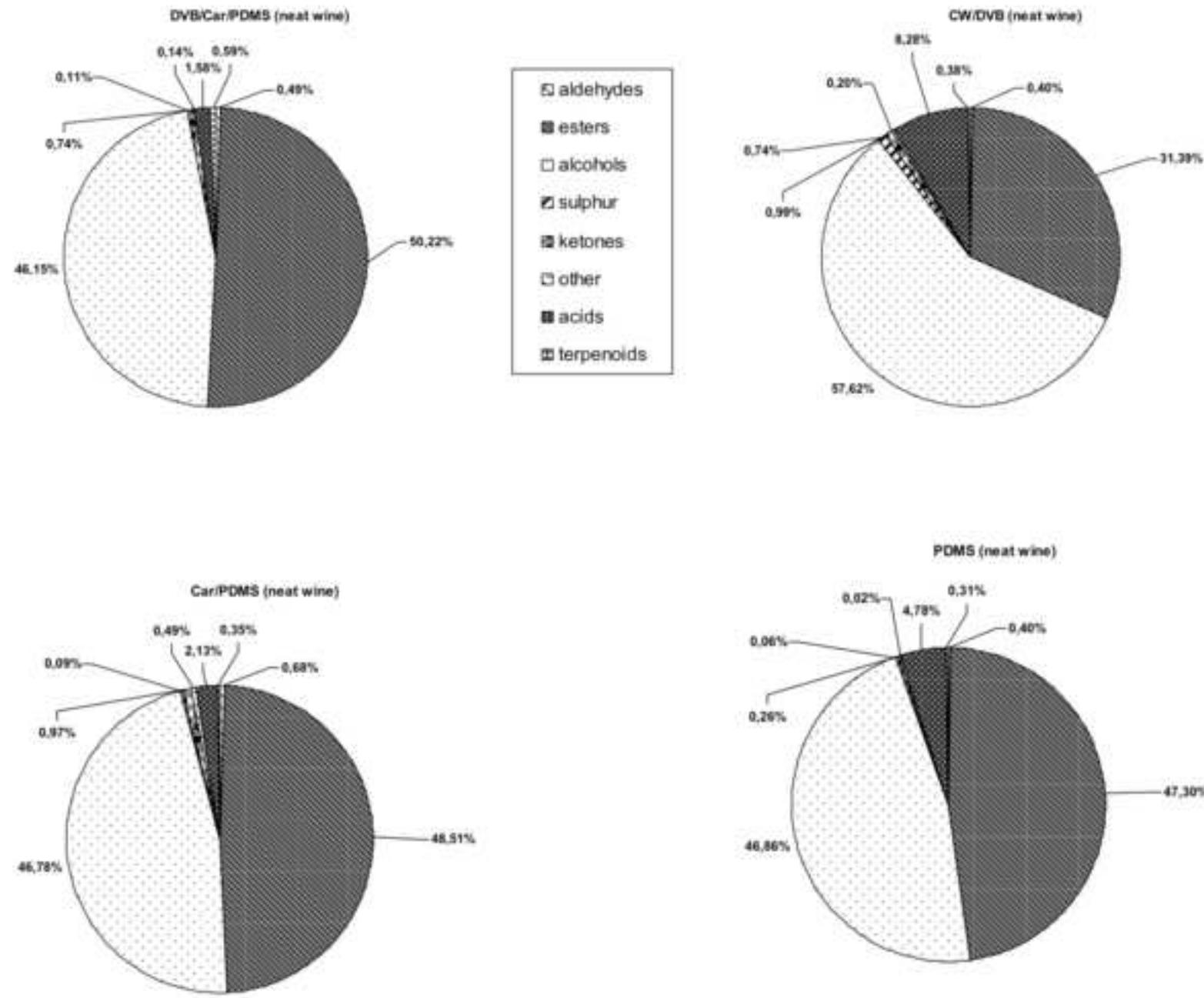
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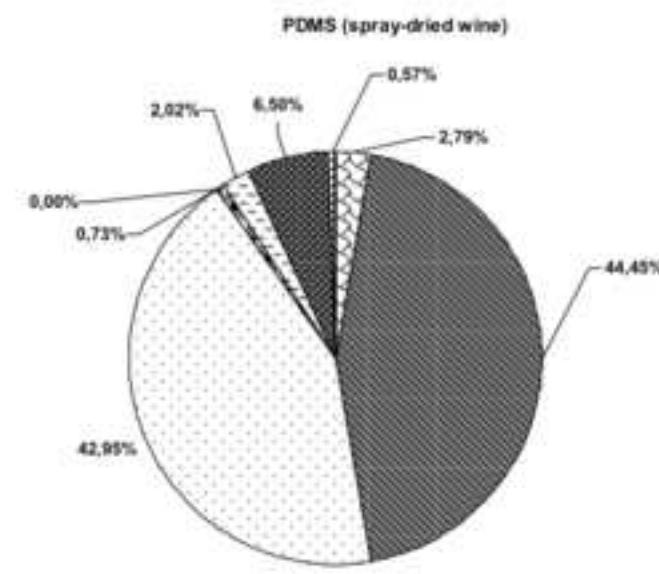
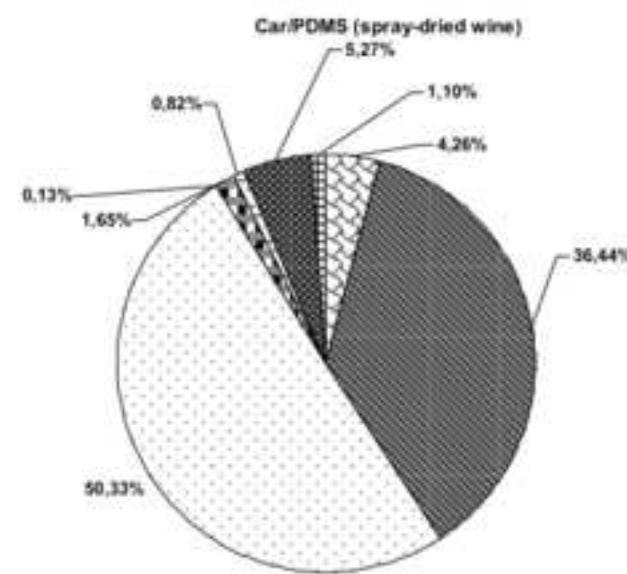
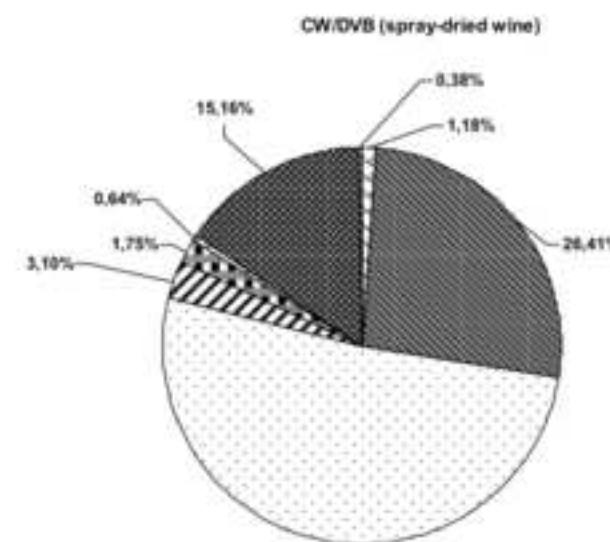
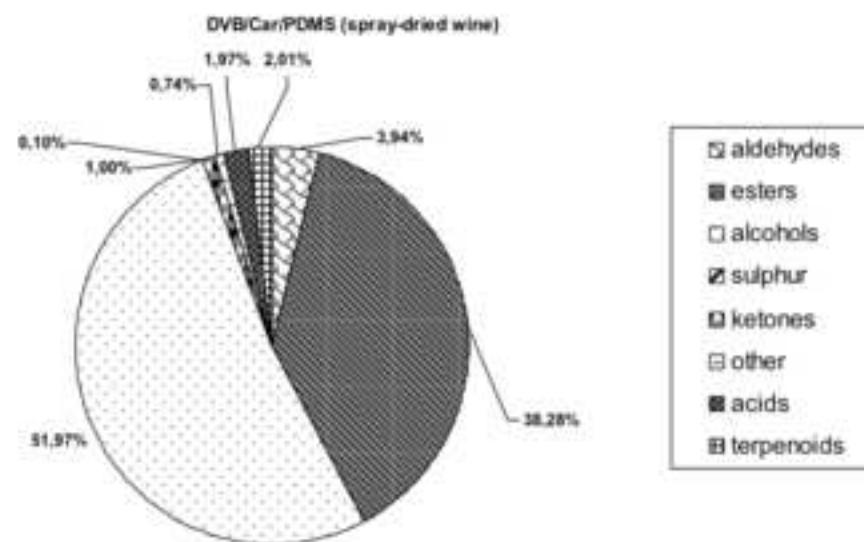


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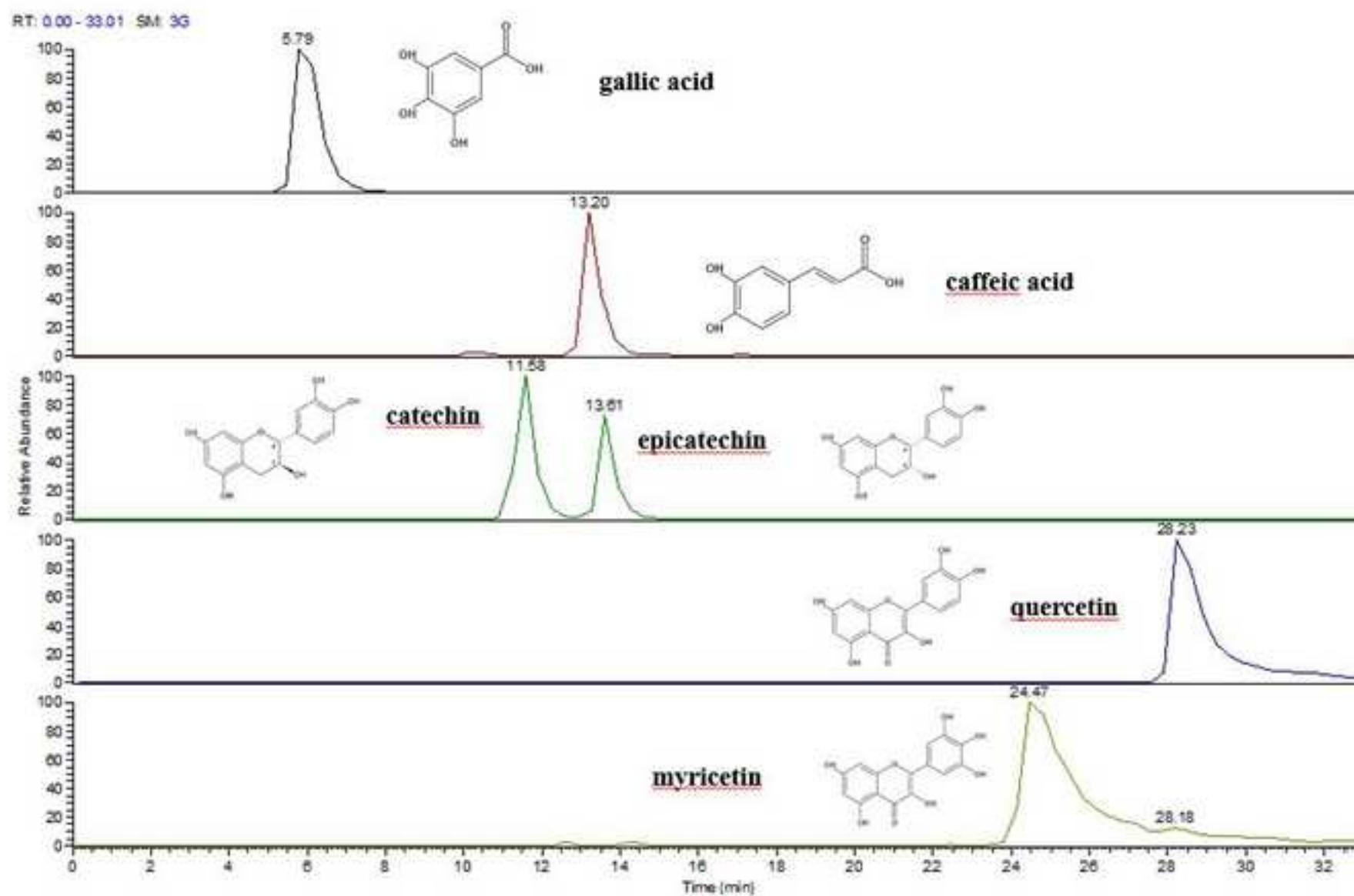
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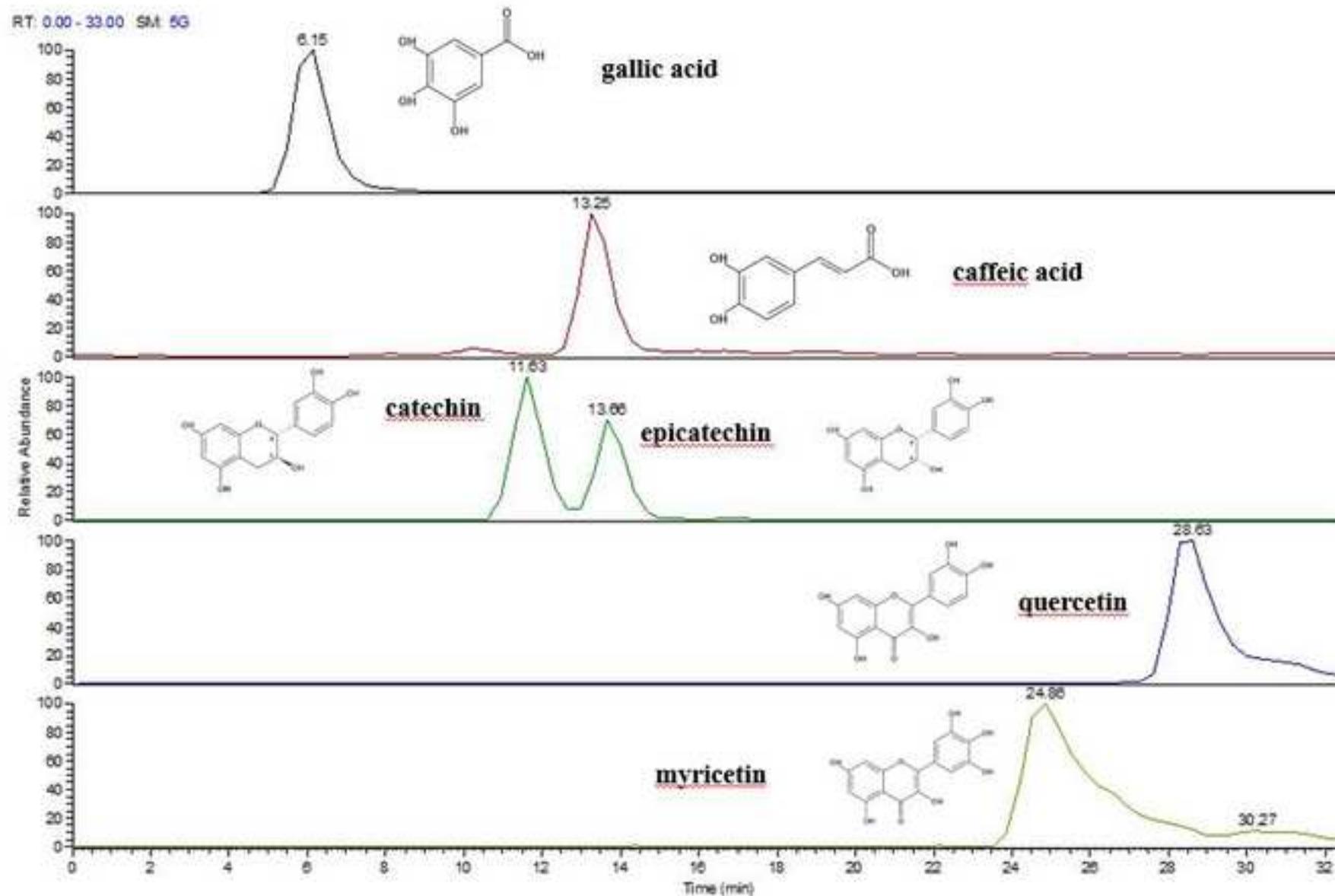
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