



SAPIENZA
UNIVERSITÀ DI ROMA

Ph.D.
CHEMICAL SCIENCES
XXX Cycle

Ph.D. Dissertation

**Application of analytical methods for characterization
of waste and recovered solvents**

Coordinator

Prof. Osvaldo Lanzalunga

Tutor

Prof. Roberta Curini

Candidate

Enrico Mattocci

693394

INDEX

PREFACE	4
INTRODUCTION	
1. Compliance with the regulations requirements	4
2. Life Cycle Assessment (LCA)	6
3. Analytical Procedures for quality control in pharmaceutical industry	8
4. Solvent Recovery	9
5. Analytical Methods for determination of VOC	10
6. Factors that can affect the analysis	12
7. Method EPA 8015D	13
8. Method EPA 8260C	14
9. Scope of the work	17
MATERIALS AND METHODS	
1. Analytical reagents and suppliers	19
2. Instruments	19
3. Operative Procedures for analysis of samples of “Solvent Recovery”	22
4. Validation of the analytical method	22
5. Experiments of Recovery	23
6. Qualitative analysis by GC-MS	23
7. Quantitative analysis by GC-MS with direct injection	23
RESULTS AND DISCUSSION	
1. Qualitative analysis by GC-MS and Quantitative analysis with GC-FID.	25
2. Calibration curves and procedures to control analytical data	26
3. LOD and LOQ	28
4. Experiments of repeatability on Certified Matrix	29
5. Recovery of the method applied to six target compounds	33
6. Analytical Method applied for the determination of 24 compounds	34
7. Reproducibility Experiments	39
8. Experiments on matrix spiked	42
CONCLUSIONS	43

APPENDIX	
A1. Calibration curves	50
A2. Determination of LOQ and LOD	54
A3. Experiments of repeatability on Certified Matrix	59
A4. Experimental Data of Recovery % of six target compounds	65
A5. Experimental Data of Recovery % of 24 compounds	66
A6. Reproducibility Tests	70
A7. Experiments on Laboratory Control Sample, Matrix Spike (MS) and Matrix Spike Duplicate (MSD)	72
BIBLIOGRAPHY	74
ACNOWLEDGMENTS	76

PREFACE

The topic of this dissertation is the research of an analytical method for the determination of volatile organic compounds, in environmental matrix. This research was performed in a pharmaceutical manufacturing plant. This pharmaceutical plant uses a large amount of organic compounds for their manufacturing processes and the implementation of a program of waste management is a key aspect of this plant.

An example of waste management is the process of "Solvent Recovery" which is fundamental to recover organic solvents to be re-used and to control the environmental impact of the processes. The process of the Solvent Recovery following the first use in the manufacturing process has been the scope of the study.

This project is focalised on defining an analytical method for the determination of several types of volatile organic compounds and it is applied to control the processes of solvent recovery.

"Solvent Recovery" consists in the distillation and purification of organic solvents, with the objective to produce pure solvent or mix of solvents that are then re-used in the manufacturing industrial processes.

It is a strategy of this pharmaceutical industry for a sustainable environment and the solvent recovery represents a growing business in waste management. In fact, environmental protection is often an important prerogative of any new process and new industrial technologies.

Official analytical methods for analysis of solvents, in environmental matrix, specified the compounds and conditions to carry out an analysis. In this work, the method EPA 8015D is evaluated as preferential method for the analysis. This method is enhanced for a great number of solvents.

Purpose of this work has been the definition of rapid and reliable analytical method for the determination of solvents in the recovered solvents.

Organic solvents take part of the pharmaceutical processes as the medium of synthesis reaction end extraction.

The determination of a solvent, in an environmental matrix, has many analytical difficulties which are related with the high volatility and sampling technique for this kind of matrix.

INTRODUCTION

2. Compliance with the regulations requirements

Legislative Decree n° 3 April 2006, n° 152 promotes a high quality of human life through the management of environmental conditions and of natural sources. Any human organization must guarantee the protection of the environment and natural ecosystem through action of prevention of pollution and by evaluation of the environmental impact of production processes. The regulations promote sustainable development to ensure that the activities that satisfy all the needs of current generations do not compromise the quality of life of future generation.

The environmental assessment of plans, programs and projects became important to ensure that human activity is compatible with sustainable development: protection of human health, better quality of life, maintain the species and maintain of the capacity of reproduction of the ecosystem.

The installation of new manufacturing plants and or substantial modification of operations need the request for Integrated Environmental Authorization that contain information about: the installation, type and scope of the

installation, raw and auxiliary materials, substances and energy used by plant, the source of emissions of the plant. The control and the prevention of emission is important for a sustainable activity.

The regulation confirms the importance of the waste management: the objective is the reduction of waste disposal through action of valorisation of the waste. The waste is not only something that is become useless, but also it could have a second life: the waste became starting material for the production of secondary raw material¹.

The Directive 2008/1/EC on Integrated Pollution Prevention and Control (IPPC Directive) achieves integrated prevention to control pollution and it lays about the adoption of specific measure to reduce emissions level in the air, in the water and in the land. The general principles of this directive are based on the safety for the environment: all member states shall take the necessary measures to provide that all the installations operate in accordance with the following requirements:

- use of the best technical solution to prevent pollution
- avoid waste production, and recovery action for wastes and reducing of emissions and environmental impact when wastes are produced
- efficient use of energy
- take necessary measures to prevent accidents and limit their consequences.

In the Annex III of the Directive 2008/1/EC there is a list of substances that can affect the safety of the environment:

- Quality of the air: sulphur dioxide and other sulphur compounds; oxides of nitrogen and other nitrogen compounds; carbon monoxide; volatile organic compounds (VOC), metals and other compounds, dust, asbestos (suspended particulates, fibres); chlorine and its compounds, fluorine and its compounds, arsenic and its compounds, cyanides; substances and preparation which have proved to have carcinogenic or mutagenic properties or properties which may affect reproduction via the air
- Quality of the water: organohalogen compounds and substances which may form such compounds in the aquatic environment; organophosphorous compounds; organotin compounds, substances and preparation which have proved to have carcinogenic or mutagenic properties or properties which may affect reproduction via the aquatic environment, persistent hydrocarbons and persistent and bioaccumulable organic toxic substances; cyanides; metals and their compounds; arsenic and its compounds; biocides and plant health products, materials in suspension, substances which contribute to eutrophication (in particular nitrates and phosphate); substances which have an unfavourable influence on the oxygen balance (and can be measured using parameters such as BOD, COD, etc)²

To persecute the scope of the safety of human health and of the environment is necessary a correct use of chemical substances in all the steps of the process. The Classification, Labelling and Packaging (CLP) Regulation (EC No 1272/2008) ensures a high level of protection of health and environment and it describes the rules for classification and labelling of substances and mixtures. CLP requires manufacturers, importers or downstream users of substances or mixtures to classify label and package their hazardous chemicals appropriately before placing them on the market and use for many kinds of applications. CLP is very

¹Decreto Legislativo 3 aprile 2006, n. 152

²Directive 2008/1/EC on Integrated Pollution Prevention and Control (IPPC Directive)

determinant to communicate the hazardous of a substance or a mixture. When relevant information, such as toxicological data, are available for a substance or a mixture, the CLP assigns a certain hazard class and category. After the classification, the hazards of a substance or a mixture must be communicated to other actors of supply chain³.

CLP Regulation No 1272/2008	Example of hazard statement	Example of precautionary statement
	Heating may cause an explosion	Keep away from heat/sparks/open flames/hot surfaces. – no smoking
	Heating may cause a fire	Keep only in original container
	May intensify fire; oxidizer	Take any precaution to avoid mixing with combustibles
	Causes serious eye damage	Wear eye protection
	Toxic if swallowed	Do not eat, drink or smoke when using this product
	Toxic to the aquatic life with long lasting effects	Avoid release to the environment
	May cause allergy or asthma symptoms or breathing difficulties if inhaled	In case of inadequate ventilation wear respiratory protection
	May cause an allergic skin reaction	Contaminated work clothing should not be allowed out of the workplace
	Contains gas under pressure; may explode when heated	Protect from sunlight. Store in a well ventilated place

Figure 1: Pictograms for hazardous substances according with Classification, Labelling and Packaging (CLP)

2. Life Cycle Assessment (LCA)

Nowadays we live in the world affected by energy crisis, because of the dependence of many human activities by non-renewable energy source and their inefficient use. Biofuels have received a great attention in order to reduce the dependence on fossil fuels⁴. Studies of “Life Cycle Assessment (LCA)” demonstrate a positive approach of the reconversion of wastes in useful materials and substances⁵.

The LCA is an important tool of making decision to explore technical solutions to reduce the use of fossil fuels and to generate a low impact on environment.

³Regulation (EC) No 1272/2008 of the parliament and of the council of 16 December 2008 on classification, labelling and packaging of substances and mixtures, amending and repealing Directives 67/548/EEC and 1999/45/EC, and amending Regulation (EC) No 1907/2006, Official Journal of the European Union L 353/1

⁴PooyaAzadi, Robert Malina, Steven R.H. Barrett, Markus Kraft, The evolution of the biofuel science, *Renewable and Sustainable Energy Reviews* 76 (2017) 1479–1484

⁵ Augustine Quek , RajasekharBalasubramanian, Life Cycle Assessment of Energy and Energy Carriers from Waste Matter e A Review, *Journal of Cleaner Production* 79 (2014) 18 - 31

Sustainable Development of productive processes imply an approach of LCA of all steps of industrial production. Modern chemical industries evaluate how their activities affect the environment and nowadays the choice of “greener” processes and products is a key aspect to minimise environmental impact⁶.

“Life Cycle Assessment” is an approach to assess industrial system that begin with the evaluation of raw material originated from the earth and it ends when all materials come back to earth. LCA evaluates all stages of product’s life, which are interdependent, so this approach enables the estimation of cumulative environmental impact that came from all stages of the production.

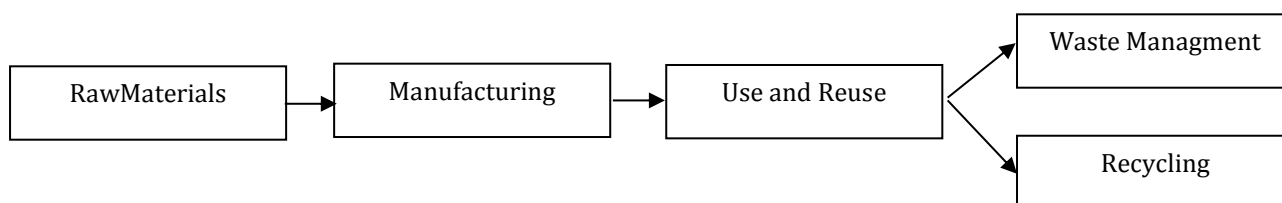


Figure 2: Steps of Life Cycle Assessment.

LCA helps to manage environmental impact underlying the importance of a correct planning of different material used in the processes: for example use and reuse of substances.

The LCA methodology includes the following steps (ISO 14040 2006):

- Goal definition: the study of the process
- Inventory analysis: identification and quantification of all energies and materials used and of all environmental emissions
- Impact assessment: involves the evaluation of human and ecological effects from the inventory.
- Interpretation: the evaluation of the data to make a decision on which process or product is environmental friendly and sustainable

LCA is applied by many kinds of industry: pharmaceutical, electronic, fuel manufacturer, ecc...⁷

Pharmaceutical industry uses a large quantity of solvent for the synthesis of active principles and, after their use, incineration of waste solvents is a common practice. The implementation of solvent recovery plant allows the minimization of emissions and the reduction of solvents deposal. Organic solvents are used as reaction medium for the synthesis of many complex molecules called “Active Pharmaceutical Ingredient (API)”. Solvents represent a very big portion of wastes produced by pharmaceutical industry: they are used for synthesis, purification and washing operation of API⁸.

⁶Adeel Tariq, Yuosre F. Badi, Waqas Tariq, Umair Saeed Bhutta, Drivers and consequences of green product and process innovation:A systematic review, conceptual framework, and future outlook, *Technology in Society* 51 (2017) 8-23

⁷M.A.Curran,Life Cycle Assessment: Principles and Practices, United States Environmental Protection Agency, National Risk Management Research Laboratory, Cincinnati, Ohio, 2006

⁸Michael J. Raymond, C. Stewart Slater and Mariano J. Savelski, LCA approach to the analysis of solvent waste issues in the pharmaceutical Industry, *GreenChem.*, 2010,12, 1826–1834

3. Analytical Procedures for quality control in pharmaceutical industry

Organic solvents are necessary for production processes of pharmaceutical products, but it is necessary to remove all solvents from the final products at the end of the process.

Analytical procedures used during quality control analysis of pharmaceutical products are based on the use of gas chromatography to separate all the components. The identification is made by Flame Ionization Detector (FID) or by Mass Spectrometry (MS). Many analytical methods are specific for a limited number of solvents and sample matrices.

It is necessary to study the analytical procedures for a large number of solvents and their residual determination. The solvents are necessary during the production processes as reaction medium and for purification operations, such as crystallization. The determination of residual solvents in pharmaceutical products is necessary, because of their toxicity and their hazardous for human health and the environmental impact. The solvents can also induce transformation of active substances, potentially losing their biological activity and causing the decomposition of the products.

Another important aspect is a forensic interest in the determination of contaminants of pharmaceutical products: the profile of contaminants could be used as a fingerprinting of manufacturer to avoid counterfeit of pharmaceuticals⁹.

A classification of solvents based on their toxicity is possible as follows¹⁰:

- Class I solvents: the use of these solvents should be avoided because they are carcinogens and very dangerous for the environment
- Class II solvents: the use of these solvents should be limited because they are non-genotoxic animal carcinogens, but they cause other irreversible toxicity such as neurotoxicity or teratogenicity.
- Class III solvents: they are solvents with low toxicity for human health.

⁹Analytical procedures for quality control of pharmaceuticals in terms of residual solvents content: Challenges and recent developments, Maciej Tankiewicz, Jacek Namiesnik, Wiesław Sawick, *Trends in Analytical chemistry* 80 (2016) 328–344

¹⁰ICH guideline Q3C (R6) on impurities: guideline for residual solvents EMA/CHMP/ICH/82260/2006

The most used solvents in pharmaceutical industry are summarized in the following Table 1:

Class of Solvents	Compounds
Alcohols	Methanol, ethanol, propanol, isopropanol, butanol, isobutanol, propylene glycol
Aliphatic Hydrocarbons	Hexane, cyclohexane, isooctane
Amides	Dimethylformamide
Amines	Pyridine
Aromatic hydrocarbons	Toluene, xylenes
Esters	Ethyl acetate, butyl acetate
Ethers	Ethyl ether, 1,4-dioxane, tetrahydrofuran, tert-butyl methyl ether, diisopropyl ether
Alogenated Solvents	Dichloromethane, chloroform, carbon tetrachloride, trichloroethane, ethylene bromide, tetrachloroethylene
Ketones	Acetone, methyl ethyl ketone, methyl isobutyl ketone, methyl isopropyl ketone
Nitriles	Acetonitrile
Sulfur containing	Dimethyl sulfoxide
Water	-

Table 1: The most used solvents in Pharmaceutical Industry

4. Solvent Recovery

The solvents are produced from petrochemicals feedstocks or by chemical processes and they are largely used in chemical plants. After their use the solvents are recycled or sent to disposal.

A typical operation of disposal of solvents is incineration: the combustion of waste solvents produces a large amount of exhaust gases and wastewater and it is necessary to treat wastewater to reduce their environmental impact.

The recovery of waste solvents is very important in the pharmaceutical, chemical and petrochemical industries. A large amount of production costs of pharmaceutical industry are imputed to downstream processing in which numerous solvents and mix of them are used to purification operations.

It is interesting for industry to find technical solutions that allow recovering solvents to be reused for their needs¹¹.

¹¹ Eduardo J., Cavanagh, Mariano J.Savelski, C.Stewart Slater, Optimization of environmental impact reduction and economic feasibility of *solventwasterecovery* using a new software tool, *Chemical Engineering Research and Design*, 92 (2014) 1942-1954

The presence of organic solvents in wastewater evidences an inefficient use of resources and this is a problem because of the hazardous for the human health and the environment.

In accordance with the general principle of Directive 2008/1/EC on Integrated Pollution Prevention and Control (IPPC Directive), chemical industries must adopt preventive measures and the best available techniques to recover and recycle wastes. The waste prevention means the adequacy of measures for the correct use of chemical substances before they became wastes and their re-use or recovery. Recycling indicates a procedure focused on reprocessing a material for the same or different kind of industrial process. Recovery indicates a technical procedure for the regeneration of a material with the same characteristics.

Chemical industries that use a large amount of solvents, such as pharmaceutical industry, have to adopt measures for the control of the presence of solvents in wastewater with analytical procedures. The development of analytical procedures is strategical to verify the adopted measures of prevention and the compliance with regulations requirements. In chemical plants, the recovery of solvents, with added value, is preferred compared to the disposal. The implementation of technical solution for recovering is the best strategy for industries that uses a large amount of solvents because of economic benefit that derives from recovery and the minimization of dangerous wastes and the emissions in the environment. The recovery of solvents from water and the wastewater depuration is the best solution to reduce environmental impact of industrial activities.

Solvent Recovery process is a technical challenge because of the difficulties of recovering high amounts of pure solvents and mix of solvents: it depends by the starting matrices that are different in solvent and water content; then many solvents generate azeotropes with difficulties in the separation by distillation.

A particular solution to separate solvents from aqueous matrix (Abejon R et al.) is the use of organic solvents resistant nanofiltration membranes. The separation of solutes by nanofiltration is based on the difference between their molecular weights and their molecular sizes: a good separation is obtained when there is a high difference in terms of molecular weights and sizes. The separation is very difficult when there is a very low difference; in this case, the separation by nanofiltration needs many steps¹².

Another example of innovative system to remove organic solvent is the operation with membranes (V. Garcia et al.) A quaternary system with water, sodium chloride, dichloromethane and n-butanol is processed by pervaporation and operation with hydrophobic membranes: the first membrane is made of polysiloxanepolymer with an active layer thickness of 0.01 mm, the second membrane is a dense membrane of 0.125 mm made of dimethyl and methyl vinyl siloxane copolymers¹³.

5. Analytical Methods for determination of VOC

Analytical testing requests for a sample of waste recovered solvents are usually: pH, density, water content (% w/w) and the concentrations of the different compounds.

These analyses are crucial to control the Solvent Recovery process to transform the waste solvent in a secondary raw material.

¹²Abejon R., Garea A., Irabien A., 2015, Organic solvent recovery and reuse in pharmaceutical purification processes by nanofiltration membrane cascades, *Chemical Engineering Transactions*, 43, 1057-1062 DOI: 10.3303/CET1543177

¹³VerónicaGarcía, Eva Pongrác, Paul S. Phillips, Riitta L. Keiski,From waste treatment to resource efficiency in the chemical industry: recovery of organic solvents from waters containing electrolytes by pervaporation, *Journal of Cleaner Production* 39 (2013) 146e153

Water content influences the prices of recovered solvents: samples with a high water percentage are less interesting in this business, while samples with a high solvents content are more interesting. The knowledge of the composition of the samples of waste solvents plays a pivotal role for the right decision-making, based on the performance of industrial process of solvent recovery.

Distillation is an operation able to separate different compounds based on the different volatility using a series of distillation columns with a particular set of temperature to perform the best and efficient separation. The presence of an azeotropic mixture contributes to make difficult the entire process. Industries have the need to control all the process through efficient analytical methods able to give a rapid response.

The development of gas chromatography is a determining factor for analysis of volatile organic compounds (VOC). The research and development of analytical methods to determinate the concentration of VOC in different environmental matrices have nowadays a growing interest because of the impact of these compounds to various environmental systems. Many effects of VOC are recognized: stratospheric ozone depletion, tropospheric photochemical ozone formation, toxic and carcinogenic human health effects, aggravate the global greenhouse effect and accumulation as pollutants in the environment.

Instrumental analysis is based on the use of gas chromatography coupled with different types of detectors: Flame Ionization Detector (FID); Electron Capture Detector (ECD), Photoionization Detector (PID) and Mass Spectrometry (MS).

There is a huge amount of interesting work for the determination of VOC in environmental matrices^{14 15 16 17}.

It's very interesting to know analytical methods able to detect a huge amount of compounds.

The determination of VOC in surface water and wastewater is carried out by Purge and Trap-Gas Chromatography-Mass Spectrometry method¹⁸. The investigation of VOC in surface water and wastewater is important to evaluate the pollution and give the opportunity to make correlation between human activities and environmental impact. The development of efficient analytical method could help to control environmental impact.

Analytical methods for determination of VOC take into account the range of concentration of interest. Dynamic headspace chromatography is based on the maximization of the contact between gas and the surface or the internal of the sample allowing the concentration of the compounds in the trap. Analytical methods based on the

¹⁴Ruth Barro, Jorge Regueiro, MaríaLlompert, Carmen Garcia-Jares, Analysis of industrial contaminants in indoor air: Part 1. Volatile organic compounds, carbonyl compounds, polycyclic aromatic hydrocarbons and polychlorinated biphenyls, *Journal of Chromatography A*, 1216 (2009) 540-566

¹⁵Jo Dewulf, Herman Van Langenhove, Anthropogenic volatile organic compounds in ambient air and natural waters: a review on recent developments of analytical methodology, performance and interpretation of field measurements, *Journal of Chromatography A*, 843 (1999) 163-177

¹⁶Kristof Demeestere, Jo Dewulf, Bavo De Witte, Herman Van Langenhove, Sample preparation for the analysis of volatile organic compounds in air and water matrices, *Journal of Chromatography A*, 1153 (2007) 130-144

¹⁷Martin Harper, Sorbent trapping of volatile organic compounds from air, *Journal of Chromatography A*, 885 (2000) 129-151

¹⁸Anastasia D. Nikolaou, Spyros K. Golfinopoulos, Maria N. Kostopoulou, George A. Kolokythas, Themistokles D. Lekkas, Determination of volatile organic compounds in surface waters and treated wastewater in Greece, *Water Research* 36 (2002) 2883-2890

use of Dynamic headspace chromatography offers the advantage of determination of very low concentration of VOC and they revealed successful to detect residual solvent in pharmaceutical preparation¹⁹.

6. Factors that can affect the analysis

A particular attention is regarded to the sampling methodologies based on the use of different sampling systems specific for different compounds: the research about adsorbent materials has nowadays a growing interest.

Analytical quality assurance for the analysis of VOC in environmental matrices is an important aspect: the compounds of interest are affected by several factors that could compromise the analysis: sampling methodology, transport conditions, maintenance, extraction and instrumental conditions.

For example, air sampling systems for VOC detection are manufactured with different sorbent matrices, such as: activated carbon, silica gel, polydimethylsiloxane.

The problem with air sampling is the absorption of water, which can affect the sampling; it is important to use material with a high specificity for the molecules of interest and able to avoid the absorption of the water.

The sampling of water for the determination of VOC is affected by different factors: transport and maintenance condition and the subsequent analysis is influenced by extraction methodologies.

The liquid-liquid extraction is ideal to extract VOC from water by organic solvent, but the disadvantage is in the correct choose of the solvent, different from analytes, and in recovery efficiency.

The development of dynamic headspace technique is ideal for VOC extraction from water samples: the extraction of compounds of interest is performed by purge and trap system. Bubbling a gas stream in the water samples, the inert gas enriches with volatile compounds, and then the enriched gas led through a sorbent matrix in which volatile compounds are retained. Subsequently the VOC are desorbed from sorbent matrix and they are injected in GC column. The major advantage of purge and trap is that a high percentage of VOC are successfully injected with a better analytical response.

To evaluate analytical performance of an analytical method, it is important to take in consideration some criteria associated with the quality of analytical data: Limit of detection, Limit of quantification, Reproducibility, Repeatability, Accuracy, Calibration curve, Specificity, Range of application²⁰.

Waste solvents samples are very complex matrices: they are heterogeneous solutions with solid and liquid phase.

Complex samples could be characterized by a biphasic liquid system with a solid residual. In these particular cases, the sampling is very problematic because of the difficulties to generate a representative sample from container with a content of thousands of litres of waste solvent.

The testing laboratory needs to complete the analysis in order to characterize the effective composition of the samples, but a waste solvent contains solid residuals, water, pharmaceutical waste, solvents mixture in one or two phases.

¹⁹Claudia Witschi, Erik Doelker, Residual solvent in pharmaceutical products: acceptable limits, influences on physicochemical properties, analytical methods and documented values, *European Journal of Pharamceutics and Biopharmaceutics* 43 (1997) 215-242

²⁰Jo Dewulf, Herman Van Langenhove, Anthropogenic volatile organic compounds in ambient air and natural waters: a review on recent developments of analytical methodology, performance and interpretation of field measurements, *Journal of Chromatography A*, 843 (1999) 163-177

The attention is concentrated on the determination of VOC content in the samples based on a specific request of the customers who are interested to agree regulations and to recover the maximum amount of solvents.

The difficulties associated with the analytical approach with this kind of samples are in the pre-treatment of the samples. For example, for a complex sample, that shows biphasic liquid system and solid, a sequence of steps is necessary: it is possible to quantify the solid by evaporation of solvent and to analyse separately the phases in the samples. In this case, the step of pre-treatment is a critical aspect for the expression of the correct analytical results.

7. Method EPA 8015D

The samples of waste solvents can be analysed with official method EPA 8015D. This method may be used to determine the concentration of volatile organic compounds (VOC) and semi-volatile organic compounds by gas chromatography. The following list shows the compounds that can be analysed by official method EPA 8015D:

- Acetone
- Acetonitrile
- Acrolein
- Acrylonitrile
- Allyl alcohol
- t-Amyl alcohol (TAA)
- t-Amyl ethyl ether (TAEE)
- t-Amyl methyl ether (TAME)
- Benzene
- t-Butyl alcohol (TBA)
- Crotonaldehyde
- Diethyl ether
- Diisopropyl ether (DIPE)
- Ethanol
- Ethyl acetate
- Ethyl Benzene
- Ethylene oxide
- Ethyl *tert*-butyl ether (ETBE)
- Isopropyl alcohol (2-Propanol)
- Methanol
- Methyl ethyl ketone (MEK, 2-Butanone)
- Methyl *tert*-butyl ether (MTBE)
- *N*-Nitroso-di-*n*-butylamine
- Paraldehyde
- 2-Pentanone
- 2-Picoline
- 1-Propanol (*n*-Propylalcohol)

- Propionitrile
- Pyridine
- Toluene
- *o*-Toluidine
- *o*-Xylene,
- *m*-Xylene,
- *p*-Xylene

This method is also for the quantitation of triethylamine and it finds application for the characterization of petroleum samples. The method allows the analysts to add other analytes, but it is important to demonstrate the performance of analytical method.

The flame ionization detector is a non-selective detector and it is necessary to consider the presence of interfering compounds. Samples with a large amount of different compounds can show problems of resolution. The confirmation of the different molecules is necessary and a GC-MS could be used to make a qualitative analysis.

Quality control procedure are necessary to demonstrate the best performance of the systems of analytical laboratory. The first control is the blank to demonstrate the absence of interfering compounds and to control instrumental signal of the blank; the control of calibration is necessary to validate the calibration curve and quantitative analysis; the matrix spike samples are necessary to control the recovery of the method²¹.

8. Method EPA 8260C

The Method EPA 8260C allows the determination of VOC by GC-MS for different kinds of environmental matrices. In the list below the principal compound of interest for this method²²:

- Acetone
- Acetonitrile
- Acrolein (Propenal)
- Acrylonitrile
- Allyl alcohol
- Allyl chloride
- *t*-Amyl ethyl ether (TAEE)
- *t*-Amyl methyl ether (TAME)
- Benzene
- Benzyl chloride
- Bis(2-chloroethyl)sulfide
- Bromoacetone
- Bromochloromethane
- Bromodichloromethane

²¹Method EPA 8015D, Non-halogenated organics using GC-FID, *Revision 4 June 2003*

²²Method 8260C, Volatile Organic Compounds by gas chromatography / Mass Spectrometry

- 4-Bromofluorobenzene (surr)
- Bromoform
- Bromomethane
- n-Butanol
- 2-Butanone (MEK)
- t-Butyl alcohol
- Carbon disulfide
- Carbon tetrachloride
- Chloral hydrate
- Chlorobenzene
- Chlorobenzene-d5 (IS)
- Chlorodibromomethane
- Chloroethane
- 2-Chloroethanol
- 2-Chloroethyl vinyl ether
- Chloroform
- Chloromethane
- Chloroprene
- Crotonaldehyde
- 1,2-Dibromo-3-chloropropane
- 1,2-Dibromoethane
- Dibromomethane
- 1,2-Dichlorobenzene
- 1,3-Dichlorobenzene
- 1,4-Dichlorobenzene
- 1,4-Dichlorobenzene-d4 (IS)
- cis-1,4-Dichloro-2-butene
- trans-1,4-Dichloro-2-butene
- Dichlorodifluoromethane
- 1,1-Dichloroethane
- 1,2-Dichloroethane
- 1,2-Dichloroethane-d4 (surr)
- 1,1-Dichloroethene
- trans-1,2-Dichloroethene
- 1,2-Dichloropropane
- 1,3-Dichloro-2-propanol
- cis-1,3-Dichloropropene
- trans-1,3-Dichloropropene
- 1,2,3,4-Diepoxybutane
- Diethyl ether

- Diisopropyl ether (DIPE)
- 1,4-Difluorobenzene (IS)
- 1,4-Dioxane
- Epichlorohydrin
- Ethanol
- Ethyl acetate
- Ethylbenzene
- Ethylene oxide
- Ethylmethacrylate
- Fluorobenzene (IS)
- Ethyl tert-butyl ether (ETBE)
- Hexachlorobutadiene
- Hexachloroethane
- 2-Hexanone
- Iodomethane
- Isobutyl alcohol
- Isopropylbenzene
- Malononitrile
- Methacrylonitrile
- Methanol
- Methylene chloride
- Methyl methacrylate
- 4-Methyl-2-pentanone (MIBK)
- Methyl tert-butyl ether (MTBE)
- Naphthalene
- Nitrobenzene
- 2-Nitropropane
- N-Nitroso-di-n-butylamine
- Paraldehyde
- Pentachloroethane
- 2-Pentanone
- 2-Picoline
- 1-Propanol
- 2-Propanol
- Propargyl alcohol
- β -Propiolactone
- Propionitrile (ethyl cyanide)
- n-Propylamine
- Pyridine
- Styrene

- 1,1,1,2-Tetrachloroethane
- 1,1,2,2-Tetrachloroethane
- Tetrachloroethene
- Toluene
- Toluene-d8 (surr)
- o-Toluidine
- 1,2,4-Trichlorobenzene
- 1,1,1-Trichloroethane
- 1,1,2-Trichloroethane
- Trichloroethene
- Trichlorofluoromethane
- 1,2,3-Trichloropropane
- Vinyl acetate
- Vinyl chloride
- o-Xylene
- m-Xylene
- p-Xylene

The application of GC-MS has the advantage of a higher instrumental sensibility allowing the simultaneous identification and quantification.

9. Scope of the work

Objective of this work was the development of an analytical method, in accordance with official method EPA 8015D, for the determination of Volatile Organic Compounds (VOC) in waste solvents and secondary raw materials. This analytical method is focused for the determination of high concentration of solvents.

This project was carried out inside a pharmaceutical plant in which a distillation plant is implemented for the recovery of waste solvents and their transformation in secondary raw materials: this is the goal of "Solvent Recovery".

This work shows the presentation and the statistical elaboration of results obtained following the iter of validation of the analytical method. The interest was focused on recovery data to evaluate the precision, accuracy and reproducibility of the analytical method.

The evaluation of *z-score*, recovery and the Analysis of Variance (ANOVA) played a pivotal role to demonstrate the validity of the purpose of this work.

The analytical method developed was focused for the determination of the following compounds:

- Methanol
- Ethanol
- Isopropanol
- Acetone
- Acetonitrile
- Methyl Acetate

- Dichloromethane
- Hexane
- 2-Butanone
- Ethyl Acetate
- Tetrahydrofuran
- Cyclohexane
- Isopropyl Acetate
- Heptane
- 1,4-Dioxane
- Cyclopentyl Methyl Ether
- Toluene
- Butyl Acetate
- N,N-Dimethylformamide
- 2-Butanol
- 1-Butanol
- Pyridine
- Diisopropyl ether
- Octane

These compounds were the most frequently requested by the customers because their diffusion in pharmaceutical processes. Their hazard for environment and for human health is known, so their recover and re-use is pivotal to reduce their impact.

The importance of this work derives from the needs to implement an analytical method able to investigate different compounds assuring an high quality of analytical data.

MATERIALS AND METHODS

1. Analytical reagents and suppliers

Methanol; Ethanol; Isopropanol; Acetone; Acetonitrile; Methyl acetate; Dichloromethane; Hexane; 2-Butanone; Ethyl acetate; Tetrahydrofuran; Cyclohexane; Isopropyl Acetate; Heptane; 1,4-Dioxane; Cyclopentyl methyl ether; Toluene; Butyl Acetate; N,N-Dimethylformamide; 2-Butanol; 1-Butanol; Pyridine; Diisopropylether; Octane are acquired from Sigma Aldrich. Pentane was purchased from Scharlab.

Certified matrix with a composition of Methanol, Ethanol, Isopropanol, Acetonitrile, Ethyl Acetate, Toluene was purchased from Restek.

Methanol with 0,0005% of water content purchased from Scharlab and Titrant for Karl Fischer without pyridine was used for determination of water by Karl Fischer's titrimetric method was purchased from Merck (Sigma Aldrich).

2. Instruments

A GC-FID-NPD 7820 A of Agilent with software Open Lab CDS Rev.C.01,07 was used for separation and quantification of analytes. This Instrument was called in this work "GC-FID_Principal".

The instrumental operative conditions are described as follow:

- GC Oven Temperature Setpoint On
- Initial Temperature 40 °C
- Hold Time 20 min
- Post Run 50 °C
- Program: Rate 15 °C/min; Final Temperature 250 °C; Hold Time 10 min
- Equilibration Time 1 min
- Max Temperature 300 °C
- Syringe Size 10 µL
- Injection Volume 1 µL
- Mode Split
- Heater On 300 °C
- Pressure On 7.3001 psi
- Gas Saver On 20 After 3 min mL/min
- Split Ratio 20 :1
- Split Flow 110 mL/min
- Column: Mega 15191 Vocol Fused Silica; diameter 530.00 µm; length 60.0 m; film thickness 3.00 µm; maximum temperature 300.0 °C; Flow 5.5 ml/min
- Make up gas: Helium
- Make up flow: 30.0 ml/min
- Hydrogen flow: 45.0 ml/min
- Air Flow: 400.0 ml/min

A GC-FID of Shimadzu with software GC Solution was used for separation and quantification of analytes. This Instrument was called in this work "GC-FID_Control".

The instrumental operative conditions are described as follow:

- GC Oven Temperature Setpoint On
- Initial Temperature 40 °C
- Hold Time 20 min
- Post Run 50 °C
- Program: Rate 15 °C/min; Final Temperature 250 °C; Hold Time 10 min
- Equilibration Time 1 min
- Max Temperature 300 °C
- Syringe Size 10 µL
- Injection Volume 1 µL
- Mode Split
- Heater On 300 °C
- Pressure On 7.3001 psi
- Gas Saver On 20 After 3 min mL/min
- Split Ratio 20 :1
- Split Flow 110 mL/min
- Column: Vocol+ BPX; length 85.0 m; diameter 530 µm; maximum temperature 300.0 °C; Flow 3.0 ml/min
- Make up gas: Helium
- Make up flow: 30.0 ml/min
- Hydrogen flow: 45.0 ml/min
- Air Flow: 400.0 ml/min

A GC-MS 5977 with headspace injector of Agilent with software Mass Hunter GC/MS Acquisition B.07.00.SP2.1654 equipped with a NIST library database was used for the identification of analytes in samples of solvent recovery.

The instrumental operative conditions are described as follow:

- GC Oven Temperature Setpoint On
- Initial Temperature 60 °C
- Hold Time 10 min
- Post Run 50 °C
- Program: Rate 10 °C/min; Final Temperature 120 °C; Hold Time 0 min; Rate 15 °C/min; Final Temperature 250 °C; Hold Time 3 min
- Equilibration Time 3 min
- Max Temperature 250 °C
- Mode Split
- Heater On 250 °C
- Pressure On 10.802 psi
- Gas Saver Off
- Split Ratio 10:1

- Split Flow 14.301 mL/min
- MSD Transfer Line, Temperature 230°C
- Column 1: Supelco 28662-U 52964-01B; diameter 180.00 µm; film thickness 1.00 µm; length 20.0 m; maximum temperature 300.0 °C; Flow 0.4806 ml/min
- Column 2: Agilent 122-1364; diameter 250 µm; film thickness 1.40 µm; length 60.0 m; maximum temperature 260.0 °C; Flow 1.4301 ml/min

A GC-MS 5977 with direct injection of Agilent with software Mass Hunter GC/MS Acquisition B.07.00.SP2.1654 equipped with a NIST library database was used for the identification of analytes in samples of solvent recovery.

The instrumental operative conditions are described as follow:

- GC Oven Temperature Setpoint On
- Initial Temperature 40 °C
- Hold Time 12 min
- Post Run 70 °C
- Program: Rate 15 °C/min; Final Temperature 150 °C; Hold Time 0 min; Rate 20 °C/min; Final Temperature 260 °C; Hold Time 6 min
- Equilibration Time 0.25 min
- Max Temperature 260 °C
- Syringe Size 10 µL
- Injection Volume 1 µL
- Mode Split
- Heater On 110 °C
- Pressure On 23.708 psi
- Gas Saver On 20 After 3 min mL/min
- Split Ratio 15 :1
- Split Flow 22.5 mL/min
- MSD Transfer Line, Temperature 235°C
- Column: Agilent 122-1364; diameter 250 µm; film thickness 1.40 µm; length 60.0 m; maximum temperature 260.0 °C; Flow 1.5 ml/min

A Karl Fischer of Mettler Toledo was used for determination of water content in the samples.

3. Operative Procedures for analysis of samples of “Solvent Recovery”

3.1. Stock Solutions and Calibration Standards for GC-FID

Stock solutions for GC-FID called respectively “*Multi-standard SR1*”, “*Multi-standard SR2*”, and “*Multi-standard SR3*” were prepared from acquired pure standard materials weighting 1 ml of each solvent in a final volume of 25 mL of pentane. The composition of the three calibration standards are:

- *Multi-standard SR1*: Methanol; Ethanol; Acetone; Acetonitrile; Hexane; Isopropanol; Methyl acetate; Dichloromethane; 2-Butanone; Ethyl acetate; Tetrahydrofuran; Cyclohexane; Isopropyl Acetate; Heptane; 1,4-Dioxane; Cyclopentyl methyl ether; Toluene; Butyl Acetate; N,N-Dimethylformamide.
- *Multi-standard SR2*: 2-Butanol; 1-Butanol; Pyridine.
- *Multi-standard SR3*: Diisopropylether; Octane.

The stock solutions had a concentration of about 30000 mg/L for each solvent. Five calibration standards were prepared from stock solutions diluting with pentane to obtain a calibration curve in a concentration range between 1000 and 30000 mg/L. For this study, dilutions for calibration curve were about 1500; 6000; 15000; 22500; 30000 mg/L.

The Appendix, section A1 shows an example of calibration curve for all the compounds.

3.2. Analysis of the samples

All the calibrations were daily controlled with known concentration standards.

Periodically a quality control check with a certified matrix was evaluated to assure the quality of analytical data.

The samples of solvents (wastes, secondary raw materials) was characterized in term of water content (% w/w) through Karl Fischer’s titrimetric method. A weighted small quantity of the sample was put inside the instrument for Karl Fischer titration through a syringe and the analysis is made automatically. The knowledge of water content is important because of the necessity of its elimination from the secondary raw material. The determination of mean water equivalence factor of Karl Fischer was carried out after 4 injection of weighted amount of water 100 % w/w and the mean value was taken in consideration for analysis. The performance of Karl Fischer method was daily verified with certified matrix with a water content of 1 % w/w.

VOC composition is determined through GC-MS with headspace injection or direct injection for qualitative analysis and GC-FID for quantitative analysis. The sample is directly injected in GC-FID without pretreatment.

4. Validation of the analytical method

4.1. Limit of Detection (LOD) and Limit of Quantification (LOQ)

The procedure for the determination of Limit of Detection (LOD) and Limit of Quantification (LOQ) implied the injection of 10 standard samples with a concentration range of 200-500 mg/L of all compounds of interest.

A stock solution, called “*Model Standard SR1*” was prepared weighting 1 ml of each compounds and diluting with acetone in a final volume of 25 mL.

- *Model Standard SR1*: Methanol; Ethanol; Acetonitrile; Hexane; Isopropanol; Methyl acetate; Dichloromethane; 2-Butanone; Ethyl acetate; Tetrahydrofuran; Cyclohexane; Isopropyl Acetate; Heptane; 1,4-Dioxane; Cyclopentyl methyl ether; Toluene; Butyl Acetate; N,N-Dimethylformamide.

The “Model Standard SR1” was diluted with acetone with a dilution factor of 20 to obtain the “Primary Dilution Solution” from which a calibration curve was obtained in a concentration range between 300-1500 mg/L for all compounds. The preparation of the “Model Standard SR1” allowed to study a model solution very similar with real samples of distillates of secondary raw materials and final products for the “Solvent Recovery”.

4.2. Experiments of repeatability on Certified Matrix

Experiments of repeatability on Certified Matrix were performed on three levels of concentration (2000, 10000, 15000 mg/L) of methanol, ethanol, isopropanol, acetonitrile, ethyl acetate, toluene; each sample was prepared from certified matrix which were acquired for this specific purpose.

Starting from the “Certified Matrix”, the preparation of three level of concentration by dilution in acetone, and the injection of seven samples for each level of concentration allowed the determination of the method’s performance.

- *Certified Matrix*: Methanol, Ethanol, Isopropanol, Acetonitrile, Ethyl Acetate, Toluene

Each level of the concentration was injected in seven different vials and they were analyzed by GC-FID.

5. Experiments of Recovery

Experiments of recovery were carried out by the preparation of the Initial Calibration Verification (ICV) with a concentration closer at the centrum of the calibration curve (≈ 15000 mg/L).

Experiments to evaluate matrix effects take in consideration a Laboratory Control Standard (LCS) and a real matrix spiked with a known concentration of the same magnitude order of the LCS.

6. Qualitative analysis by GC-MS

A diluted sample with a dilution factor of 1000 in water was analyzed by GC-MS with headspace injection to identify all the components. In alternative, a diluted sample with a dilution factor in methanol was analyzed by GC-MS with direct injection to identify all the components. After the identification, all the samples are quantified by GC-FID.

7. Quantitative analysis by GC-MS with direct injection

Stock solutions called respectively “*Multi-standard MS-SR1*”, “*Multi-standard MS-SR2*” were prepared from acquired pure standard materials weighting 100 μ l of each solvent and diluted in methanol in a final volume of 25 mL:

- *Multi-standard MS-SR1*: Ethanol; Acetone; Acetonitrile; Hexane; Isopropanol; Methyl acetate; Dichloromethane; 2-Butanone; Ethyl acetate; Tetrahydrofuran; Cyclohexane; Isopropyl Acetate; Heptane; 1,4-Dioxane; Cyclopentyl methyl ether; Toluene; Butyl Acetate; N,N-Dimethylformamide.
- *Multi-standard MS-SR2*: Diisopropyleter; 2-Butanol; 1-Butanol; Pyridine; Octane.

The stock solutions had a concentration of about 3000 mg/L for each solvent. Five calibration standards are prepared from stock solutions diluting with methanol (dilution factor 10) to obtain a calibration curve in a concentration range between 15 and 3000 mg/L. For our study, dilutions for calibration curves were about: 15; 60; 150; 225; 300 mg/L.

RESULTS AND DISCUSSION

1. Qualitative analysis by GC-MS and Quantitative analysis with GC-FID.

The analysis operative procedure of a sample of "Solvent Recovery" describes all the steps to carry out a testing analysis. The qualitative screening with GC-MS is fundamental to recognize all chemical compounds of the sample.

To overcome the difficulties of co-elution of different compounds, the qualitative analysis by GC-MS is fundamental for assuring the best identification, improving the quality of analytical data.

The following figures show the chromatograms obtained by GC-MS for "Multi-standard MS-SR1" and "Multi-standard MS-SR2".

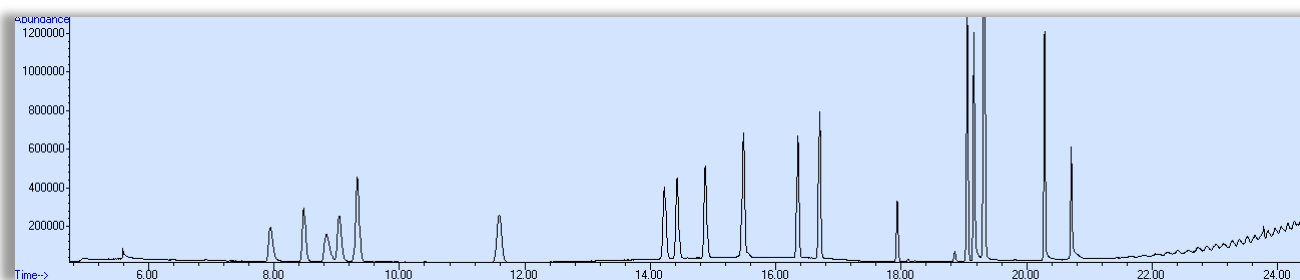


Figure 3: Chromatogram obtained by GC-MS with direct injection of "Multi-standard MS-SR1". The elution order from left to right is: Ethanol; Isopropanol; Acetone; Acetonitrile; Methyl acetate; Dichloromethane; Hexane; 2-Butanone; Ethyl acetate; Tetrahydrofuran; Cyclohexane; Isopropyl Acetate; Heptane; 1,4-Dioxane; Cyclopentyl methyl ether; Toluene; Butyl Acetate; N,N-Dimethylformamide.

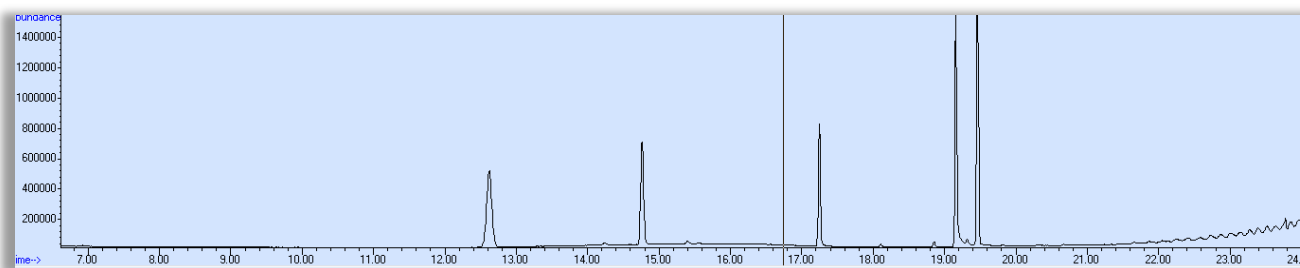


Figure 4: Chromatogram obtained with GC-MS with direct injection of for "Multi-standard MS-SR2".

The elution order from left to right is Diisopropyl ether; 2-Butanol; 1-Butanol; Pyridine; Octane.

The preparation of two stock solutions to calibrate the GC-MS is necessary because some compounds showed comparable retention times. If necessary, it is possible to perform the quantitative analysis by GC-MS, following the initial identification with NIST libraries.

After the screening with GC-MS, the samples can be analysed through GC-FID.

The following figures show the chromatograms of "Multi-standard SR1"; "Multi-standard SR2" and "Multi-standard SR3" obtained with GC-FID. The preparation of three different stock solutions, with their own calibration, is

necessary to evaluate a high number of compounds and to distinguish different peaks with the same retention time.

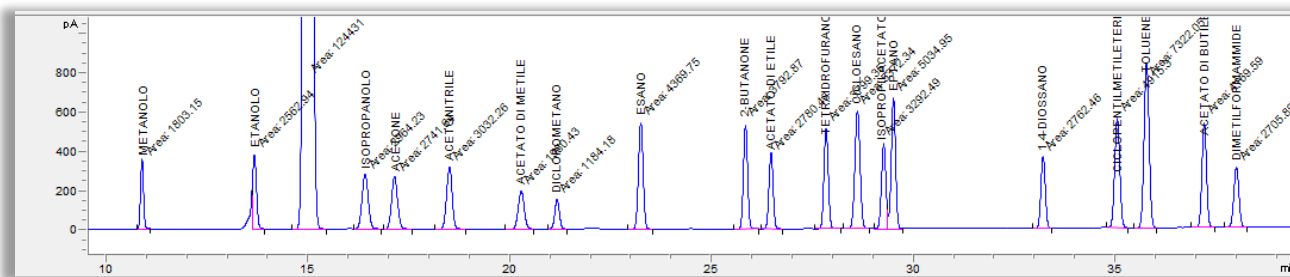


Figure 5: Chromatogram obtained with GC-FID of “Multi-standard SR1”

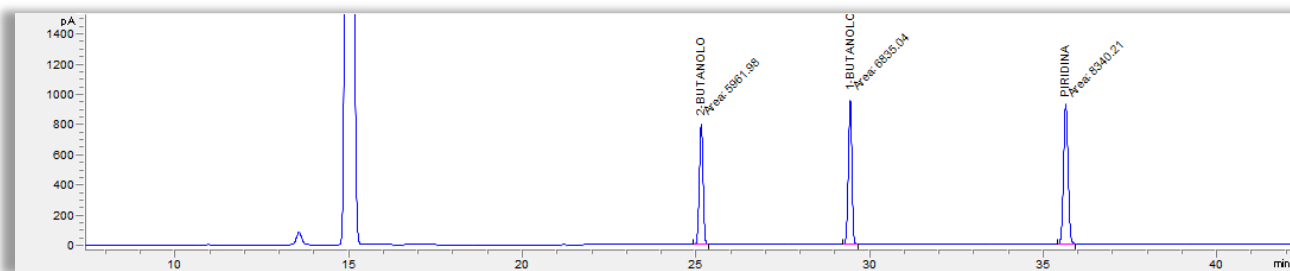


Figure 6: Chromatogram obtained with GC-FID of “Multi-standard SR2”

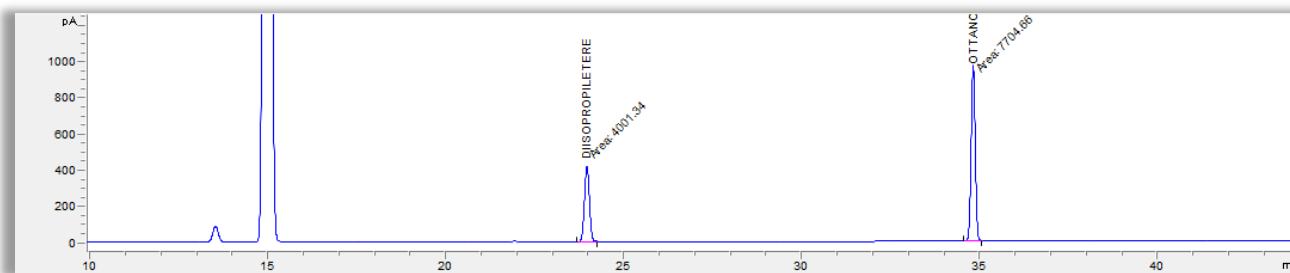


Figure 7: Chromatogram obtained with GC-FID of “Multi-standard SR3”

The preparation of “Multi-standard SR1”, “Multi-standard SR2” and “Multi-standard SR3”, followed during the experiments, as described in Materials and Methods, was the best manner to manage the analysis in accordance with customers’ request.

2. Calibration curves and procedures to control analytical data

All compounds of the stock solutions (“Multistandard SR1”, “Multistandard SR2”, “Multistandard SR3”) resulted in a good linearity between the range of analysis (1000 – 30000 mg/L), as demonstrated in the calibration curves described in Appendix, section A1. This calibration, characterized by a high range of concentration, allowed the analysis of liquid waste’s samples, which presented a high concentration of different solvents.

Before starting with any analytical testing, it is necessary to ensure the best instrumental performance through the execution of the “Initial Calibration Verification (ICV)” standard testing, with an acceptance range between 80-120% of the given value of the standard.

The following tables shows an example of daily quality control, performed to validate the calibration for quantitative analysis.

COMPUNDS	RETENTION TIME (min)	CONC. STOCK SOLUTION (mg/L)	DILUTION	ICV (mg/L)	VALUE (mg/L)	RANGE (80-120)%	VALIDATION
METHANOL	10.915	30024	1	15012	17041.7	113.52	TRUE
ETHANOL	13.690	30496	1	15248	18145.3	119.00	TRUE
ISOPROPANOL	16.466	29264	1	14632	16591	113.39	TRUE
ACETONE	17.166	29280	1	14640	16472.5	112.52	TRUE
ACETONITRILE	18.500	31188	1	15594	17651.8	113.20	TRUE
METHYL ACETATE	20.274	34616	1	17308	19591.3	113.19	TRUE
DICHLOROMETANE	21.164	45228	1	22614	25653.7	113.44	TRUE
HEXANE	23.240	25416	1	12708	13867.7	109.13	TRUE
2-BUTANONE	25.832	31824	1	15912	17923.6	112.64	TRUE
ETHYL ACETATE	26.465	34804	1	17402	19581	112.52	TRUE
TETRAHYDROFURAN	27.831	33640	1	16820	18715.5	111.27	TRUE
CYCLOHEXANE	28.605	29724	1	14862	16046.8	107.97	TRUE
ISOPROPYL ACETATE	29.202	34356	1	17178	19103.4	111.21	TRUE
HEPTANE	29.501	26840	1	13420	14762.6	110.00	TRUE
1,4-DIOXANE	33.203	39440	1	19720	21950.7	111.31	TRUE
CYCLOPENTYL METHYL ETHER	35.044	33528	1	16764	18484	110.26	TRUE
TOLUENE	35.771	33688	1	16844	18613.2	110.50	TRUE
BUTYL ACETATE	37.198	34412	1	17206	19349.5	112.46	TRUE
DIMETHYL FORMAMIDE	37.994	38540	1	19270	21591.3	112.05	TRUE

Table 2.: Quality control of "Multi-standard SR1"

COMPUNDS	RETENTION TIME (min)	STOCK SOLUTION (mg/L)	DILUTION	ICV (mg/L)	VALUE (mg/L)	RANGE (80-120)%	VALIDATION
2-BUTANOL	25.136	30872	1	18523	20551.8	110.95	TRUE
1-BUTANOL	29.430	31544	1	18926	21036.1	111.15	TRUE
PYRIDINE	35.635	39292	1	23575	24367.7	103.36	TRUE

Table 3.: Quality control of "Multi-standard SR2"

COMPUNDS	RETENTION TIME (min)	STOCK SOLUTION (mg/L)	DILUTION	ICV (mg/L)	VALUE (mg/L)	RANGE (80-120)%	VALIDATION
DIISOPROPTL ETHER	23.954	26992	1	13496	13966.5	103.49	TRUE
OCTANE	34.810	28108	1	14054	15886.5	113.04	TRUE

Table 4.: Quality control of "Multi-standard SR3"

3. LOD and LOQ

Limit of Detection (LOD) and Limit of Quantification (LOQ) were evaluated for the following compounds:

- Methanol; Ethanol; Acetonitrile; Hexane; Isopropanol; Methyl acetate; Dichloromethane; 2-Butanone; Ethyl acetate; Tetrahydrofuran; Cyclohexane; Isopropyl Acetate; Heptane; 1,4-Dioxane; Cyclopentyl methyl ether; Toluene; Butyl Acetate; N,N-Dimethylformamide.

The LOD and LOQ were calculated by:

$$LOD = (3 * \sigma) / k$$

$$LOQ = (10 * \sigma) / k$$

where σ is the standard deviation of peak areas and k is the slope of calibration curve.

The Appendix, section A2 shows the experimental data obtained for the determination of LOD and LOQ.

The values are summarized in the following table:

COMPOUNDS	LOD (mg/L)	LOQ (mg/L)
METHANOL	31.7	105.7
ETHANOL	30.8	102.7
ISOPROPANOL	22.2	74.1
ACETONITRILE	37.8	126.0
METHYL ACETATE	31.3	104.4
DICHLOROMETANE	52.1	173.6
HEXANE	31.7	105.6
2-BUTANONE	29.4	97.8
ETHYL ACETATE	30.6	102.0
TETRAHYDROFURAN	30.1	100.3
CYCLOHEXANE	28.2	94.0
ISOPROPYL ACETATE	83.0	276.8
HEPTANE	37.7	125.8
1,4-DIOXANE	50.0	166.7
CYCLOPENTYL-METHYL-ETHER	28.1	93.7
TOLUENE	33.0	109.9
BUTHYL ACETATE	39.0	129.8
N,N-DIMETHYLFORMAMIDE	35.8	119.2

Table 5: Limit of Detection (LOD) and Limit of Quantification (LOQ)

This analytical method resulted to be adequate and it showed successful results, in terms of reliability and performance, in order to analyse “In Process Control” samples, which are important to set and control the industrial production processes. The sampling and the analysis of “In Process Control” samples, during the different steps of production, provide information about the performance of the industrial processes and support the meeting of the process specification setting.

This method was successfully applied to analyse waste solvents before and after the distillation treatment.

One of the most important product of the “Solvent Recovery” process in the industrial plant, object of this study, is pure acetone. Key important factor of the industrial distillation of pure acetone is to ensure the presence of

low impurities concentration. It is obvious that the analytical testing method needs to be capable to detect low level of impurities. This developed analytical method allows the quantification of compounds present in low concentration (0.01%) in pure acetone.

4. Experiments of repeatability on Certified Matrix

The performance of this analytical method was evaluated using a certified matrix.

A certified matrix with all the compounds (24) of interest for this study was not available in the market and the experiments of repeatability were carried out on a certified matrix that contained six compounds: methanol, ethanol, isopropanol, acetonitrile, ethyl acetate and toluene.

In the Appendix, the section A3 shows the experimental data of repeatability and *z-scores* obtained during the tests carried out with certified matrix.

The following tables show the summary of the results:

	Conc (mg/L)_Level 1	Conc (mg/L)_Level 2	Conc (mg/L)_Level 3
	METHANOL	METHANOL	METHANOL
Spike (mg/L)	2531.646	12658.228	16911.392
Mean	2932.550	11408.318	14887.523
STD Dev.	63.924	752.898	1050.219
CV%	2.180	6.600	7.054
Recovery%_Mean	115.84	90.13	88.03
Recovery%_STD Dev.	2.52	5.95	6.21

Table 6: Repeatability tests carried out on Certified Matrix for Methanol

	Conc (mg/L)_Level 1	Conc (mg/L)_Level 2	Conc (mg/L)_Level 3
	ETHANOL	ETHANOL	ETHANOL
Spike (mg/L)	2534.85	12674.27	16932.83
Mean	2820.206	12720.623	16082.926
STD Dev.	134.308	1084.120	995.123
CV%	4.762	8.523	6.187
Recovery%_Mean	111.26	100.37	94.98
Recovery%_STD Dev.	5.30	8.55	5.88

Table 7: Repeatability tests carried out on Certified Matrix for Ethanol

	Conc (mg/L)_Level 1	Conc (mg/L)_Level 2	Conc (mg/L)_Level 3
	ISOPROPANOL	ISOPROPANOL	ISOPROPANOL
Spike (mg/L)	2544.53	12722.65	16997.46
Mean	2692.84	11457.29	14749.18
STD Dev.	153.21	845.65	1110.37
CV%	5.69	7.38	7.53
Recovery%_Mean	105.83	90.05	86.77
Recovery%_STD Dev.	6.02	6.65	6.53

Table 8: Repeatability tests carried out on Certified Matrix for Isopropanol

	Conc (mg/L)_Level 1	Conc (mg/L)_Level 2	Conc (mg/L)_Level 3
	ACETONITRILE	ACETONITRILE	ACETONITRILE
Spike (mg/L)	2544.53	12722.65	16997.46
Mean	3005.816	10693.021	14239.913
STD Dev.	126.980	531.895	1072.346
CV%	4.224	4.974	7.531
Recovery%_Mean	118.13	84.05	83.78
Recovery%_STD Dev.	4.99	4.18	6.31

Table 9: Repeatability tests carried out on Certified Matrix for Acetonitrile

	Conc (mg/L)_Level 1	Conc (mg/L)_Level 2	Conc (mg/L)_Level 3
	ETHYL ACETATE	ETHYL ACETATE	ETHYL ACETATE
Spike (mg/L)	2229.65	11148.27	15161.65
Mean	2104.141	10459.946	13410.256
STD Dev.	84.132	859.046	1010.877
CV%	3.998	8.213	7.538
Recovery%_Mean	94.37	93.83	88.45
Recovery%_STD Dev.	3.77	7.71	6.67

Table 10: Repeatability tests carried out on Certified Matrix for Ethyl Acetate

	Conc (mg/L)_Level 1	Conc (mg/L)_Level 2	Conc (mg/L)_Level 3
	TOLUENE	TOLUENE	TOLUENE
Spike (mg/L)	2306.81	11534.03	15409.46
Mean	2524.831	11568.133	15216.345
STD Dev.	130.725	1050.384	1139.090
CV%	5.178	9.080	7.486
Recovery%_Mean	109.45	100.30	98.75
Recovery%_STD Dev.	5.67	9.11	7.39

Table 11: Repeatability tests carried out on Certified Matrix for Toluene

For all the compounds of certified matrix, the experiments confirmed good performance of the method.

The evaluation of *z-score* is important to evaluate the precision and accuracy of the method.

The *z-score* was calculated by:

$$z\ score = \frac{x - X_m}{\sigma}$$

where *x* is the experimental result, *X_m* is the mean value and σ is the standard deviation.

The *z-score* is considered acceptable when its value is within the range $-2 < Z\text{-score} > +2$.

The Recovery is calculated as the percentage ratio between the obtained analytical result and the certified spike's value. All the values of recovery collected in these experiments were in the acceptance range between 80-120%.

The following plots show the *z-scores* associated with all the compounds of certified matrix and the percentage of recovery (Appendix, section A3).



Figure 8: Plot of z-score obtained for Level 1



Figure 9: Plot of z-score obtained for Level 2



Figure 10: Plot of z-score obtained for Level 3

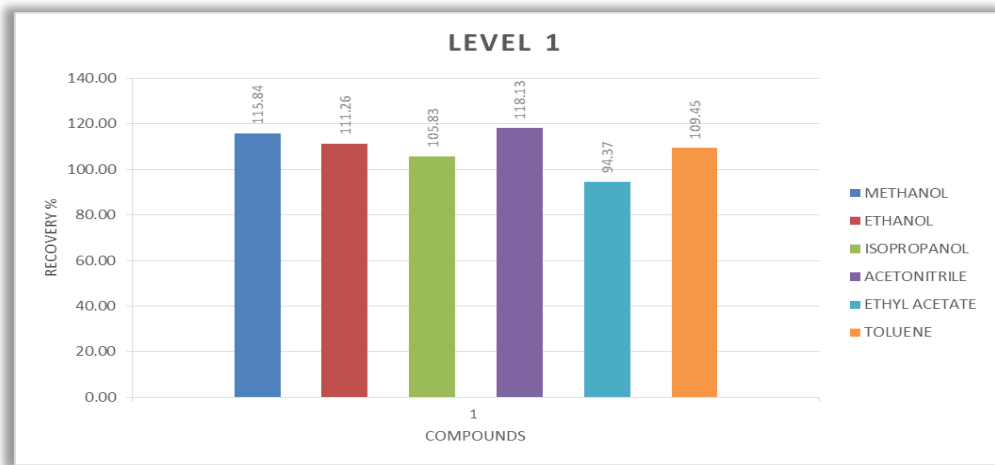


Figure 11: Percentage of Recovery obtained for Level 1

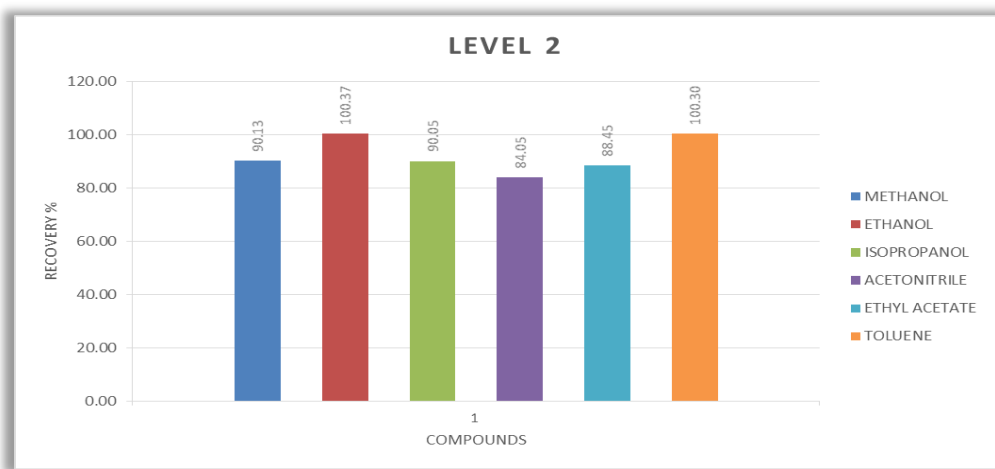


Figure 12: Percentage of Recovery obtained for Level 2

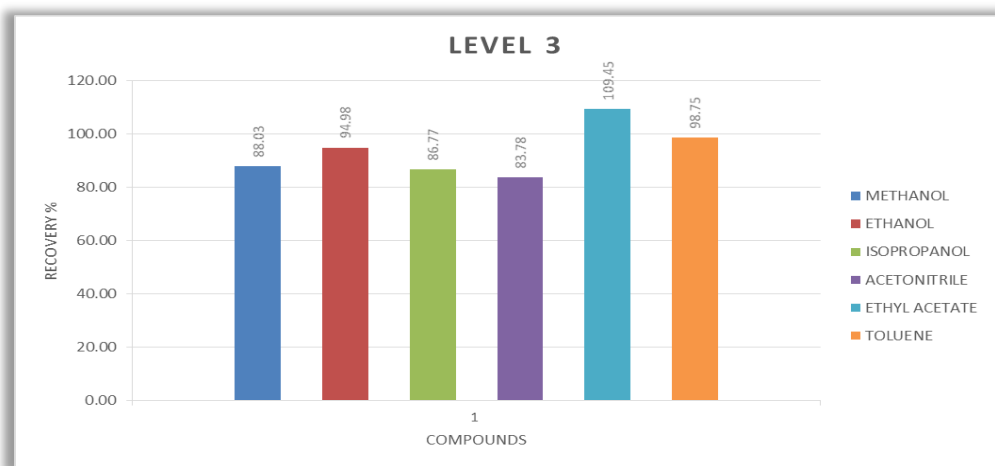


Figure 13: Percentage of Recovery obtained for Level 3

5. Recovery of the method applied to six target compounds

The objective of this study was the evaluation of the quality of analytical data using a statistical approach.

In this study, a particular attention was focused on the control of the recovery percentage, obtained during a long period.

The comparison between data obtained from the tests carried out on certified matrix and data obtained by daily calibration's control (ICV) gave the opportunity to demonstrate the accuracy of the analytical method.

Recovery data of ICV for six target compounds (methanol, ethanol, isopropanol, acetonitrile, ethyl acetate and toluene) were collected after 25 tests, as reported in the Appendix, section A4.

The following tables show the mean, the standard deviation and the Analysis of Variance (ANOVA) obtained for the 6 compounds under observation after 25 measures of ICV.

RECOVERY %	GROUPS					
	METHANOL	ETHANOL	ISOPROPANOL	ACETONITRILE	ETHYL ACETATE	TOLUENE
DIMENSION	25	25	25	25	25	25
MEAN	111.07	112.97	112.34	108.22	111.06	108.46
STD DEV.	6.30	9.29	6.45	7.00	5.56	6.35

Table 12: Mean and Standard Deviation obtained after 25 measures for methanol, ethanol, isopropanol, acetonitrile, ethyl acetate and toluene.

Analysis of Variance: one factor
RESUME

Groups	Count	Sum	Mean	Variance
METHANOL	25.00	2776.80	111.07	39.65
ETHANOL	25.00	2824.23	112.97	86.22
ISOPROPANOL	25.00	2808.59	112.34	41.61
ACETONITRILE	25.00	2705.53	108.22	49.02
ETHYL ACETATE	25.00	2776.45	111.06	30.87
TOLUENE	25.00	2711.49	108.46	40.38

ANOVA						
Origin of the variation	SQ	Degrees of Freedom	MQ	F	Significancevalue	F crit
Between groups	481.99	5.00	96.40	2.01	0.08	3.15
Within groups	6906.06	144.00	47.96			
Total	7388.05	149.00				

Table 13: Test of Analysis of Variance (ANOVA)

ANOVA demonstrated that $F < F_{crit}$ and there is the same parametric variance between the groups and so the six compounds belong to the same population.

A further comparison between data obtained from certified matrix and from measures carried out on ICV is given by Z test. The Z test evaluates whether the mean for the 6 target compounds of certified matrix is closer and comparable with the mean of percentage of recovery (recovery %) obtained for ICV. The mean values of recovery for Level 1, Level 2 and Level 3 of concentration of certified matrix were compared with the mean value of recovery of ICV.

Z test was calculated by:

$$Z = \frac{X - \mu}{\sigma\sqrt{n}}$$

where X is the mean value and n is the number of measures (Tables 6-11); μ is the mean value and σ is the standard deviation of a reference population of measures (Table 12).

The table below summarizes the outcome of the above-mentioned Z test.

LEVEL 1	GROUPS					
	METHANOL	ETHANOL	ISOPROPANOL	ACETONITRILE	ETHYL ACETATE	TOLUENE
DIMENSION	7.00	7.00	7.00	7.00	7.00	7.00
MEAN	115.84	111.26	105.83	118.13	94.37	109.45
STD DEV.	2.52	5.30	6.02	4.99	3.77	5.67
MEAN_Population	111.07	112.97	112.34	108.22	111.06	108.46
STD DEV_Population	6.30	9.29	6.45	7.00	5.56	6.35
TEST Z (Z=2.2326)	0.29	-0.07	-0.38	0.53	-1.14	0.06

LEVEL 2	GROUPS					
	METHANOL	ETHANOL	ISOPROPANOL	ACETONITRILE	ETHYL ACETATE	TOLUENE
DIMENSION	7.00	7.00	7.00	7.00	7.00	7.00
MEAN	90.13	100.37	90.05	84.05	93.83	100.30
STD DEV.	5.95	8.55	6.65	4.18	7.71	9.11
MEAN_Population	111.07	112.97	112.34	108.22	111.06	108.46
STD DEV_Population	6.30	9.29	6.45	7.00	5.56	6.35
TEST Z (Z=2.2326)	-1.26	-0.51	-1.31	-1.30	-1.17	-0.49

LEVEL 3	GROUPS					
	METHANOL	ETHANOL	ISOPROPANOL	ACETONITRILE	ETHYL ACETATE	TOLUENE
DIMENSION	7.00	7.00	7.00	7.00	7.00	7.00
MEAN	88.03	94.98	86.77	83.78	88.45	98.75
STD DEV.	6.21	5.88	6.53	6.31	6.67	7.39
MEAN_Population	111.07	112.97	112.34	108.22	111.06	108.46
STD DEV_Population	6.30	9.29	6.45	7.00	5.56	6.35
TEST Z (Z=2.2326)	-1.38	-0.73	-1.50	-1.32	-1.54	-0.58

Table 14: Comparison between data obtained from certificate matrix and from measures carried out on ICV by Z test

The acceptance range for Z is: $-2.2326 < Z < +2.2326$ which corresponds to a significance level $\alpha=0.01$.

Test Z applied in this experiment demonstrate that a good comparison exists between the mean of the recovery % obtained with tests on certified matrix for 3 levels of concentration and the mean of the recovery % obtained from the population of ICV.

6. Analytical Method applied for the determination of 24 compounds

The results obtained opened the way to a development of an analytical method for the determination of 24 compounds in waste solvents and in mix of organic solvents.

The compounds of interests were:

- Methanol; Ethanol; Isopropanol; Acetone; Acetonitrile; Methyl Acetate; Dichloromethane; Hexane; 2-Butanone; Ethyl Acetate; Tetrahydrofuran; Cyclohexane; Isopropyl Acetate; Heptane; 1,4-Dioxane; Cyclopentyl methyl ether; Toluene; Butyl Acetate; N,N-Dimethylformamide; 2-Butanol; 1-Butanol; Pyridine; Diisopropylether; Octane.

These compounds are organized in three stock solutions, as described in Materials and Methods: "Multi-standard SR1", "Multi-standard SR2", and "Multi-standard SR3".

The Appendix, section A5 reported the results of 70 tests characterized by a population of 1680 measures of recovery of 24 compounds of interest.

The following table shows the mean and standard deviation of the distribution of the recovery (%) for 24 compounds.

RECOVERY %	Group 1	Group 2	Group 3	Group 4	Group 5	Group 6	Group 7	Group 8
	METHANOL	ETHANOL	ISOPROPRANOL	ACETONE	ACETONITRILE	METHYL ACETATE	DICHLOROMETANE	HEXANE
DIMENSION	70	70	70	70	70	70	70	70
MEAN	98.57	101.55	98.07	100.83	99.39	101.93	100.98	107.81
STD DEV.	9.85	10.43	10.14	10.00	9.30	10.34	8.76	14.57
CONFIDENCE 99%	3.03	3.21	3.12	3.08	2.86	3.18	2.70	4.48
L 1	95.53	99.20	95.68	98.59	97.31	99.54	98.98	104.17
L 2	101.60	103.89	100.46	103.06	101.46	104.31	102.97	111.45
LCL	95.03	97.80	94.44	97.24	96.05	98.22	97.84	102.58
UCL	102.10	105.29	101.71	104.41	102.72	105.64	104.12	113.03

RECOVERY %	Group 9	Group 10	Group 11	Group 12	Group 13	Group 14	Group 15	Group 16
	2-BUTANONE	ETHYL ACETATE	TETRAHYDROFURAN	CYCLOHEXANE	ISOPROPHYL ACETATE	HEPTANE	1,4-DIOXAN	CYCLOPENTYL METHYL ETHER
DIMENSION	70	70	70	70	70	70	70	70
MEAN	99.16	100.07	100.15	109.79	99.68	101.90	98.30	101.14
STD DEV.	8.77	10.22	8.65	24.73	10.68	18.66	9.50	8.27
CONFIDENCE 99%	2.70	3.15	2.66	7.61	3.29	5.75	2.93	2.55
L 1	96.99	97.72	98.19	104.04	97.09	97.64	96.03	99.23
L 2	101.33	102.42	102.11	115.55	102.26	106.16	100.58	103.04
LCL	96.02	96.40	97.05	100.93	95.85	95.21	94.90	98.17
UCL	102.31	103.73	103.25	118.66	103.51	108.59	101.71	104.10

RECOVERY %	Group 17	Group 18	Group 19	Group 20	Group 21	Group 22	Group 23	Group 24
	TOLUENE	BUTHYL ACETATE	DIMETHYLFORMAMIDE	2-BUTANOL	1-BUTANOL	PYRIDINE	DIISOPROPYL ETHER	OCTANE
DIMENSION	70	70	70	70	70	70	70	70
MEAN	101.34	99.99	99.87	100.08	99.02	98.78	96.31	98.11
STD DEV.	8.45	9.36	9.70	11.76	12.16	9.81	10.10	12.52
CONFIDENCE 99%	2.60	2.88	2.99	3.62	3.74	3.02	3.11	3.85
L 1	99.37	97.79	97.67	97.41	95.32	95.79	93.31	94.83
L 2	103.32	102.20	102.07	102.74	102.71	101.76	99.32	101.39
LCL	98.31	96.64	96.39	95.86	94.65	95.26	92.69	93.62
UCL	104.37	103.35	103.35	104.30	103.38	102.29	99.94	102.60

Table 15: Mean and Standard Deviation of the population of 70 tests characterized by a population of 1680 measures of recovery % of 24 compounds of interest

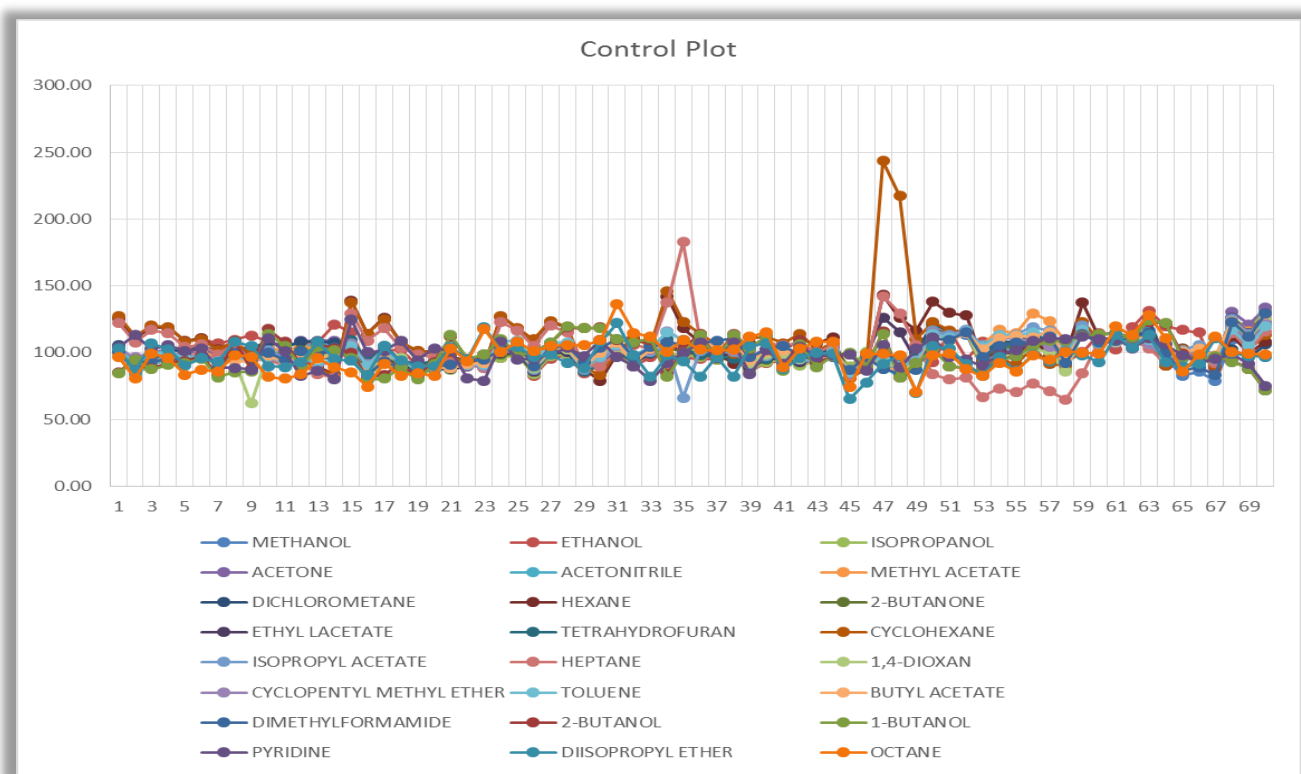


Figure 14: Flow Chart to control recovery % of 24 analytes

These measures gave a normal distribution of recovery for 24 compounds. A flow chart (Figure 14) is used to evaluate the fluctuations of the values around the mean value.

By reviewing the plot, it is possible to evaluate the presence of any outliers for very few compounds.

It is important to know the value of Lower Control Limit (LCL) and of the Upper Control Limit (UCL) obtained by:

$$LCL = X - \alpha \frac{\sigma}{\sqrt{n}}$$

$$UCL = X + \alpha \frac{\sigma}{\sqrt{n}}$$

The application of test of ANOVA demonstrates that, in this case, the total variance is not representative for all compounds.

The Appendix, section A5 shows the experimental data obtained for the determination of recovery.

Analysis of Variance; one factor

Resume

<i>Groups</i>	<i>Count</i>	<i>Sum</i>	<i>Mean</i>	<i>Variance</i>
METHANOL	70	6899.58	98.57	97.04
ETHANOL	70	7108.16	101.55	108.88
ISOPROPANOL	70	6865.18	98.07	102.75
ACETONE	70	7057.76	100.83	100.01
ACETONITRILE	70	6957.05	99.39	86.58
METHYL ACETATE	70	7134.89	101.93	106.96
DICHLOROMETANE	70	7068.49	100.98	76.70
HEXANE	70	7546.46	107.81	212.20
2-BUTANONE	70	6941.48	99.16	76.93
ETHYL LACETATE	70	7004.76	100.07	104.51
TETRAHYDROFURAN	70	7010.44	100.15	74.89
CYCLOHEXANE	70	7685.53	109.79	611.65
ISOPROPYL ACETATE	70	6977.39	99.68	114.06
HEPTANE	70	7132.90	101.90	348.33
1,4-DIOXAN	70	6881.19	98.30	90.29
CYCLOPENTYL METHYL ETHER	70	7079.49	101.14	68.42
TOLUENE	70	7094.08	101.34	71.41
BUTHYL ACETATE	70	6999.64	99.99	87.68
DIMETHYLFORMAMIDE	70	6990.96	99.87	94.08
2-BUTANOL	70	7005.40	100.08	138.38
1-BUTANOL	70	6931.05	99.02	147.86
PYRIDINE	70	6914.29	98.78	96.23
DIISOPROPYL ETHER	70	6741.96	96.31	102.07
OCTANE	70	6867.59	98.11	156.73

ANALYSIS OF VARIANCE

<i>Origin of variazione</i>	<i>SQ</i>	<i>Degrees of Freedom</i>	<i>MQ</i>	<i>F</i>	<i>Significance Value</i>	<i>F crit</i>
BetweenGroups	13578.63	23	590.38	4.33	0.00	1.82
WithinGroups	225950.12	1656	136.44			
Total	239528.76	1679				

Table 16: Test of Analysis of Variance (ANOVA) applied to all 24 compounds

In this ANOVA, $F > F_{crit}$ and the different compounds belong to different populations.

The ANOVA allows the classification of the compounds in three groups to explain the total variance:

- Recovery % less than 100
- Recovery % closer to 100
- Recovery higher than 100

ANOVA tests were applied to classify these populations in which $F < F_{crit}$.

Analysis of Variance; one factor

Resume				
Groups	Count	Sum	Mean	Variance
METHANOL	70	6899.58	98.57	97.04
ISOPROPANOL	70	6865.18	98.07	102.75
1,4-DIOXAN	70	6881.19	98.30	90.29
PYRIDINE	70	6914.29	98.78	96.23
DIISOPROPYL ETHER	70	6741.96	96.31	102.07
OCTANE	70	6867.59	98.11	156.73

ANALYSIS OF VARIANCE

Origin of variation	SQ	Degrees of Freedom	MQ	F	Significance Value	F crit
BetweenGroups	270.91	5	54.18	0.50	0.77	3.06
WithinGroups	44512.61	414	107.52			
Total	44783.52	419				

Table 17: ANOVA test for compounds that showed the mean of Recovery less than 100%

Analysis of Variance; one factor

Resume				
Groups	Count	Sum	Mean	Variance
ETHANOL	70	7108.16	101.55	108.88
ACETONE	70	7057.76	100.83	100.01
ACETONITRILE	70	6957.05	99.39	86.58
METHYL ACETATE	70	7134.89	101.93	106.96
DICHLOROMETANE	70	7068.49	100.98	76.70
2-BUTANONE	70	6941.48	99.16	76.93
ETHYL LACETATE	70	7004.76	100.07	104.51
TETRAHYDROFURAN	70	7010.44	100.15	74.89
ISOPROPYL ACETATE	70	6977.39	99.68	114.06
HEPTANE	70	7132.90	101.90	348.33
CYCLOPENTYL METHYL ETHER	70	7079.49	101.14	68.42
TOLUENE	70	7094.08	101.34	71.41
BUTHYL ACETATE	70	6999.64	99.99	87.68
DIMETHYLFORMAMIDE	70	6990.96	99.87	94.08
2-BUTANOL	70	7005.40	100.08	138.38
1-BUTANOL	70	6931.05	99.02	147.86

ANALYSIS OF VARIANCE

Origin of variation	SQ	Degrees of Freedom	MQ	F	Significance Value	F crit
BetweenGroups	946.84	15	63.12	0.56	0.91	2.05
WithinGroups	124591.99	1104	112.86			
Total	125538.83	1119				

Table 18: ANOVA test for compounds that showed the mean of Recovery closer to 100%

Analysis of Variance; one factor

Resume				
Groups	Count	Sum	Mean	Variance
HEXANE	70	7546.46	107.81	212.20
CYCLOHEXANE	70	7685.53	109.79	611.65

ANALYSIS OF VARIANCE

Origin of variation	SQ	Degrees of Freedom	MQ	F	Significance Value	F crit
BetweenGroups	138.15	1	138.15	0.34	0.56	6.82
WithinGroups	56845.53	138	411.92			
Total	56983.68	139				

Table 19: ANOVA test for compounds that showed the mean of Recovery higher than 100%

Recovery less than 100 %	Recovery closer to 100 %	Recovery higher than 100 %
METHANOL	ETHANOL	HEXANE
ISOPROPANOL	ACETONE	CYCLOHEXANE
1,4-DIOXAN	ACETONITRILE	
PYRIDINE	METHYL ACETATE	
DIISOPROPYL ETHER	DICHLOROMETANE	
OCTANE	2-BUTANONE	
	ETHYL LACETATE	
	TETRAHYDROFURAN	
	ISOPROPHYL ACETATE	
	HEPTANE	
	CYCLOPENTYL METHYL ETHER	
	TOLUENE	
	BUTHYL ACETATE	
	DIMETHYLFORMAMIDE	
	2-BUTANOL	
	1-BUTANOL	

Mean of Recovery %_1	Mean of Recovery %_2	Mean of Recovery %_3
98.02	100.44	108.80
STD Dev of Recovery %_1	STD Dev of Recovery %_2	STD Dev of Recovery %_3
1.10	2.49	7.19

Table 20: Classification of all compounds based on ANOVA applied to recovery values.

7. Reproducibility Experiments

Experiments to evaluate the reproducibility of analytical data were carried out through the analysis of ICV with two different instruments GC-FID (called “GC-FID_Principal” and “GC-FID_Control”). The “GC-FID_Principal” was considered as the principal instrument for all the tests and the “GC-FID_Control” was used to compare the series of analytical data.

The ICV were analysed with “GC-FID_Principal” and “GC-FID_Control” to generate two series of recovery % of seven data for each compound. The Appendix, section A6 shows the complete population of data.

The Test Z applied to both series of the data demonstrated that they were representative of the population of recovery data of 24 compounds, as described in paragraph 6 and in Appendix, section A5.

According with the formula of Test Z:

$$Z = \frac{X - \mu}{\sigma\sqrt{n}}$$

where μ and σ were respectively the mean and the standard deviation of the entire population of recovery data; X was the mean of the two series of measures of recovery and n is the number of measures.

The comparison between data of the two instruments was described in term of Bravais' coefficient R:

$$R = \frac{\sum(xi - mean(x)) * (yi - mean(y))}{\sqrt{\sum (xi - mean(x))^2 * \sum (yi - mean(y))^2}}$$

The following tables show the results obtained in term of mean and standard deviation:

GC-FID_Control	METHANOL	ETHANOL	ISOPROPANOL	ACETONE	ACETONITRILE	METHYLACETATE
DIMENSION	7	7	7	7	7	7
MEAN	98.81	99.83	99.97	102.05	100.19	98.65
STANDARD DEVIATION	11.27	14.42	14.64	11.51	13.82	9.97
MEAN_all	98.57	101.55	98.07	100.83	99.39	101.93
STD DEV_all	9.85	10.43	10.14	10.00	9.30	10.34
TEST Z (Z=2.2326)	0.01	-0.06	0.07	0.05	0.03	-0.12

GC-FID_Control	DICHLOROMETANE	HEXANE	2-BUTANONE	ETHYL LACETATE	TETRAHYDROFURAN	CYCLOHEXANE
DIMENSION	7	7	7	7	7	7
MEAN	102.93	98.80	95.89	99.83	87.32	100.22
STANDARD DEVIATION	10.26	13.62	17.87	10.82	23.47	13.86
MEAN_all	100.98	107.81	99.16	100.07	100.15	109.79
STD DEV_all	8.76	14.57	8.77	10.22	8.65	24.73
TEST Z (Z=2.2326)	0.08	-0.23	-0.14	-0.01	-0.56	-0.15

GC-FID_Control	ISOPROPYL ACETATE	HEPTANE	1,4-DIOXAN	CYCLOPENTYL METHYL ETHER	TOLUENE	BUTYL LACETATE
DIMENSION	7	7	7	7	7	7
MEAN	101.67	101.21	100.09	101.46	101.11	100.69
STANDARD DEVIATION	11.89	12.64	11.21	7.93	8.25	12.17
MEAN_all	99.68	101.90	98.30	101.14	101.34	99.99
STD DEV_all	10.68	18.66	9.50	8.27	8.45	9.36
TEST Z (Z=2.2326)	0.07	-0.01	0.07	0.01	-0.01	0.03

GC-FID_Control	DIMETHYLFORMAMIDE	2-BUTANOL	1-BUTANOL	PYRIDINE	DIISOPROPYL ETHER	OCTANE
DIMENSION	7	7	7	7	7	7
MEAN	101.16	96.99	97.54	94.78	101.68	102.58
STANDARD DEVIATION	13.37	22.20	22.74	8.68	13.07	12.89
MEAN_all	99.87	100.08	99.02	98.78	96.31	98.11
STD DEV_all	9.70	11.76	12.16	9.81	10.10	12.52
TEST Z (Z=2.2326)	0.05	-0.10	-0.05	-0.15	0.20	0.14

Table 21: Results of reproducibility test for 24 compounds obtained by "GC-FID_Control"

GC-FID_Principal	METHANOL	ETHANOL	ISOPROPANOL	ACETONE	ACETONITRILE	METHYLACETATE
DIMENSION	7	7	7	7	7	7
MEAN	98.55	104.83	95.42	101.61	97.75	99.49
STANDARD DEVIATION	7.24	9.13	7.25	7.82	7.72	5.86
MEAN_all	98.57	101.55	98.07	100.83	99.39	101.93
STD DEV_all	9.85	10.43	10.14	10.00	9.30	10.34
TEST Z (Z=2.2326)	0.00	0.12	-0.10	0.03	-0.07	-0.09

GC-FID_Principal	DICHLOROMETANE	HEXANE	2-BUTANONE	ETHYL LACETATE	TETRAHYDROFURAN	CYCLOHEXANE
DIMENSION	7	7	7	7	7	7
MEAN	102.70	106.87	96.63	95.24	97.52	106.35
STANDARD DEVIATION	6.69	17.86	5.55	5.21	5.56	18.35
MEAN_all	100.98	107.81	99.16	100.07	100.15	109.79
STD DEV_all	8.76	14.57	8.77	10.22	8.65	24.73
TEST Z (Z=2.2326)	0.07	-0.02	-0.11	-0.18	-0.11	-0.05

GC-FID_Principal	ISOPROPYL ACETATE	HEPTANE	1,4-DIOXAN	CYCLOPENTYL METHYL ETHER	TOLUENE	BUTYL LACETATE
DIMENSION	7	7	7	7	7	7
MEAN	95.30	102.35	89.78	96.00	95.66	93.42
STANDARD DEVIATION	6.19	16.58	13.12	5.70	5.05	4.53
MEAN_all	99.68	101.90	98.30	101.14	101.34	99.99
STD DEV_all	10.68	18.66	9.50	8.27	8.45	9.36
TEST Z (Z=2.2326)	-0.16	0.01	-0.34	-0.23	-0.25	-0.27

GC-FID_Principal	DIMETHYLFORMAMIDE	2-BUTANOL	1-BUTANOL	PYRIDINE	DIISOPROPYL ETHER	OCTANE
DIMENSION	7	7	7	7	7	7
MEAN	94.29	93.67	91.83	94.90	96.45	87.68
STANDRD DEVIATION	5.70	9.99	9.51	8.97	9.95	9.60
MEAN_all	99.87	100.08	99.02	98.78	96.31	98.11
STD DEV_all	9.70	11.76	12.16	9.81	10.10	12.52
TEST Z (Z=2.2326)	-0.22	-0.21	-0.22	-0.15	0.00	-0.31

Table 22: Results of reproducibility test for 24 compounds obtained by "GC-FID_Principal"

The following plots shows the mean recovery obtained by "GC-FID_Control" and "GC-FID_Principal"

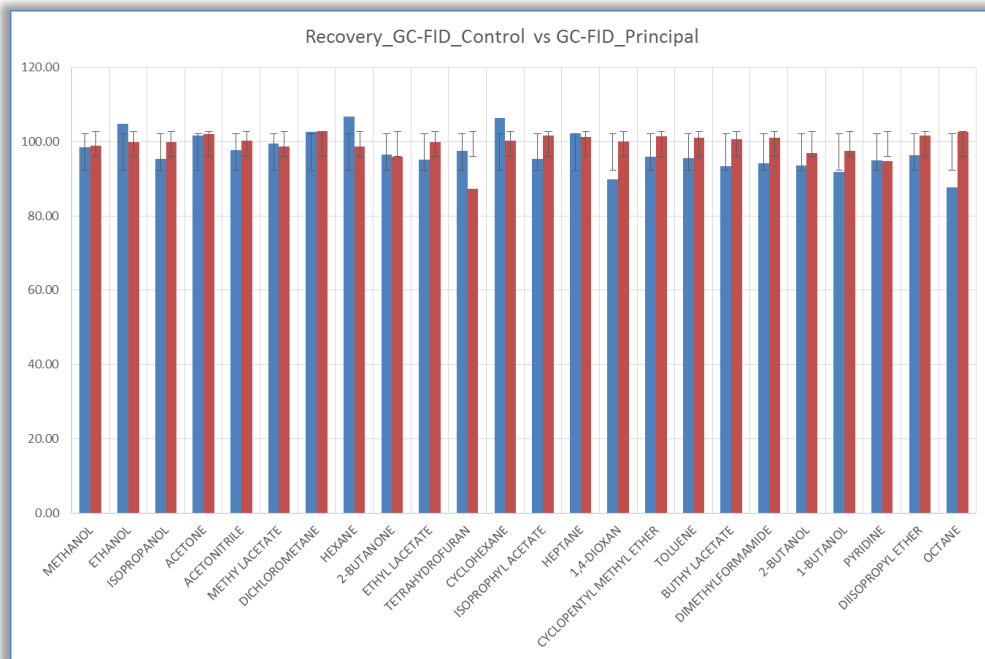


Figure 15: Comparison between recovery obtained with GC-FID_Control (Red) and GC-FID_Principal (Blue).

	METHANOL	ETHANOL	ISOPROPANOL	ACETONE	ACETONITRILE	METHYL ACETATE	DICHLOROMETANE	HEXANE
BRAVAIS_R	0.59	0.16	0.79	0.68	0.74	0.44	0.59	0.79
	ETHYL ACETATE	TETRAHYDROFURAN	CYCLOHEXANE	ISOPROPYL ACETATE	HEPTANE	1,4-DIOXAN	CYCLOPENTYL METHYL ETHER	
BRAVAIS_R	0.56	-0.46	0.82	0.61	0.83	0.63	0.04	
	BUTHYL ACETATE	DIMETHYLFORMAMIDE	2-BUTANOL	1-BUTANOL	PYRIDINE	DIISOPROPYL ETHER	OCTANE	
BRAVAIS_R	0.54	0.72	0.73	0.73	-0.72	0.38	0.23	

Table 23: Test of Bravais for comparison of recovery obtained through analytical data from two instruments

The results obtained showed the existence of reproducibility of analytical recoveries for all compounds.

8. Experiments on matrix spiked

This study was focused to evaluate the quality of analytical data for the method used and the tests were carried out on standard solutions. It is important to evaluate the data in the case of a real matrix to evaluate how matrix can affect the analysis.

Matrix effects can be detected comparing a Laboratory Control Sample (LCS) with Matrix (M), Matrix Spike (MS) and Matrix Spike Duplicate (MSD). The LCS, MS and MSD were characterized by similar concentration to evaluate the difference between recoveries: the composition of a real matrix could affect the quality of analytical data. These experiments had the objective to verify that analytical data obtained from real matrices were under control in a specific range of acceptance 70-130%. This range is a little different from ICV's acceptability (80-120%) because of the matrix effects. Real matrices were spiked with a low concentration of standard and were analysed to evaluate recovery of all compounds of interest. In these experiments, the spike was made with six target compounds: methanol, ethanol, isopropanol, acetonitrile, ethyl acetate and toluene.

The Appendix, section A7 shows the results obtained in all tests carried out on LCS, M, MS and MSD.

The following tables show the results

	METHANOL	ETHANOL	ISOPROPANOL	ACETONITRILE	ETHYL ACETATE	TOLUENE
DIMENSION	20.00	20.00	20.00	20.00	20.00	20.00
AVERAGE VALUE	107.34	113.80	105.61	100.56	105.28	104.21
STANDRD DEVIATION	11.14	9.94	9.86	10.22	8.10	10.81
RSD%	10.37	8.73	9.34	10.16	7.70	10.37

Table 24: Recovery for LCS obtained after 20 tests

	METHANOL	ETHANOL	ISOPROPANOL	ACETONITRILE	ETHYL ACETATE	TOLUENE
DIMENSION	20.00	20.00	20.00	20.00	20.00	20.00
AVERAGE VALUE	117.28	120.87	117.07	115.78	84.59	113.79
STANDRD DEVIATION	24.82	20.42	25.82	25.61	18.62	26.32
RSD %	21.16	16.89	22.05	22.12	22.02	23.13

Table 25: Recovery for MS obtained after 20 tests

	METHANOL	ETHANOL	ISOPROPANOL	ACETONITRILE	ETHYL ACETATE	TOLUENE
DIMENSION	20.00	20.00	20.00	20.00	20.00	20.00
AVERAGE VALUE	116.91	122.61	118.45	115.59	85.60	114.45
STANDRD DEVIATION	24.26	17.85	24.15	24.59	16.96	24.52
RSD %	20.75	14.56	20.39	21.27	19.81	21.42

Table 26: Recovery for MSD obtained after 20 tests

MS, MSD and LCS showed that recovery are acceptable for analysis and the method was successfully applied for the analysis of real matrices.

CONCLUSIONS

The topic of this work was the development of an analytical method for the determination of volatile organic compounds (VOC) in waste recovered solvents from a pharmaceutical manufacturing plant.

This analytical method was developed in compliance with official method EPA 8015D and shows some advantage in term of specificity and reduction of time analysis.

In this work new compounds, other than those listed in official method EPA 8015D, were introduced:

- Methyl Acetate
- Dichloromethane
- Hexane
- Tetrahydrofuran
- Cyclohexane
- Isopropyl Acetate
- Heptane
- 1,4-Dioxane
- Cyclopentyl Methyl Ether
- Butyl Acetate
- N,N-Dimethylformamide
- 2-Butanol
- 1-Butanol
- Octane

The results obtained in this work demonstrated the validity of the analytical conditions for all examined compounds. Official method EPA 8015D is restricted to no halogenated organic compounds, but this study showed good results also for dichloromethane.

The pentane was a good solvent for all compounds and its peak appeared separated from all compounds of interest. In this method the use of a column with a length of 60 m, as an alternative to those recommended in the method EPA 8015D with a length of 30m, showed a good chromatographic separation and it was preferential for the application of the method for the determination of numerous compounds.

This new approach with analysis of samples of waste and recovered solvents is according with customer's requests who are looking for laboratories able to analyse complex matrices, with the presence of a high number of molecules present in different concentration, reducing time and costs.

The synergic use of GC-MS with headspace injection, or direct injection, for qualitative analysis and GC-FID for quantitative analysis allowed the development of a method with advantages in term of specific identification and the quantification of different molecules.

This developed analytical method was successfully applied for VOC determination in waste and recovered solvents and it is very useful in the "Solvent Recovery" plant to control processes and to make correlations between the composition of different matrices (waste solvents, recovered solvents and wastewaters). Quantitative approach with GC-FID allows to determinate the composition of samples of waste solvents, products "in process" and final products (pure solvents or mixture).

The following table shows some typical determinations of different compounds in waste and recovered solvents.

Waste solvents are complex matrices with different compound, while recovered solvents are matrices characterized by purified mix or pure solvents.

	Waste Solvents_sample 1	Waste Solvents_sample 2	Waste Solvents_sample 3	Waste Solvents_sample 4
COMPOUNDS	mg/L	mg/L	mg/L	mg/L
METHANOL	13059.9	6989.15	123472	84511.30
ETHANOL			28529.4	53067.1
ISOPROPANOL	12543.1		19728.2	62288.2
ACETONE	27174	315613	203656	145403
ACETONITRILE			10105.7	18379.2
METHYL ACETATE		260997	39022.1	
DICHLOROMETANE		3582.7	57654.9	20241.8
HEXANE				
2-BUTANONE			1023.08	
ETHYL LACETATE	1469.12	2480.34	121236	185996
TETRAHYDROFURAN	13227.2		16725.9	11616.5
CYCLOHEXANE				3551.98
ISOPROPYL ACETATE			10046.4	5317.89
HEPTANE				503.13
1,4-DIOXAN				
CYCLOPENTYL METHYL ETHER				
TOLUENE	7595.8		76645.5	102757
BUTYL ACETATE				
DIMETHYLFORMAMIDE				
2-BUTANOL	12799.1			
1-BUTANOL				
PYRIDINE				
DIISOPROPYL ETHER			1392.67	
OCTANE				
Totals (mg/L)	87868.22	589662.19	709237.85	693633.10

Table n. 27: Typical composition of waste solvents

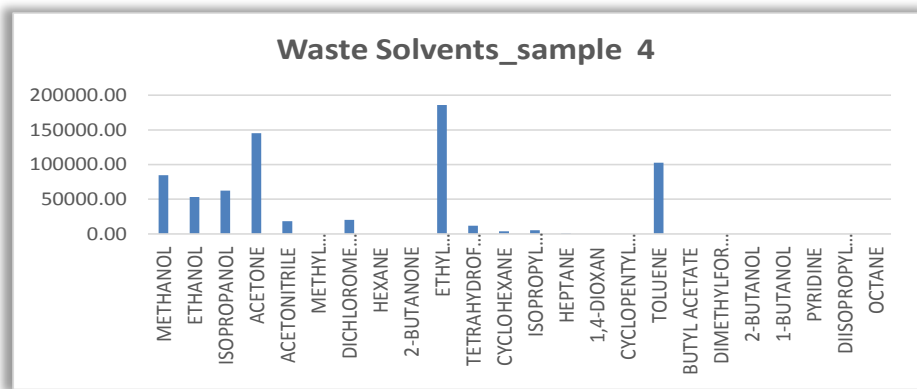
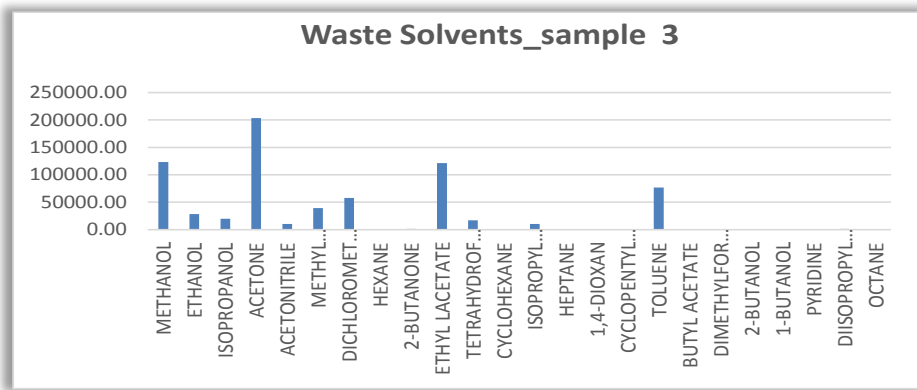
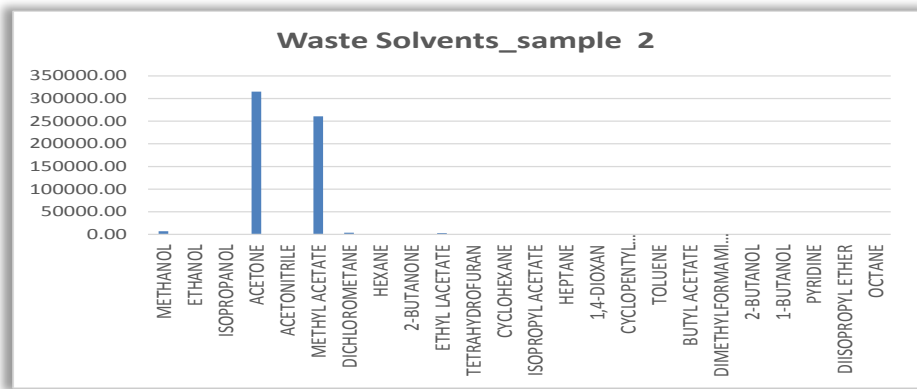
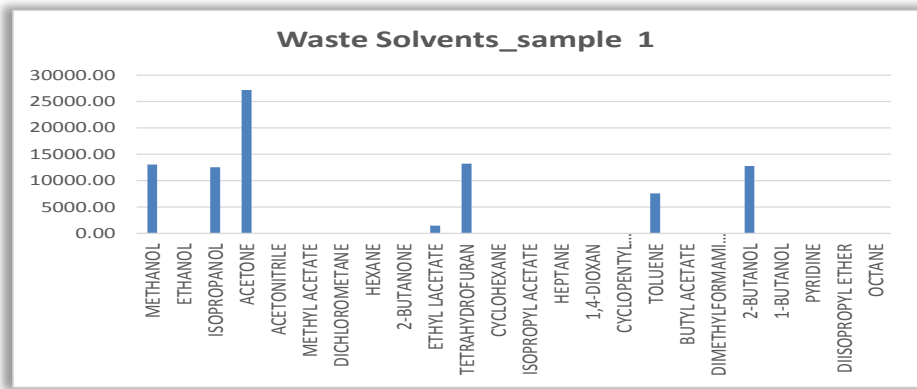


Figure 16: Plots of the compositions of 4 samples of waste solvents

	Recovered Solvents_sample 1	Recovered Solvents_sample 2	Recovered Solvents_sample 3	Recovered Solvents_sample 4
COMPOUNDS	mg/L	mg/L	mg/L	mg/L
METHANOL	164545		36406.5	6781.24
ETHANOL	193499			867.8
ISOPROPANOL	77157.1	163325	12566	
ACETONE	6679.08	225132	65806.8	804158
ACETONITRILE				
METHYL ACETATE				
DICHLOROMETANE			15180.8	
HEXANE				
2-BUTANONE				
ETHYL LACETATE	21570		562348	4930.53
TETRAHYDROFURAN			878.05	
CYCLOHEXANE	5479.53	1835.52		
ISOPROPYL ACETATE				
HEPTANE				
1,4-DIOXAN				
CYCLOPENTYL METHYL ETHER				
TOLUENE	25419.7	188737	2863.1	
BUTYL ACETATE			2472.43	
DIMETHYLFORMAMIDE				
2-BUTANOL				63791.7
1-BUTANOL				
PYRIDINE				
DIISOPROPYL ETHER				
OCTANE				
Totals (mg/L)	494349.41	579029.52	698521.68	880529.27

Table n. 28: Typical composition of recovered solvents

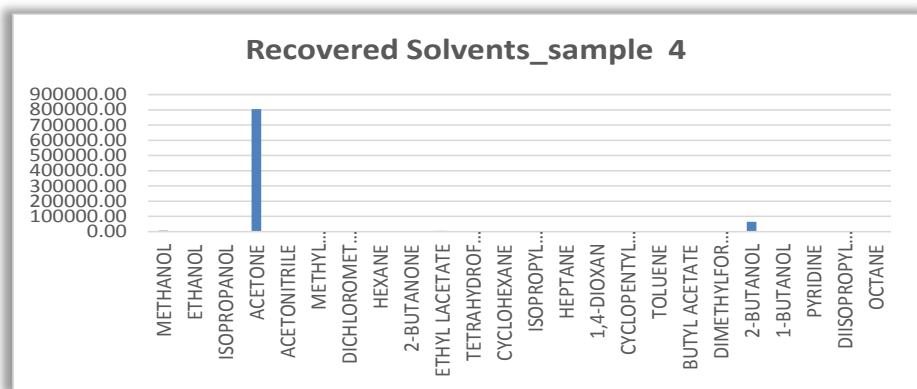
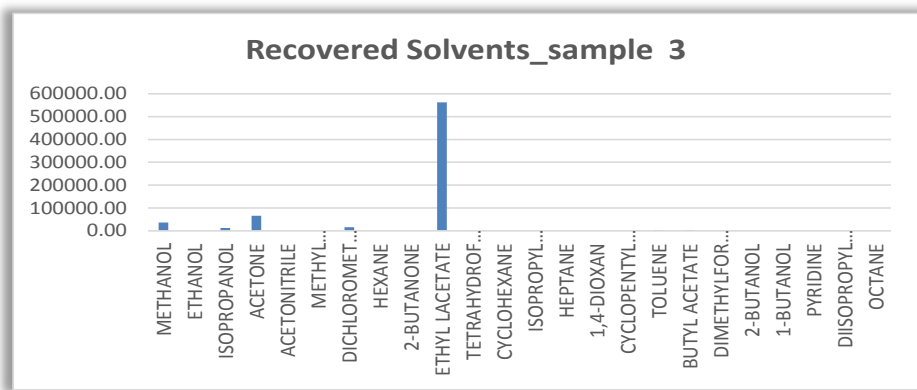
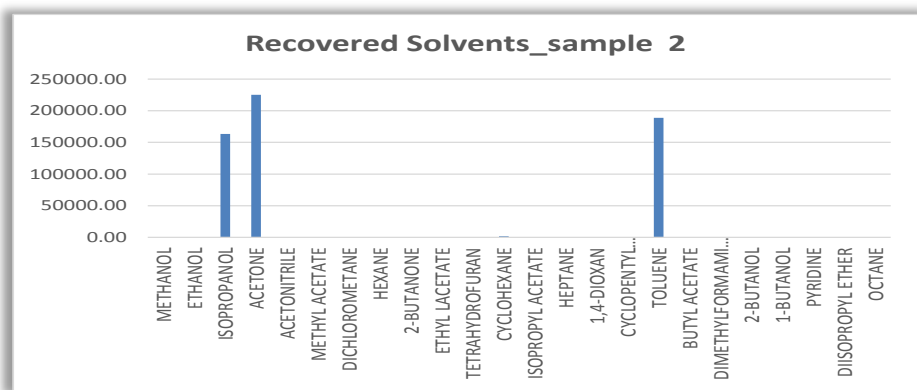
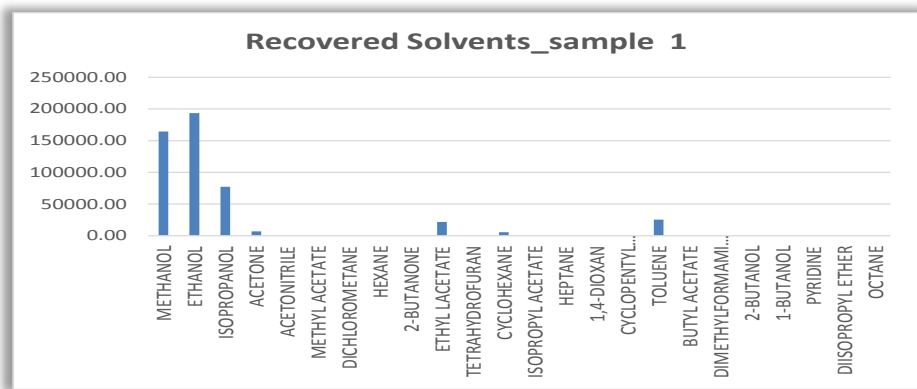


Figure 17: Plots of the compositions of 4 samples of recovered solvents

This method is successfully applied for many industrial implications:

- the analysis of waste solvents is important for the characterization of the samples and to assure the compliance with laws
- the analysis of products "in process" plays a pivotal role for the control of the efficiency of the separation: for example the analysis could put in evidence the presence of azeotropes during the distillation
- the analysis of the final products is determinant for the distribution of the product on the market
- the analysis of waste of production, for example a wastewater with a high concentration of solvents before the depuration, is important to control the pre-treatment to reduce the solvent's content in wastewater and to preserve the industrial depurator
- this method allows to execute analysis in a short time in accordance with industrial needs to have an immediate control of all the steps of the processes.

Future developments of this analytical method are always opened to determination of different compounds, for example: tert-butanol, hexamethyldisiloxane, triethylamine, methyl isobutyl ketone, o-xylene, m-xylene, p-xylene.

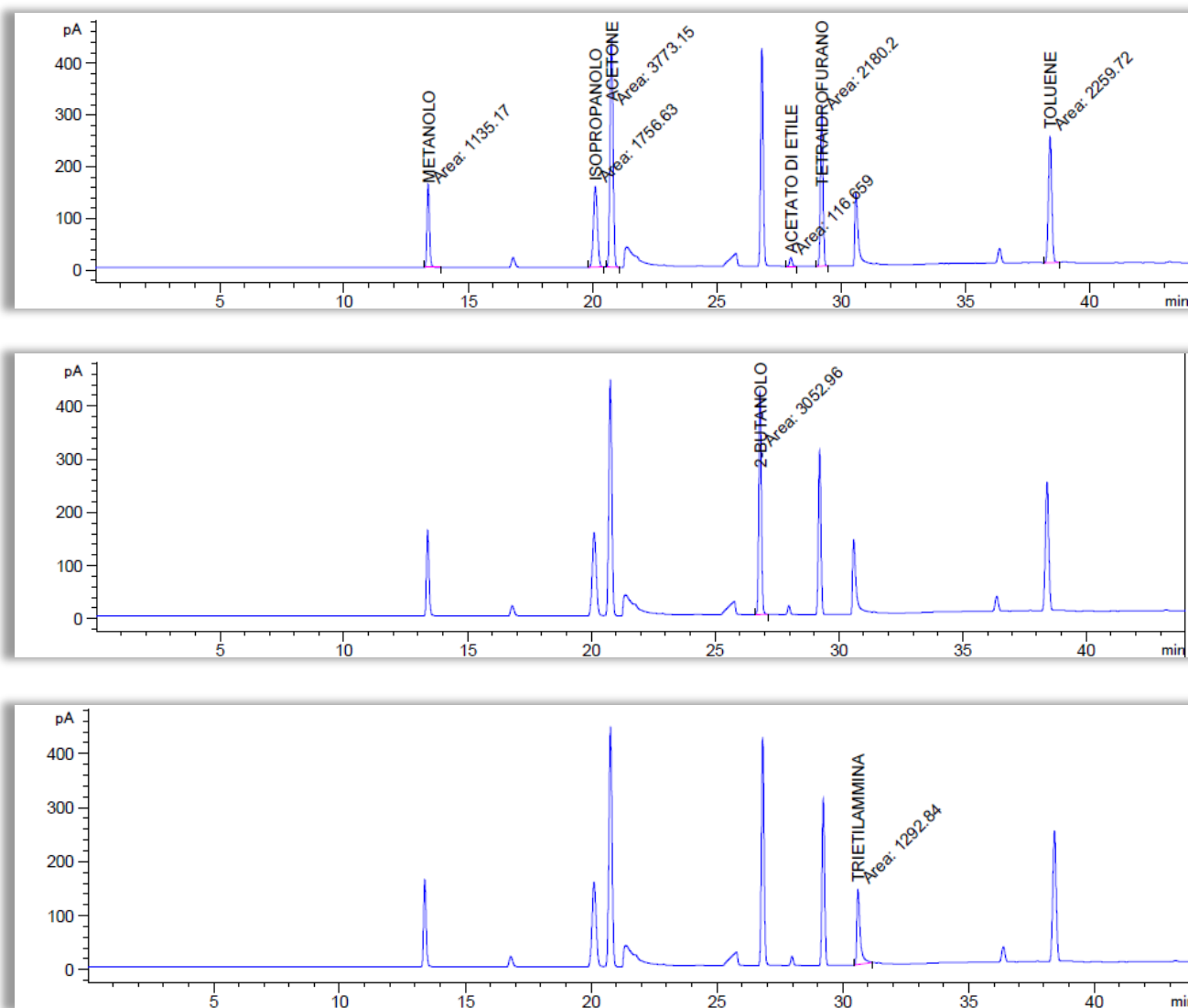


Figure 18: Chromatograms of a waste solvent with the presence of triethylamine

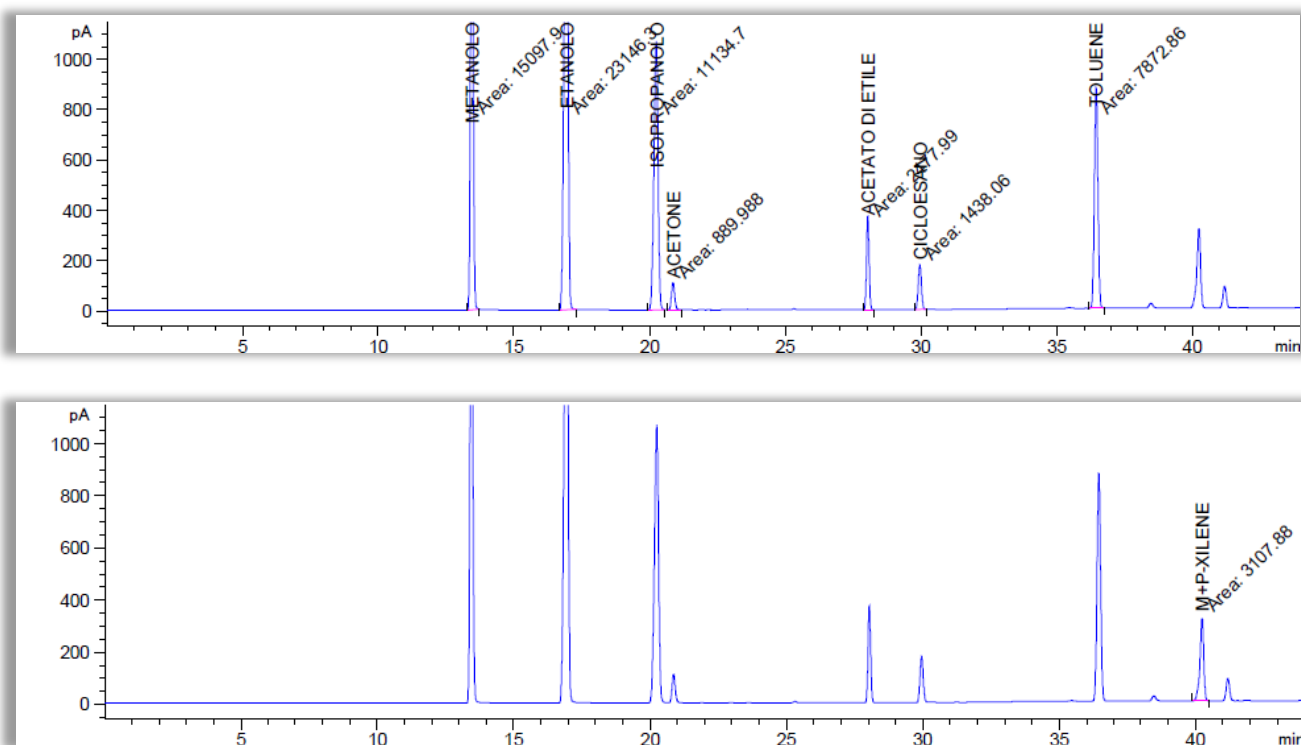


Figure 19: Chromatograms of a waste solvent with the presence of m-xylene and p-xylene

The statistical approach described in this work could be considered as an additional quality assurance used with method EPA 8015D. The development of this method with all the statistical determinations was important to assure a good quality of analytical data. The statistical approach of this study allowed to the analysts to have a good control of the response of the analysis in term of precision and accuracy for all the analytes. The wastes are very complex matrices and the evaluation of recovery is fundamental for the correct expression of the analytical data.

In this work the statistical approach gave the opportunity to understand possible differences between the analytes: each compound was monitored for a long period of time and the results of recovery obtained from ICV, elaborated with ANOVA test, confirmed that all the analytes can be classified in three groups to explain the total variance. This approach is necessary to demonstrate the validity of the analysis carried out in short time and without any pre-treatment of the samples.

APPENDIX

A1. Calibration curves

MEYHANOL	
Conc (mg/L)	Peak Area
1510	99.03934
6040	618.08781
15100	1998.02847
22650	3066.503989
30200	4356.41784

ETHANOL	
Conc (mg/L)	Peak Area
1469.4	199.50518
5877.6	1246.49988
14694	3514.46794
22041	4962.54585
29388	6578.41416

ISOPROPANOL	
Conc (mg/L)	Peak Area
1486.6	208.57998
5946.4	1237.07032
14866	3899.57394
22299	5514.39092
29732	7648.94753

ACETONE	
Conc (mg/L)	Peak Area
1472.2	217.40515
5888.8	1042.07697
14722	3013.10309
22083	4609.35109
29444	6327.02956

ACETONITRILE	
Conc (mg/L)	Peak Area
1516.6	177.37162
6066.4	1070.01883
15166	3407.85297
22749	5006.57719
30332	7322.4987

METHYL ACETATE	
Conc (mg/L)	Peak Area
1745.2	136.41474
6980.8	724.74055
17452	2267.45021
26178	3207.16904
34904	4686.24589

DICHLOROMETANE	
Conc (mg/L)	Peak Area
2176	75.85747
8704	440.40489
21760	1166.66634
32640	2001.67028
43520	2539.84947

HEXANE	
Conc (mg/L)	Peak Area
1255	286.40729
5020	1516.36292
12550	4004.41419
18825	5594.20959
25100	7710.93644

2-BUTANONE	
Conc (mg/L)	Peak Area
1280.6	235.41515
5122.4	1254.4084
12806	3764.35421
19209	5452.88801
25612	7407.34518

ETHYLACETATE	
Conc (mg/L)	Peak Area
1758.2	289.96897
7032.8	1180.43054
17582	3422.84588
26373	4942.15874
35164	6858.82133

TETRAHYDROFURAN	
Conc (mg/L)	Peak Area
931.2	156.48483
3724.8	825.24937
9312	2405.4199
13968	3439.60198
18624	4854.69105

CYCLOHEXANE	
Conc (mg/L)	Peak Area
1476.6	362.50913
5906.4	1817.3864
14766	4960.7652
22149	7109.79997
29532	9782.00064

ISOPROPILACETATE	
Conc (mg/L)	Peak Area
1711.8	274.20331
6847.2	1428.3141
17118	4050.32477
25677	5936.01996
34236	8100.23507

HEPTANE	
Conc (mg/L)	Peak Area
1345.8	354.47705
5383.2	1826.87607
13458	5000.33761
20187	7010.44203
26916	9709.13447

1,4-DIOXANE	
Conc (mg/L)	Peak Area
2029.4	230
8117.6	1207.94165
20294	3605.09985
30441	5071.2391
40588	7097.3501

CYCLO PENTYL METHYL ETHER	
Conc (mg/L)	Peak Area
1694.4	375.10716
6777.6	2079.13783
16944	5884.1299
25416	8249.42954
33888	11303.8

TOLUENE	
Conc (mg/L)	Peak Area
1703.6	595.74809
6814.4	3073.9899
17036	8874.49762
25554	12368.4
34072	17087.2

BUTHYL ACETATE	
Conc (mg/L)	Peak Area
1740.2	356.02305
6960.8	1939.76129
17402	5478.71433
26103	7606.73477
34804	10467

DIMETHYLFORMAMIDE	
Conc (mg/L)	Peak Area
1929	222.49948
7716	1236.31026
19290	3345.94849
28935	4856.46435
38580	6831.08788

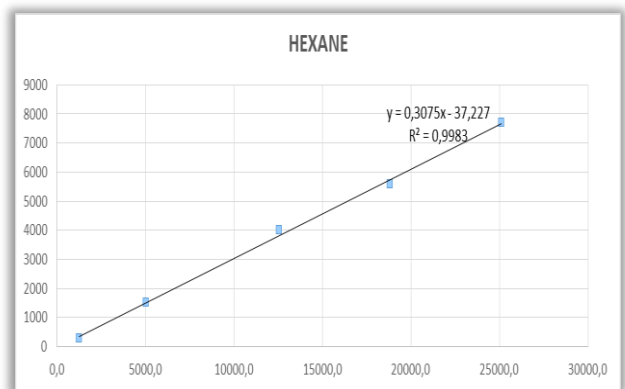
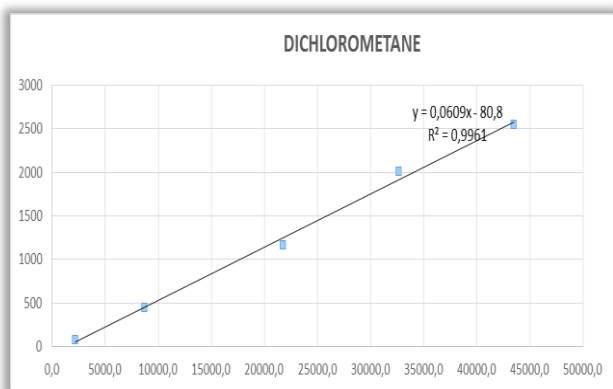
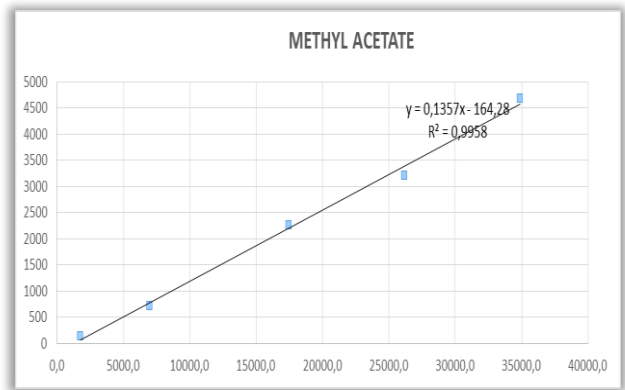
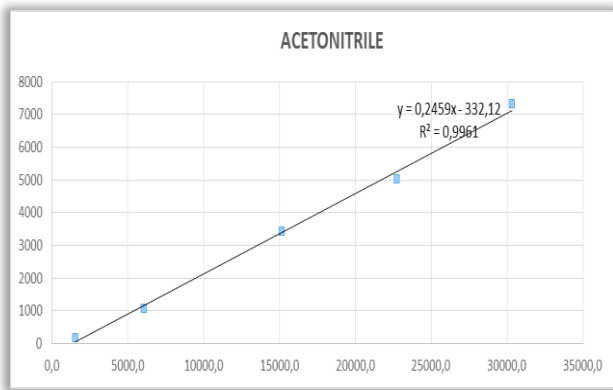
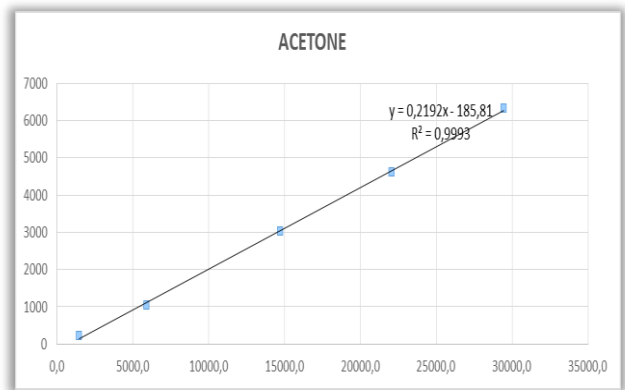
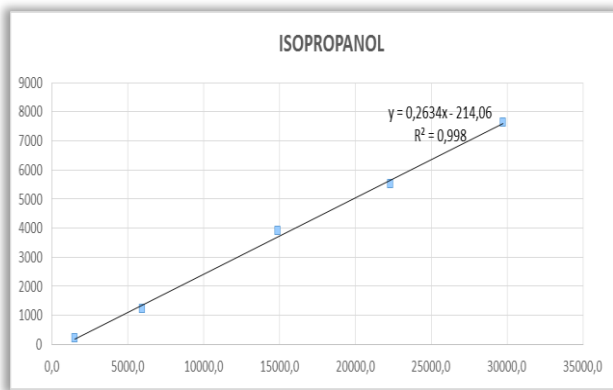
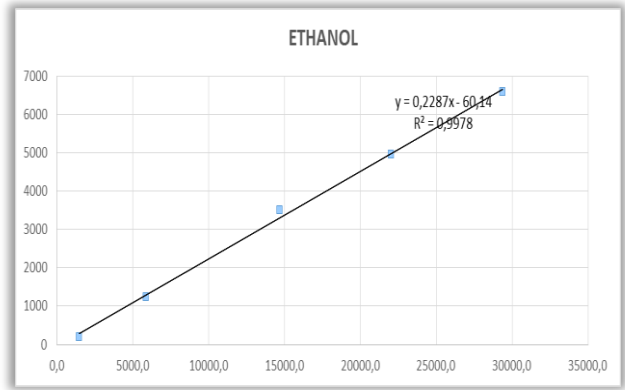
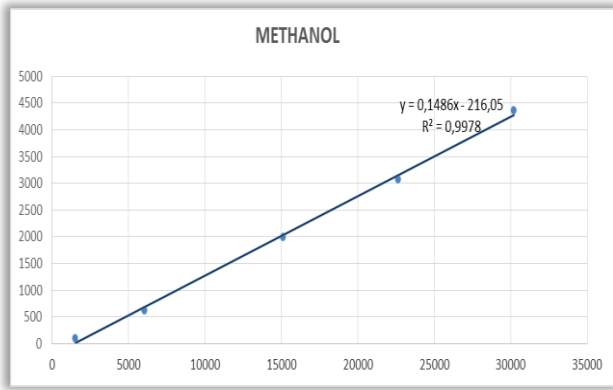
2-BUTANOL	
Conc (mg/L)	Peak Area
1543.6	253.5717
6174.4	1391.18762
15436	3701.89966
23154	6020.95752
30872	8236.18945

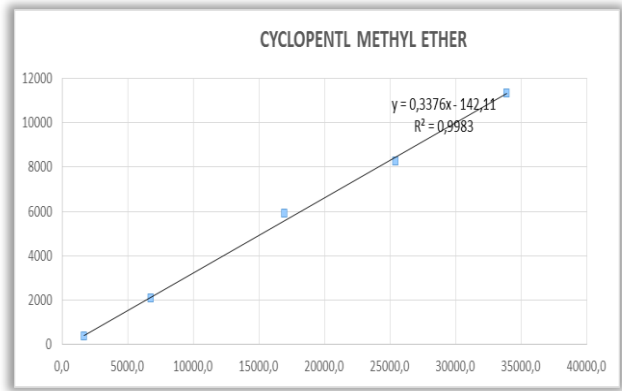
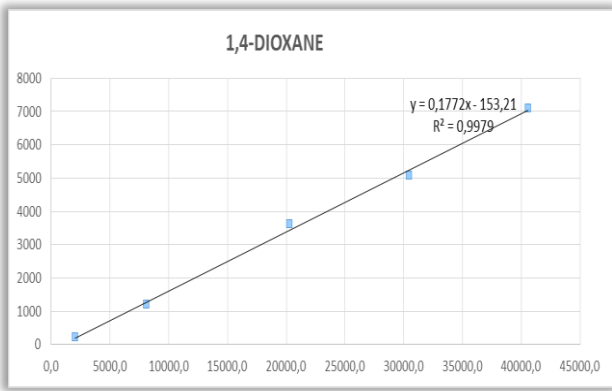
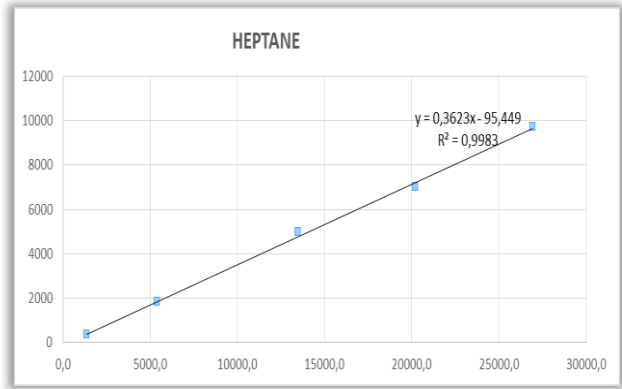
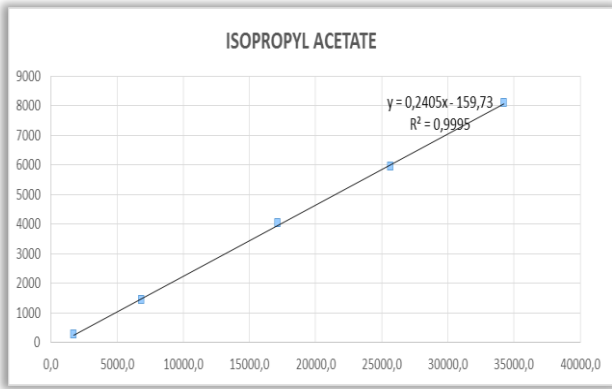
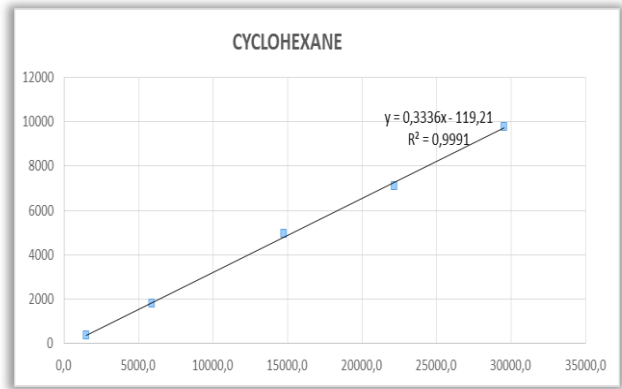
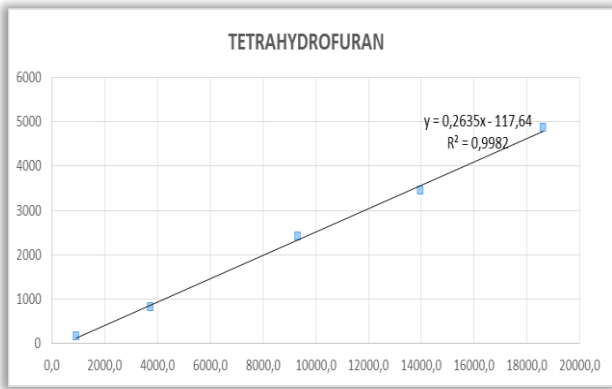
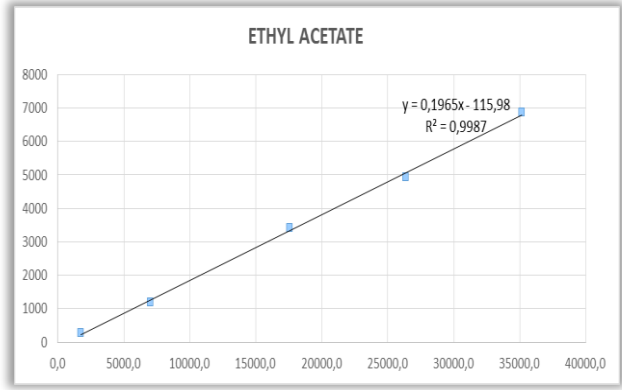
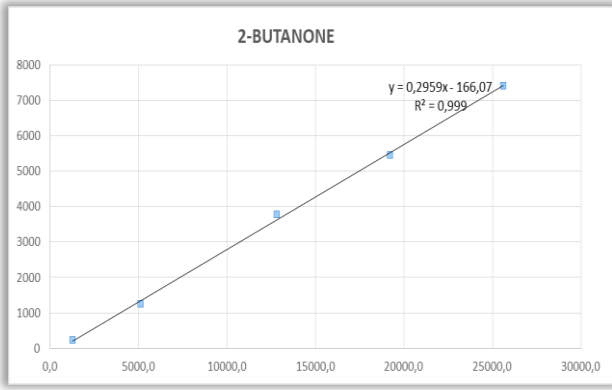
1-BUTANOL	
Conc (mg/L)	Peak Area
1567.2	287.90253
6268.8	1603.1842
15672	4275.42432
23508	6886.77393
31344	9402.22461

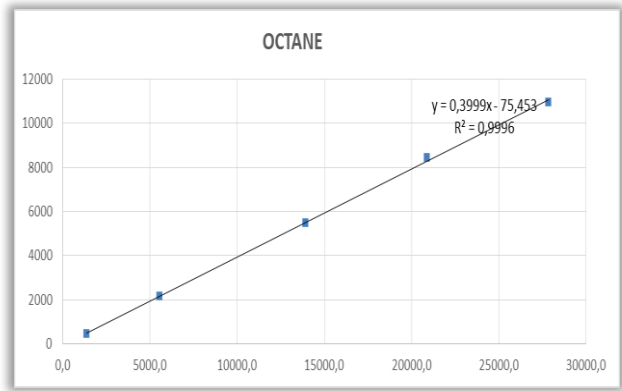
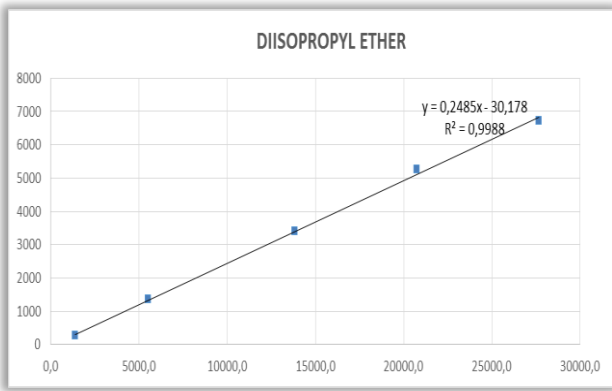
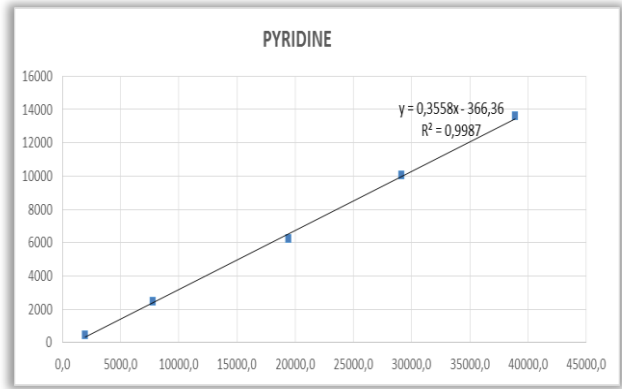
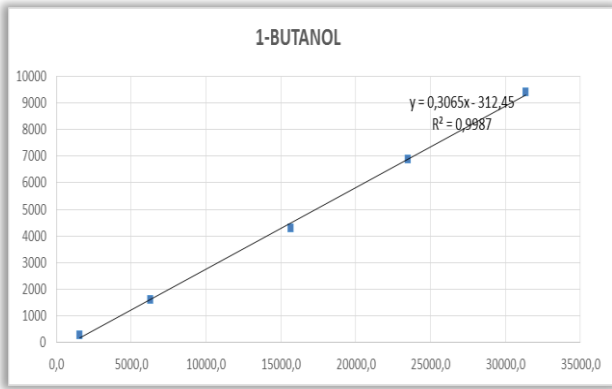
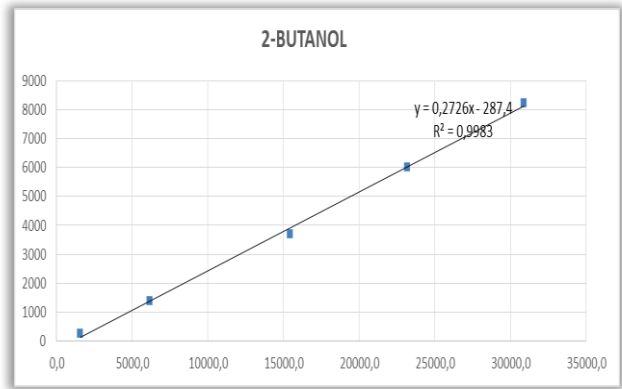
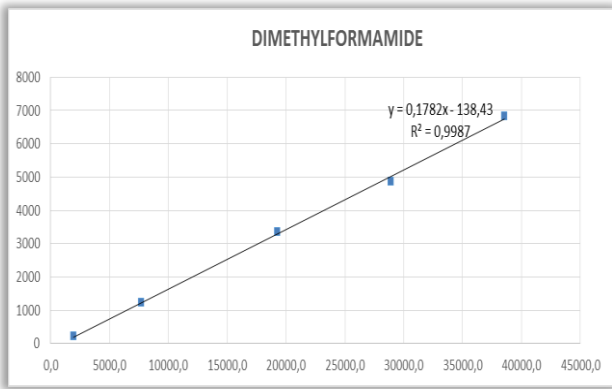
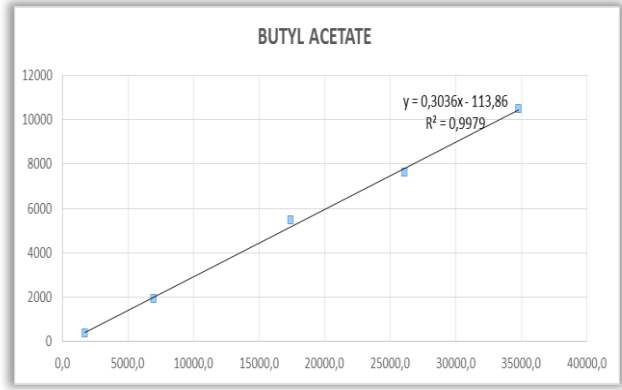
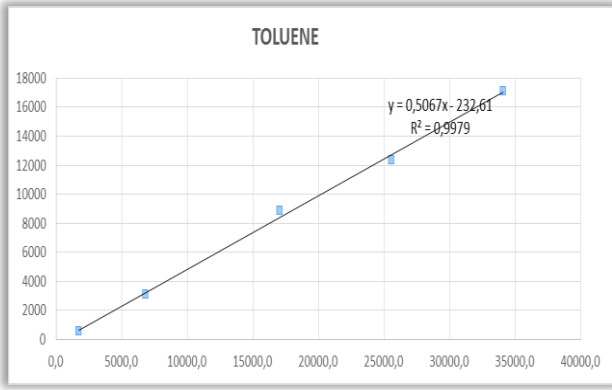
PYRIDINE	
Conc (mg/L)	Peak Area
1943.8	455.33185
7775.2	2446.01127
19438	6220.23998
29157	10023.2
38876	13601.7

DIISOPROPYL ETHER	
Conc (mg/L)	Peak Area
1382.4	268.33295
5529.6	1349.2771
13824	3410.05542
20736	5264.50098
27648	6731.11377

OCTANE	
Conc (mg/L)	Peak Area
1392	439.51883
5568	2163.5874
13920	5476.95166
20880	8418.46094
27840	10954.2







A2. Determination of LOQ and LOD

METHANOL		n.	Peak Area
Conc (mg/L)	Peak Area	1	70.91
308.00	68.80	2	67.18
615.00	153.72	3	70.09
923.00	226.10	4	75.54
1230.00	294.67	5	70.58
1538.00	389.79	6	72.61
		7	71.20
		8	72.53
Slope	0.25	9	76.47
Intercept	-8.34	10	70.64
LOD mg/L	31.72	STD Dev.	2.69
LOQ (mg/L)	105.74	Mean	71.78

ISOPROPANOL		n.	Peak Area
Conc (mg/L)	Peak Area	1	76.69
311.00	71.05	2	74.58
622.00	160.94	3	77.26
933.00	261.32	4	77.39
1244.00	349.65	5	78.70
1555.00	473.78	6	75.51
		7	81.90
		8	81.40
Slope	0.32	9	76.39
Intercept	-34.90	10	78.60
LOD mg/L	22.23	STD Dev.	2.37
LOQ (mg/L)	74.10	Mean	77.84

METHYL ACETATE		n.	Peak Area
Conc (mg/L)	Peak Area	1	73.02
346.00	67.33	2	69.91
692.00	151.84	3	74.61
1038.00	230.87	4	70.33
1384.00	298.10	5	73.61
1730.00	390.41	6	69.72
		7	71.73
		8	75.49
Slope	0.23	9	76.67
Intercept	-10.02	10	72.32
LOD mg/L	31.33	STD Dev.	2.39
LOQ (mg/L)	104.42	Mean	72.74

HEXANE		n.	Peak Area
Conc (mg/L)	Peak Area	1	148.18
260.00	152.29	2	135.50
519.00	334.10	3	148.88
779.00	534.61	4	141.19
1038.00	661.19	5	145.31
1298.00	874.54	6	141.90
		7	152.19
		8	153.03
Slope	0.68	9	130.42
Intercept	-20.35	10	146.93
LOD mg/L	31.68	STD Dev.	7.21
LOQ (mg/L)	105.62	Mean	144.35

ETHANOL		n.	Peak Area
Conc (mg/L)	Peak Area	1	105.30
315.00	100.03	2	99.77
631.00	216.25	3	111.24
946.00	338.93	4	104.38
1261.00	457.39	5	107.79
1576.00	572.23	6	107.69
		7	109.07
		8	113.89
Slope	0.38	9	108.62
Intercept	-18.77	10	108.79
LOD mg/L	30.82	STD Dev.	3.86
LOQ (mg/L)	102.75	Mean	107.65

ACETONITRILE		n.	Peak Area
Conc (mg/L)	Peak Area	1	111.50
317.00	104.03	2	103.09
633.00	227.87	3	119.60
950.00	354.76	4	110.38
1266.00	482.00	5	114.14
1582.00	592.22	6	113.03
		7	116.94
		8	120.05
Slope	0.39	9	114.90
Intercept	-17.25	10	114.29
LOD mg/L	37.81	STD Dev.	4.90
LOQ (mg/L)	126.04	Mean	113.79

DICHLOROMETANE		n.	Peak Area
Conc (mg/L)	Peak Area	1	53.89
527.00	50.31	2	50.57
1054.00	109.05	3	54.64
1581.00	168.24	4	50.66
2108.00	218.93	5	53.73
2635.00	273.93	6	55.32
		7	53.73
		8	54.13
Slope	0.11	9	50.55
Intercept	-3.04	10	51.64
LOD mg/L	52.08	STD Dev.	1.84
LOQ (mg/L)	173.59	Mean	52.89

2-BUTANONE		n.	Peak Area
Conc (mg/L)	Peak Area	1	150.43
327.00	137.24	2	141.49
654.00	291.66	3	155.08
982.00	448.64	4	147.15
1309.00	588.27	5	150.07
1637.00	775.89	6	146.09
		7	148.90
		8	156.04
Slope	0.48	9	155.96
Intercept	-23.50	10	147.99
LOD mg/L	29.35	STD Dev.	4.70
LOQ (mg/L)	97.85	Mean	149.92

ETHYL ACETATE		n.	Peak Area
Conc (mg/L)	Peak Area	1	107.69
360.00	99.85	2	104.90
721.00	222.33	3	110.14
1081.00	347.98	4	109.46
1441.00	441.51	5	109.44
1802.00	583.67	6	112.41
		7	108.16
		8	116.58
Slope	0.33	9	107.84
Intercept	-16.92	10	105.73
LOD mg/L	30.59	STD Dev.	3.36
LOQ (mg/L)	101.98	Mean	109.24

CYCLOHEXANE		n.	Peak Area
Conc (mg/L)	Peak Area	1	137.49
302.00	137.65	2	131.40
603.00	312.17	3	145.79
905.00	513.74	4	139.83
1206.00	681.15	5	136.04
1508.00	850.36	6	148.53
		7	139.96
		8	149.01
Slope	0.60	9	141.75
Intercept	-39.49	10	138.52
LOD mg/L	28.20	STD Dev.	5.60
LOQ (mg/L)	94.01	Mean	140.83

HEPTANE		n.	Peak Area
Conc (mg/L)	Peak Area	1	162.17
267.00	155.24	2	159.97
533.00	316.43	3	167.07
800.00	490.62	4	156.07
1067.00	704.10	5	163.35
1333.00	806.27	6	166.76
		7	168.56
		8	172.30
Slope	0.63	9	183.43
Intercept	-12.54	10	157.99
LOD mg/L	37.73	STD Dev.	7.97
LOQ (mg/L)	125.78	Mean	165.77

CYCLOPENTYL METHYL ETHER		n.	Peak Area
Conc (mg/L)	Peak Area	1	192.48
344.00	174.15	2	185.25
688.00	389.42	3	192.43
1032.00	591.06	4	186.48
1376.00	787.24	5	196.46
1720.00	1019.51	6	197.48
		7	199.67
		8	204.08
Slope	0.61	9	192.73
Intercept	-34.29	10	194.24
LOD mg/L	28.11	STD Dev.	5.69
LOQ (mg/L)	93.71	Mean	194.13

BUTYL ACETATE		n.	Peak Area
Conc (mg/L)	Peak Area	1	151.11
354.00	140.79	2	147.13
707.00	308.47	3	164.54
1061.00	482.35	4	152.52
1415.00	651.87	5	157.12
1769.00	805.73	6	160.82
		7	165.54
		8	159.65
Slope	0.47	9	158.25
Intercept	-24.05	10	151.11
LOD mg/L	38.95	STD Dev.	6.14
LOQ (mg/L)	129.84	Mean	156.78

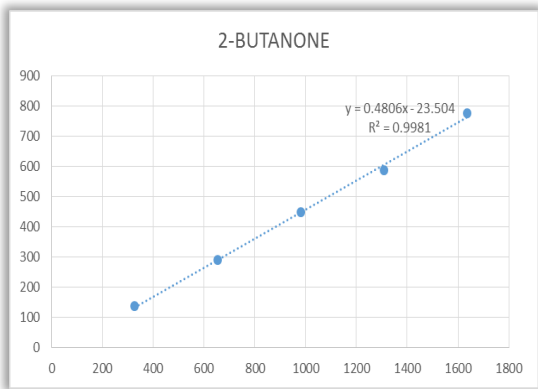
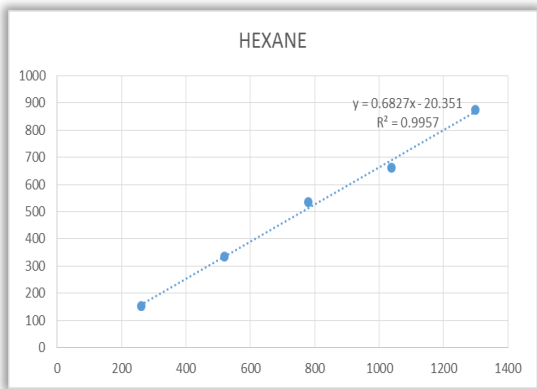
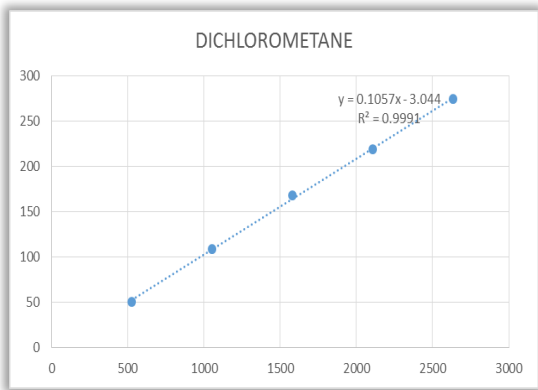
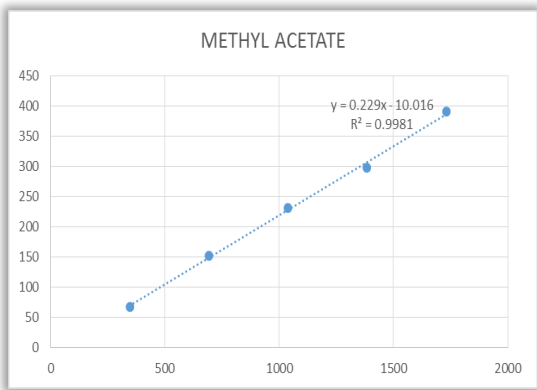
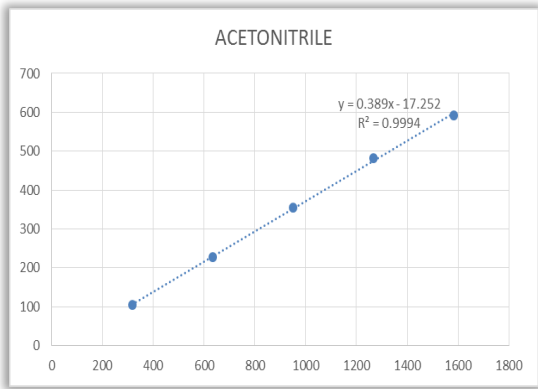
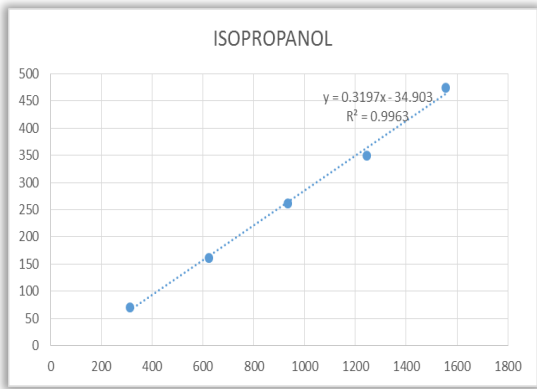
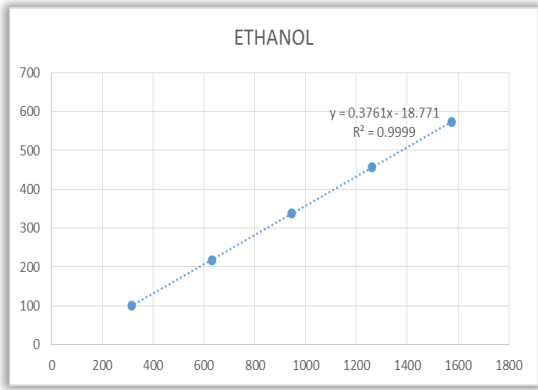
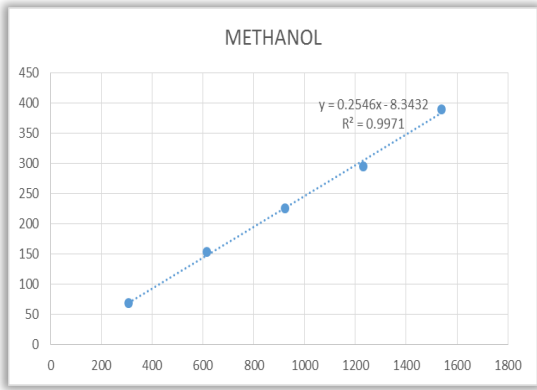
TETRAHYDROFURAN		n.	Peak Area
Conc (mg/L)	Peak Area	1	119.69
345.00	107.83	2	114.81
691.00	251.82	3	123.06
1036.00	400.42	4	114.87
1382.00	528.20	5	119.94
1727.00	678.40	6	127.09
		7	125.27
		8	124.64
Slope	0.41	9	120.90
Intercept	-31.80	10	120.08
LOD mg/L	30.09	STD Dev.	4.11
LOQ (mg/L)	100.30	Mean	121.04

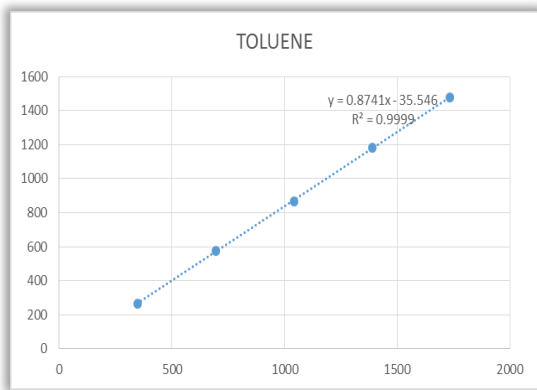
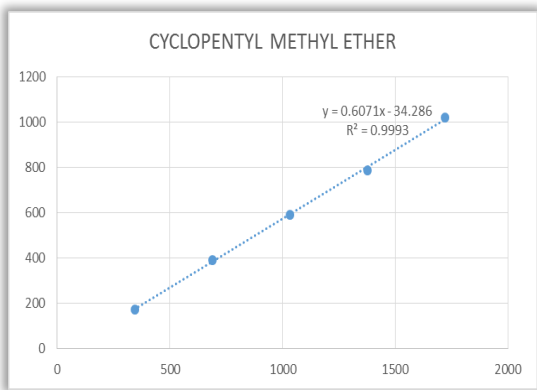
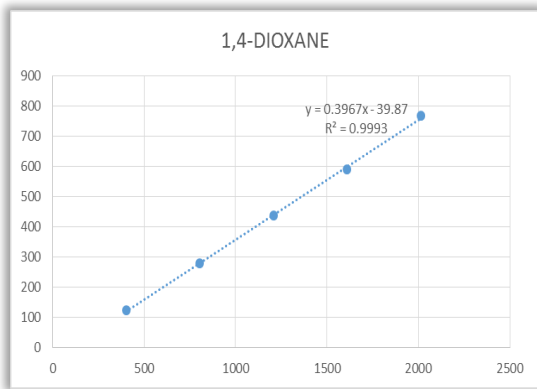
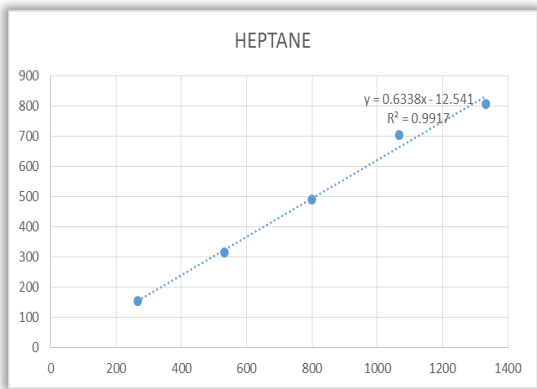
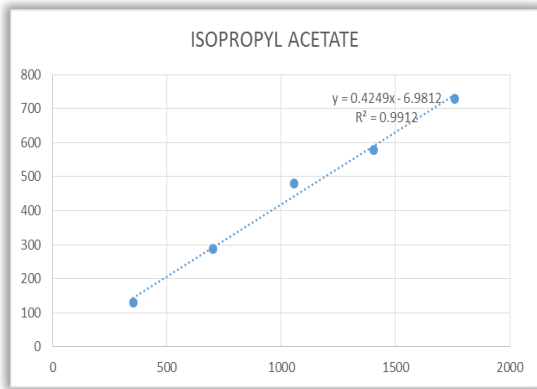
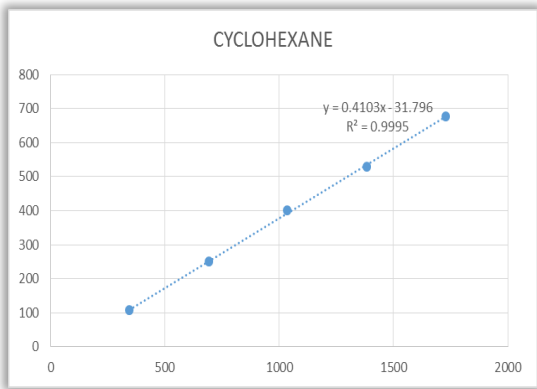
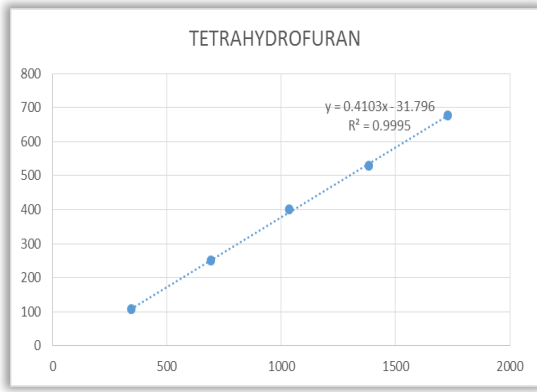
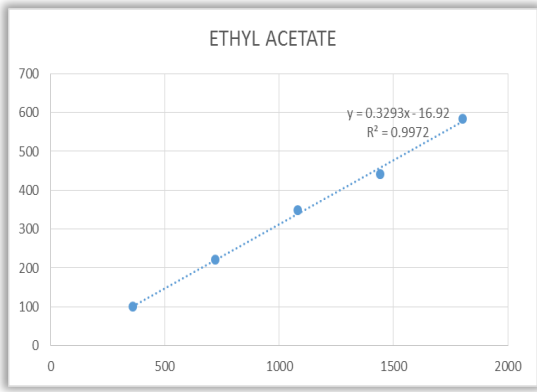
ISOPROPYL ACETATE		n.	Peak Area
Conc (mg/L)	Peak Area	1	131.23
352.00	128.87	2	130.66
703.00	287.91	3	147.38
1055.00	480.45	4	137.36
1406.00	578.60	5	141.39
1758.00	730.32	6	147.07
		7	153.62
		8	148.25
Slope	0.42	9	114.27
Intercept	-6.98	10	130.82
LOD mg/L	83.04	STD Dev.	11.76
LOQ (mg/L)	276.81	Mean	138.21

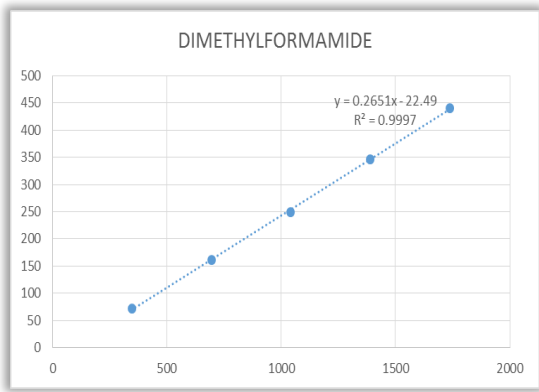
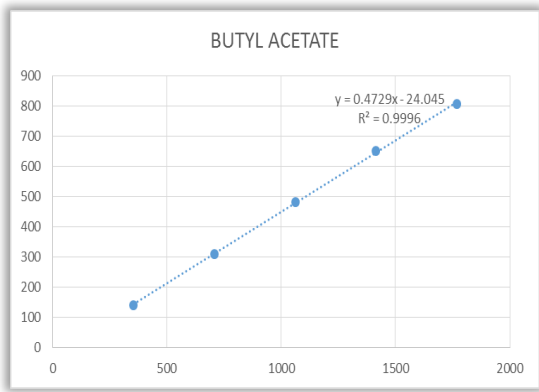
1,4-DIOXANE		n.	Peak Area
Conc (mg/L)	Peak Area	1	138.54
403.00	124.56	2	123.91
805.00	278.39	3	144.56
1208.00	436.52	4	142.38
1611.00	590.21	5	141.43
2013.00	767.30	6	141.81
		7	145.89
		8	147.45
Slope	0.40	9	144.87
Intercept	-39.87	10	141.11
LOD mg/L	50.01	STD Dev.	6.61
LOQ (mg/L)	166.70	Mean	141.20

TOLUENE		n.	Peak Area
Conc (mg/L)	Peak Area	1	286.16
347.00	266.91	2	275.64
694.00	575.12	3	306.88
1041.00	868.07	4	291.01
1388.00	1181.92	5	295.13
1735.00	1480.14	6	297.70
		7	299.48
		8	308.61
Slope	0.87	9	292.13
Intercept	-35.55	10	295.62
LOD mg/L	32.98	STD Dev.	9.61
LOQ (mg/L)	109.92	Mean	294.84

DIMETHYLFORMAMIDE		n.	Peak Area
Conc (mg/L)	Peak Area	1	77.83
347.00	71.77	2	80.50
695.00	161.35	3	86.39
1042.00	249.40	4	79.93
1390.00	347.04	5	82.93
1737.00	439.59	6	84.92
		7	88.53
		8	82.23
Slope	0.27	9	83.77
Intercept	-22.49	10	83.77
LOD mg/L	35.76	STD Dev.	3.16
LOQ (mg/L)	119.19	Mean	83.08







A3. Experiments of repeatability on Certified Matrix

Experimental Data	Conc (mg/L)_Level 1	Conc (mg/L)_Level 2	Conc (mg/L)_Level 3
EXPERIMENTS	METHANOL	METHANOL	METHANOL
1	2843.04	10644.30	13592.41
2	2850.63	10651.90	14039.24
3	2931.65	10713.92	14050.63
4	2953.16	11353.16	14882.28
5	2954.43	12007.59	15454.43
6	2982.28	12192.41	15726.58
7	3012.66	12294.94	16467.09

Spike (mg/L)	2531.65	12658.23	16911.39
---------------------	---------	----------	----------

Mean	2932.55	11408.32	14887.52
STD Dev.	63.92	752.90	1050.22
CV%	2.18	6.60	7.05
Variance	4086.27	566855.13	1102959.69

Recovery %	Recovery %_Level 1	Recovery %_Level 2	Recovery %_Level 3
EXPERIMENTS	METHANOL	METHANOL	METHANOL
1	112.30	84.09	80.37
2	112.60	84.15	83.02
3	115.80	84.64	83.08
4	116.65	89.69	88.00
5	116.70	94.86	91.38
6	117.80	96.32	92.99
7	119.00	97.13	97.37

Mean	115.84	90.13	88.03
STD Dev.	2.52	5.95	6.21
CV%	0.02	0.07	0.07
Variance	6.38	35.38	38.57

Z-SCORE	Conc (mg/L)_Level 1	Conc (mg/L)_Level 2	Conc (mg/L)_Level 3
EXPERIMENTS	METHANOL	METHANOL	METHANOL
1	-1.40	-1.01	-1.23
2	-1.28	-1.00	-0.81
3	-0.01	-0.92	-0.80
4	0.32	-0.07	0.00
5	0.34	0.80	0.54
6	0.78	1.04	0.80
7	1.25	1.18	1.50

Experimental Data	Conc (mg/L)_Level 1	Conc (mg/L)_Level 2	Conc (mg/L)_Level 3
EXPERIMENTS	ETHANOL	ETHANOL	ETHANOL
1	2664.13	11581.75	15007.60
2	2737.64	11676.81	15330.80
3	2792.14	11787.07	15338.40
4	2795.94	12515.84	15889.73
5	2826.36	13604.56	16342.21
6	2830.16	13918.88	16877.06
7	3095.06	13959.44	17794.68

Spike (mg/L)	2534.85	12674.27	16932.83
--------------	---------	----------	----------

Mean	2820.21	12720.62	16082.93
STD Dev.	134.31	1084.12	995.12
CV%	4.76	8.52	6.19
Variance	18038.71	1175315.56	990269.06

Recovery %	Recovery %_Level 1	Recovery %_Level 2	Recovery %_Level 3
EXPERIMENTS	ETHANOL	ETHANOL	ETHANOL
1	105.10	91.38	88.63
2	108.00	92.13	90.54
3	110.15	93.00	90.58
4	110.30	98.75	93.84
5	111.50	107.34	96.51
6	111.65	109.82	99.67
7	122.10	110.14	105.09

Mean	111.26	100.37	94.98
STD Dev.	5.30	8.55	5.88
CV%	0.05	0.09	0.06
Variance	28.07	73.17	34.54

Z-SCORE	Conc (mg/L)_Level 1	Conc (mg/L)_Level 2	Conc (mg/L)_Level 3
EXPERIMENTS	ETHANOL	ETHANOL	ETHANOL
1	-1.16	-1.05	-1.08
2	-0.61	-0.96	-0.76
3	-0.21	-0.86	-0.75
4	-0.18	-0.19	-0.19
5	0.05	0.82	0.26
6	0.07	1.11	0.80
7	2.05	1.14	1.72

Experimental Data	Conc (mg/L)_Level 1	Conc (mg/L)_Level 2	Conc (mg/L)_Level 3
EXPERIMENTS	ISOPROPANOL	ISOPROPANOL	ISOPROPANOL
1	2553.44	10527.99	13016.54
2	2601.78	10568.70	13998.73
3	2622.14	10623.41	14108.14
4	2661.58	11793.89	14891.86
5	2669.21	12075.06	15226.46
6	2723.92	12248.09	15863.87
7	3017.81	12363.87	16138.68

Spike (mg/L)	2544.53	12722.65	16997.46
---------------------	---------	----------	----------

Mean	2692.84	11457.29	14749.18
STD Dev.	153.21	845.65	1110.37
CV%	5.69	7.38	7.53
Variance	23472.07	715123.25	1232922.86

Recovery %	Recovery %_Level 1	Recovery %_Level 2	Recovery %_Level 3
EXPERIMENTS	ISOPROPANOL	ISOPROPANOL	ISOPROPANOL
1	100.35	82.75	76.58
2	102.25	83.07	82.36
3	103.05	83.50	83.00
4	104.60	92.70	87.61
5	104.90	94.91	89.58
6	107.05	96.27	93.33
7	118.60	97.18	94.95

Mean	105.83	90.05	86.77
STD Dev.	6.02	6.65	6.53
CV%	0.06	0.07	0.08
Variance	36.25	44.18	42.67

Z-SCORE	Conc (mg/L)_Level 1	Conc (mg/L)_Level 2	Conc (mg/L)_Level 3
EXPERIMENTS	ISOPROPANOL	ISOPROPANOL	ISOPROPANOL
1	-0.91	-1.10	-1.56
2	-0.59	-1.05	-0.68
3	-0.46	-0.99	-0.58
4	-0.20	0.40	0.13
5	-0.15	0.73	0.43
6	0.20	0.94	1.00
7	2.12	1.07	1.25

Experimental Data	Conc (mg/L)_Level 1	Conc (mg/L)_Level 2	Conc (mg/L)_Level 3
EXPERIMENTS	ACETONITRILE	ACETONITRILE	ACETONITRILE
1	2896.95	9936.39	12435.11
2	2907.12	9963.10	13160.31
3	2912.21	10792.62	14217.56
4	3008.91	10792.62	14718.83
5	3012.72	11016.54	14743.00
6	3041.98	11134.86	14936.39
7	3260.81	11215.01	15468.19

Spike (mg/L)	2544.53	12722.65	16997.46
--------------	---------	----------	----------

Mean	3005.82	10693.02	14239.91
STD Dev.	126.98	531.90	1072.35
CV%	4.22	4.97	7.53
Variance	16123.91	282912.54	1149926.44

Recovery %	Recovery %_Level 1	Recovery %_Level 2	Recovery %_Level 3
EXPERIMENTS	ACETONITRILE	ACETONITRILE	ACETONITRILE
1	113.85	78.10	73.16
2	114.25	78.31	77.43
3	114.45	84.83	83.65
4	118.25	84.83	86.59
5	118.40	86.59	86.74
6	119.55	87.52	87.87
7	128.15	88.15	91.00

Mean	118.13	84.05	83.78
STD Dev.	4.99	4.18	6.31
CV%	0.04	0.05	0.08
Variance	24.90	17.48	39.80

Z-SCORE	Conc (mg/L)_Level 1	Conc (mg/L)_Level 2	Conc (mg/L)_Level 3
EXPERIMENTS	ACETONITRILE	ACETONITRILE	ACETONITRILE
1	-0.86	-1.42	-1.68
2	-0.78	-1.37	-1.01
3	-0.74	0.19	-0.02
4	0.02	0.19	0.45
5	0.05	0.61	0.47
6	0.28	0.83	0.65
7	2.01	0.98	1.15

Experimental Data	Conc (mg/L)_Level 1	Conc (mg/L)_Level 2	Conc (mg/L)_Level 3
EXPERIMENTS	ETHYL ACETATE	ETHYL ACETATE	ETHYL ACETATE
1	2008.92	9516.16	11885.17
2	2023.41	9560.76	12598.66
3	2088.07	9626.53	12897.44
4	2100.33	10711.26	13554.07
5	2115.94	11212.93	14033.44
6	2127.09	11241.92	14060.20
7	2265.22	11350.06	14842.81

Spike (mg/L)	2229.65	11148.27	15161.65
---------------------	---------	----------	----------

Mean	2104.14	10459.95	13410.26
STD Dev.	84.13	859.05	1010.88
CV%	4.00	8.21	7.54
Variance	7078.14	737960.03	1021871.91

Recovery %	Recovery %_Level 1	Recovery %_Level 2	Recovery %_Level 3
EXPERIMENTS	ETHYL ACETATE	ETHYL ACETATE	ETHYL ACETATE
1	90.10	85.36	78.39
2	90.75	85.76	83.10
3	93.65	86.35	85.07
4	94.20	96.08	89.40
5	94.90	100.58	92.56
6	95.40	100.84	92.74
7	101.60	101.81	97.90

Mean	94.37	93.83	88.45
STD Dev.	3.77	7.71	6.67
CV%	0.04	0.08	0.08
Variance	14.24	59.38	44.45

Z-SCORE	Conc (mg/L)_Level 1	Conc (mg/L)_Level 2	Conc (mg/L)_Level 3
EXPERIMENTS	ETHYL ACETATE	ETHYL ACETATE	ETHYL ACETATE
1	-1.13	-1.10	-1.51
2	-0.96	-1.05	-0.80
3	-0.19	-0.97	-0.51
4	-0.05	0.29	0.14
5	0.14	0.88	0.62
6	0.27	0.91	0.64
7	1.91	1.04	1.42

Experimental Data	Conc (mg/L)_Level 1	Conc (mg/L)_Level 2	Conc (mg/L)_Level 3
EXPERIMENTS	TOLUENE	TOLUENE	TOLUENE
1	2404.84	10309.11	13396.77
2	2433.68	10831.60	14506.34
3	2472.90	10900.81	14587.08
4	2506.34	10982.70	15475.20
5	2522.49	12298.73	15814.30
6	2531.72	12805.07	15867.36
7	2801.85	12848.90	16867.36

Spike (mg/L)	2306.81	11534.03	15409.46
---------------------	---------	----------	----------

Mean	2524.83	11568.13	15216.35
STD Dev.	130.73	1050.38	1139.09
CV%	5.18	9.08	7.49
Variance	17089.04	1103306.85	1297525.02

Recovery %	Recovery %_Level 1	Recovery %_Level 2	Recovery %_Level 3
EXPERIMENTS	TOLUENE	TOLUENE	TOLUENE
1	104.25	89.38	86.94
2	105.50	93.91	94.14
3	107.20	94.51	94.66
4	108.65	95.22	100.43
5	109.35	106.63	102.63
6	109.75	111.02	102.97
7	121.46	111.40	109.46

Mean	109.45	100.30	98.75
STD Dev.	5.67	9.11	7.39
CV%	0.05	0.09	0.07
Variance	32.11	82.93	54.64

Z-SCORE	Conc (mg/L)_Level 1	Conc (mg/L)_Level 2	Conc (mg/L)_Level 3
EXPERIMENTS	TOLUENE	TOLUENE	TOLUENE
1	-0.92	-1.20	-1.60
2	-0.70	-0.70	-0.62
3	-0.40	-0.64	-0.55
4	-0.14	-0.56	0.23
5	-0.02	0.70	0.52
6	0.05	1.18	0.57
7	2.12	1.22	1.45

A4. Experimental Data of Recovery % of six target compounds

RECOVERY %	GROUPS					
EXPERIMENTS	METHANOL	ETHANOL	ISOPROPANOL	ACETONITRILE	ETHYL ACETATE	TOLUENE
1	116.37	108.65	99.48	111.52	114.71	109.69
2	99.88	122.82	100.48	97.24	101.12	97.71
3	111.35	130.43	111.73	108.92	113.18	109.08
4	112.37	93.82	113.03	109.26	113.21	107.92
5	109.88	117.37	109.72	107.60	108.01	105.71
6	109.35	114.38	112.55	107.21	111.14	106.54
7	119.42	114.38	112.55	118.11	111.14	118.09
8	111.37	112.56	113.57	110.08	114.03	112.42
9	117.54	101.75	114.73	117.71	120.39	120.58
10	105.82	96.69	107.45	104.96	106.57	105.84
11	96.94	96.43	100.19	97.00	98.53	98.77
12	104.16	102.62	108.68	104.55	105.81	105.13
13	105.45	107.40	109.67	105.81	106.76	104.73
14	116.75	113.14	119.44	116.47	116.12	110.75
15	120.97	125.23	123.31	120.17	119.69	113.84
16	118.48	121.57	121.34	118.36	115.70	109.51
17	102.26	109.30	107.61	104.11	104.59	102.41
18	107.50	117.25	112.12	109.42	104.68	94.40
19	117.85	119.75	117.49	113.48	115.79	119.28
20	113.67	117.39	122.72	116.08	115.16	109.72
21	112.18	120.21	113.05	101.01	111.86	109.31
22	106.84	109.32	108.51	96.25	106.85	103.56
23	113.37	120.24	115.72	103.01	113.77	112.54
24	111.34	116.17	114.31	101.54	111.91	110.44
25	115.69	115.37	119.16	105.68	115.70	113.53

RECOVERY %	GROUPS					
DIMENSION	METHANOL	ETHANOL	ISOPROPANOL	ACETONITRILE	ETHYL ACETATE	TOLUENE
25	25	25	25	25	25	25
MEAN	111.07	112.97	112.34	108.22	111.06	108.46
STD DEV.	6.30	9.29	6.45	7.00	5.56	6.35

RECOVERY %	METHANOL	ETHANOL	ISOPROPANOL	ACETONITRILE	ETHYL ACETATE	TOLUENE
Minimum	96.94	93.82	99.48	96.25	98.53	94.40
Maximum	120.97	130.43	123.31	120.17	120.39	120.58
Sum	2776.80	2824.23	2808.59	2705.53	2776.45	2711.49
Points	25.00	25.00	25.00	25.00	25.00	25.00
Mean	111.07	112.97	112.34	108.22	111.06	108.46
Median	111.37	114.38	112.55	107.60	111.91	109.31
RMS	111.24	113.33	112.52	108.44	111.19	108.64
StdDeviation	6.30	9.29	6.45	7.00	5.56	6.35
Variance	39.65	86.22	41.61	49.02	30.87	40.38
StdError	1.26	1.86	1.29	1.40	1.11	1.27
Skewness	-0.46	-0.46	-0.31	0.04	-0.45	-0.17
Kurtosis	-0.52	-0.39	-0.27	-0.92	-0.47	-0.09

A5. Experimental Data of Recovery % of 24 compounds

RECOVERY %	Group 1	Group 2	Group 3	Group 4	Group 5	Group 6	Group 7	Group 8
n.	METHANOL	ETHANOL	ISOPROPANOL	ACETONE	ACETONITRILE	METHYL ACETATE	DICHLOROMETANE	HEXANE
1	100.24	104.07	98.36	102.65	99.73	100.94	105.83	125.15
2	84.14	88.54	84.88	88.42	84.43	88.08	92.00	111.24
3	91.22	95.85	91.71	95.78	91.21	94.81	99.30	118.76
4	95.17	99.07	94.70	99.48	94.92	98.19	103.00	118.67
5	94.56	97.57	93.07	98.20	93.92	95.53	100.09	108.64
6	99.73	110.55	96.84	103.02	98.16	101.02	105.34	110.84
7	95.51	106.91	93.36	98.33	94.38	96.24	100.99	102.39
8	97.28	109.56	94.95	100.09	95.98	97.89	103.15	105.82
9	98.62	112.90	96.33	102.46	98.60	99.70	103.64	99.66
10	103.71	108.70	101.35	107.39	103.46	104.05	108.01	100.59
11	100.83	104.68	98.46	104.50	100.93	101.03	104.49	91.67
12	104.72	107.51	102.99	108.82	105.24	104.40	108.11	96.23
13	105.62	108.10	102.18	109.10	105.48	102.73	108.32	89.13
14	105.90	121.12	104.64	109.12	106.97	104.31	107.47	83.91
15	102.61	115.12	94.53	103.54	98.73	105.01	105.85	138.66
16	86.25	89.65	81.71	87.04	84.05	88.71	89.20	113.97
17	100.14	101.29	93.41	101.13	96.99	101.69	102.47	126.14
18	89.81	102.25	85.73	89.78	87.86	90.54	90.48	107.86
19	85.64	90.61	83.36	86.20	84.74	87.05	86.81	100.88
20	90.40	91.73	87.43	90.77	89.73	91.04	91.09	100.76
21	90.55	92.57	87.43	91.28	89.68	90.81	90.67	97.24
22	91.92	95.53	90.38	92.25	92.20	92.21	91.40	93.56
23	92.90	96.07	90.75	92.93	93.29	92.33	92.04	90.62
24	99.83	96.25	96.21	101.67	99.86	102.67	102.95	126.93
25	100.11	104.45	98.39	100.95	100.79	101.54	101.22	118.10
26	86.29	82.88	83.73	87.62	86.08	88.14	88.80	109.19
27	102.39	95.50	97.52	103.42	101.84	103.56	103.85	122.93
28	102.23	98.11	98.70	103.67	102.54	103.88	103.94	118.68
29	86.95	84.30	86.98	87.80	88.60	87.05	86.80	87.39
30	92.98	99.38	96.03	92.93	96.45	91.35	90.06	78.93
31	101.97	98.39	101.07	101.78	104.23	102.30	101.92	99.29
32	95.79	89.98	91.08	97.81	95.51	97.09	97.20	111.07
33	102.83	96.60	99.49	103.89	103.86	102.77	102.24	105.31
34	104.42	97.67	99.34	106.36	104.70	108.13	109.46	141.90
35	100.67	96.69	98.27	101.38	102.00	101.82	102.46	118.23
36	97.54	95.43	97.03	98.15	99.76	98.95	98.87	107.85
37	95.54	95.12	97.16	96.22	99.06	96.60	96.19	99.42
38	96.14	95.61	98.37	95.93	99.94	95.83	94.91	91.74
39	89.48	85.93	86.77	89.68	90.25	90.62	91.47	106.94
40	94.19	92.47	93.20	95.17	95.86	95.52	95.57	106.99
41	95.18	93.84	94.81	95.36	97.41	95.91	95.54	101.98
42	93.43	90.95	90.25	94.29	94.46	94.84	95.21	110.00
43	96.05	95.60	94.67	96.80	94.82	97.60	98.33	106.85
44	104.15	105.58	105.08	104.03	101.22	104.77	105.02	111.45
45	84.35	85.74	89.52	84.69	84.85	84.81	83.28	82.12
46	95.27	100.05	100.49	95.24	95.10	95.72	94.77	89.52
47	93.57	96.19	91.00	96.20	90.27	99.71	96.10	143.28
48	83.81	88.15	84.05	85.42	81.91	88.54	85.21	126.00
49	100.16	98.64	94.83	99.85	93.52	100.33	100.60	116.53
50	117.11	92.91	115.43	119.95	118.08	119.98	121.85	138.23
51	111.14	116.04	111.88	113.92	112.81	113.71	115.44	129.92
52	113.47	94.99	116.70	115.65	116.42	115.66	116.65	127.89
53	89.11	108.00	90.15	92.06	93.35	105.74	95.46	94.25
54	99.05	109.39	99.44	102.00	103.54	116.87	105.84	102.83
55	97.58	113.01	99.55	101.18	102.56	114.57	103.40	96.54
56	104.27	108.72	102.13	107.20	109.85	129.48	113.06	112.68
57	100.66	107.01	102.14	102.23	107.29	123.58	108.12	97.30
58	93.24	93.40	87.18	94.74	97.40	107.92	98.82	98.53
59	118.74	100.50	119.24	121.16	119.20	120.79	122.37	137.58
60	108.11	114.47	108.69	107.70	107.45	109.57	109.35	110.22
61	109.79	102.50	109.79	109.59	109.00	110.71	110.37	109.99
62	113.52	119.00	113.39	112.52	113.20	113.19	113.44	109.13
63	116.37	131.35	113.24	112.40	111.52	113.67	124.76	108.72
64	96.14	119.89	99.79	95.29	97.29	97.18	95.56	90.15
65	82.88	117.10	95.27	97.96	95.80	100.71	101.44	102.94
66	85.65	115.22	98.38	98.79	97.71	100.75	101.52	98.90
67	78.85	87.73	92.84	89.77	90.18	93.45	94.17	89.61
68	109.80	118.12	125.57	130.41	121.52	125.30	102.36	112.10
69	119.61	109.93	118.41	121.22	112.25	115.96	96.49	99.53
70	131.71	119.46	133.40	133.47	125.05	127.74	106.29	108.43

RECOVERY %	Group 9	Group 10	Group 11	Group 12	Group 13	Group 14	Group 15	Group 16
n.	2-BUTANONE	ETHYL ACETATE	TETRAHYDROFURAN	CYCLOHEXANE	ISOPROPYL ACETATE	HEPTANE	1,4-DIOXAN	CYCLOPENTYL METHYL ETHER
1	98.90	97.72	101.39	127.53	97.73	122.51	98.39	102.95
2	86.98	86.30	89.57	112.59	89.04	107.78	87.08	96.43
3	93.22	92.93	96.32	120.22	92.78	116.95	93.67	100.53
4	96.41	95.62	98.19	119.32	98.22	114.32	95.67	101.40
5	94.05	93.34	94.90	108.49	94.65	104.25	93.06	96.16
6	97.86	96.80	98.23	110.06	97.16	106.24	96.49	98.64
7	93.92	92.49	93.40	101.88	93.37	98.71	92.08	93.52
8	95.82	94.56	96.13	108.58	95.23	106.50	94.40	97.49
9	96.71	95.01	95.04	98.64	96.84	94.57	61.93	93.90
10	101.29	99.34	98.58	99.56	102.08	95.19	98.73	97.06
11	97.82	96.17	94.07	91.30	96.33	90.22	94.04	92.05
12	101.92	99.89	97.87	94.96	100.61	92.46	98.83	95.83
13	101.14	98.95	95.93	87.39	100.66	83.64	96.94	91.57
14	102.70	99.79	95.64	83.44	101.19	81.60	99.40	92.53
15	99.66	98.89	109.72	137.87	97.19	129.48	98.02	106.67
16	85.95	84.85	93.70	114.10	82.66	109.11	85.04	91.42
17	97.88	96.81	105.29	124.86	95.55	118.33	96.25	101.89
18	88.94	87.24	94.56	107.91	86.36	103.82	95.85	92.27
19	86.36	84.37	91.35	101.30	84.72	98.79	85.86	88.94
20	89.99	88.07	94.02	100.48	88.70	96.44	88.99	90.41
21	88.95	87.56	89.86	96.58	88.64	93.00	88.71	89.18
22	91.13	89.73	90.88	93.49	90.98	90.51	91.26	90.44
23	91.65	89.89	90.54	90.38	90.56	88.72	91.52	89.91
24	98.37	99.69	103.33	127.43	100.47	122.81	101.32	108.48
25	98.85	99.66	102.28	118.33	101.53	116.79	102.94	108.78
26	85.78	85.95	89.54	110.01	87.41	104.91	88.38	93.80
27	99.77	99.76	103.01	123.59	99.38	120.20	101.81	107.22
28	100.42	100.52	102.73	119.11	102.61	115.26	102.54	107.17
29	87.14	86.24	85.86	89.45	86.90	91.57	88.67	90.61
30	94.58	93.07	89.69	83.12	93.17	89.76	96.57	95.88
31	101.99	100.78	99.47	101.20	103.80	100.31	104.30	104.92
32	92.87	92.76	94.63	110.26	94.35	105.72	93.81	97.97
33	100.53	99.21	99.68	105.31	100.78	102.74	101.98	102.04
34	103.70	105.68	110.12	146.13	104.07	137.32	107.03	116.03
35	100.61	101.38	102.60	123.15	66.28	182.67	103.53	109.43
36	98.30	99.04	98.90	114.21	101.34	110.71	102.41	106.92
37	98.47	98.46	96.78	106.26	100.26	106.84	103.70	106.66
38	98.93	98.39	95.47	99.21	100.45	101.55	103.09	105.76
39	88.72	89.33	90.79	111.30	91.50	104.79	91.98	97.55
40	94.99	95.35	95.60	111.98	97.56	107.29	98.38	102.57
41	96.50	95.92	97.29	107.22	96.92	105.40	99.41	102.57
42	93.26	92.81	96.49	113.97	94.61	107.07	95.61	99.75
43	95.19	95.14	95.41	101.64	93.30	100.45	94.48	94.70
44	102.48	103.15	103.62	107.29	101.16	105.11	101.41	100.71
45	87.06	86.59	84.57	80.85	86.40	82.65	87.43	84.83
46	97.39	96.77	94.29	87.38	95.91	89.52	96.20	93.35
47	92.97	126.22	97.03	243.45	93.23	142.15	94.36	97.88
48	85.47	115.53	88.59	217.35	86.86	129.39	87.64	91.29
49	94.68	94.12	98.15	110.38	92.97	105.64	92.01	94.35
50	115.58	116.33	117.90	122.81	116.20	83.65	113.55	115.01
51	111.45	111.95	113.36	116.73	113.47	79.97	110.37	111.73
52	115.79	115.81	115.70	115.68	117.00	81.60	115.29	115.86
53	95.42	99.15	98.82	88.29	105.36	66.97	89.87	102.54
54	104.92	108.72	108.33	96.32	112.91	73.27	97.90	111.50
55	104.03	107.81	106.34	90.89	113.43	70.24	97.87	110.51
56	107.19	110.77	114.11	104.03	119.09	76.92	102.18	108.22
57	105.46	108.82	109.06	91.56	116.32	70.86	100.60	104.67
58	94.41	95.73	98.84	90.48	102.52	65.08	86.16	97.26
59	117.82	118.69	119.72	123.09	120.17	84.44	115.40	117.12
60	108.56	108.92	108.58	109.14	105.89	109.55	107.14	106.67
61	109.34	109.80	109.16	108.59	110.02	107.33	108.02	107.62
62	112.64	112.52	111.27	107.97	111.21	110.00	111.31	110.26
63	114.91	114.71	112.60	107.13	110.38	103.34	110.69	106.24
64	98.23	97.76	95.62	90.30	97.34	92.57	97.33	96.69
65	100.78	101.13	101.14	102.32	101.78	101.32	99.88	100.84
66	102.12	102.18	101.07	98.95	105.82	98.44	101.30	101.13
67	95.90	95.97	94.15	90.06	97.09	92.43	94.90	94.88
68	122.06	126.22	121.30	115.01	125.36	114.90	122.12	119.11
69	111.08	114.57	109.73	102.49	114.24	102.41	110.87	107.57
70	125.53	129.33	122.55	113.15	127.32	115.85	125.27	121.61

RECOVERY %	Group 17	Group 18	Group 19	Group 20	Group 21	Group 22	Group 23	Group 24
EXPERIMENTS	TOLUENE	BUTYL ACETATE	DIMETHYLFORMAMIDE	2-BUTANOL	1-BUTANOL	PYRIDINE	DIISOPROPYL ETHER	OCTANE
1	102.41	97.35	100.28	85.25	84.48	104.35	102.88	96.85
2	92.84	87.65	87.16	94.69	94.62	113.05	86.49	81.04
3	99.19	93.81	93.33	88.28	87.50	102.94	106.72	99.45
4	100.74	95.88	95.96	92.36	90.77	105.56	102.69	95.95
5	95.87	93.02	93.84	93.84	91.98	101.51	90.40	83.38
6	97.81	95.39	95.91	97.80	95.62	102.94	96.03	87.31
7	93.36	91.85	92.69	83.01	81.39	88.48	92.86	85.68
8	96.69	94.14	99.79	109.17	85.36	88.45	108.37	98.17
9	93.57	93.23	94.84	87.89	85.73	87.32	104.95	96.84
10	97.28	97.94	100.18	117.61	114.19	110.97	89.87	81.93
11	92.49	93.22	94.72	108.21	105.21	100.97	89.29	80.54
12	96.19	98.11	101.34	90.95	89.29	82.53	93.05	83.87
13	92.58	95.43	98.72	106.33	104.12	86.66	108.05	95.94
14	93.66	98.98	103.79	102.80	101.51	80.16	95.38	88.89
15	105.13	96.17	93.91	100.14	96.14	124.52	94.09	85.39
16	90.49	83.96	83.16	81.14	79.60	100.49	83.22	74.67
17	101.55	94.54	93.55	83.47	80.59	101.28	105.00	91.61
18	91.88	87.51	88.23	91.09	88.76	108.67	94.04	82.97
19	89.07	85.71	86.49	81.41	80.00	94.26	89.87	84.70
20	90.96	88.74	90.07	92.89	91.37	103.37	89.83	82.45
21	89.77	88.72	90.79	112.04	113.10	101.37	104.19	103.22
22	91.13	91.39	94.07	95.10	95.08	80.67	95.02	93.83
23	90.82	92.31	95.09	98.61	98.81	78.94	118.94	117.79
24	108.27	102.14	101.85	109.68	110.41	108.15	101.28	100.75
25	109.04	105.56	106.37	97.91	98.16	94.88	104.37	108.08
26	94.02	89.42	89.92	100.42	100.50	95.18	96.99	100.97
27	107.58	102.51	103.71	107.25	107.44	99.24	98.60	105.00
28	107.62	103.75	105.32	119.03	119.95	103.59	92.29	105.72
29	90.86	90.26	92.77	118.41	118.36	97.52	88.55	105.47
30	96.86	99.74	103.63	119.05	118.54	109.24	108.21	109.32
31	105.82	107.33	110.40	109.50	109.93	96.71	122.21	136.53
32	98.20	91.95	96.38	107.21	107.29	89.85	98.18	114.68
33	103.88	102.93	104.14	109.37	110.89	78.94	82.00	111.79
34	115.01	107.10	107.39	84.48	82.30	92.41	97.79	100.24
35	109.69	104.39	106.93	102.44	99.82	99.74	93.81	109.63
36	106.97	104.97	106.81	114.27	113.01	107.56	81.83	102.23
37	106.52	106.32	109.08	95.99	94.92	96.04	96.81	101.56
38	105.84	107.20	110.58	114.27	113.01	107.56	81.83	102.23
39	97.20	94.26	96.40	102.70	103.40	83.87	104.97	111.76
40	102.58	100.75	102.47	112.09	112.67	101.61	107.17	115.53
41	102.58	101.94	104.80	89.32	86.43	91.81	87.48	88.97
42	99.55	96.77	98.28	109.42	106.01	104.60	96.10	103.33
43	94.83	94.31	93.01	91.65	89.20	96.89	100.02	108.07
44	100.08	98.92	96.64	97.39	97.51	98.12	99.12	108.33
45	85.14	86.96	87.02	98.83	99.97	98.90	65.22	74.08
46	93.07	95.78	96.47	88.87	89.04	86.78	77.72	99.12
47	96.41	90.10	87.82	115.78	114.20	105.45	91.71	99.26
48	89.92	85.86	84.24	81.39	81.17	88.96	97.64	97.84
49	93.26	89.63	87.05	92.24	92.49	103.05	70.07	70.67
50	114.21	110.85	110.17	102.47	102.16	111.30	104.71	97.94
51	111.26	108.61	109.56	89.83	89.91	96.89	102.79	99.08
52	115.68	114.91	115.31	88.07	88.18	94.53	91.53	87.51
53	104.15	102.67	96.77	85.35	85.04	90.14	82.89	82.36
54	113.45	111.09	104.79	94.69	94.03	98.61	96.94	92.47
55	112.49	112.59	107.35	97.78	97.22	101.58	86.39	86.06
56	115.48	111.61	106.56	105.56	104.92	108.77	98.55	97.69
57	113.09	115.28	112.16	109.48	108.75	111.03	93.84	94.72
58	98.48	96.36	92.17	109.44	108.63	110.20	101.16	100.55
59	116.94	113.81	113.65	112.89	112.69	112.26	98.03	100.23
60	107.20	107.79	107.51	113.60	113.90	110.36	92.81	99.19
61	107.79	108.54	108.72	114.85	114.36	110.43	113.23	119.47
62	110.50	112.46	112.05	110.95	111.15	103.36	103.49	113.04
63	109.69	110.91	114.39	122.96	124.11	110.74	113.33	128.10
64	96.83	100.17	99.73	121.66	122.56	106.11	93.09	110.95
65	101.01	100.40	86.07	99.20	96.51	98.17	89.53	85.61
66	101.87	103.10	89.13	93.54	94.59	92.63	91.54	98.42
67	95.52	97.75	83.10	96.71	97.65	95.38	110.80	112.14
68	117.01	123.81	123.14	93.77	93.13	99.00	99.88	100.33
69	105.65	112.27	111.84	87.77	87.98	91.57	97.42	99.51
70	119.49	128.77	129.36	71.80	71.74	75.11	96.80	98.60

	Group 1	Group 2	Group 3	Group 4	Group 5	Group 6	Group 7	Group 8
	METHANOL	ETHANOL	ISOPROPANOL	ACETONE	ACETONITRILE	METHYL ACETATE	DICHLOROMET ANE	HEXANE
DIMENSION	70.00	70.00	70.00	70.00	70.00	70.00	70.00	70.00
MEAN	98.57	101.55	98.07	100.83	99.39	101.93	100.98	107.81
STD DEV.	9.85	10.43	10.14	10.00	9.30	10.34	8.76	14.57
CONFIDENCE 99%	3.03	3.21	3.12	3.08	2.86	3.18	2.70	4.48
L 1	95.53	99.20	95.68	98.59	97.31	99.54	98.98	104.17
L 2	101.60	103.89	100.46	103.06	101.46	104.31	102.97	111.45
LCL	95.03	97.80	94.44	97.24	96.05	98.22	97.84	102.58
UCL	102.10	105.29	101.71	104.41	102.72	105.64	104.12	113.03

	Group 9	Group 10	Group 11	Group 12	Group 13	Group 14	Group 15	Group 16
	2-BUTANONE	ETHYL ACETATE	TETRAHYDROFU RAN	CYCLOHEX ANE	ISOPROPHYL ACETATE	HEPTANE	1,4-DIOXAN	CYCLOPENTYL METHYL ETHER
DIMENSION	70.00	70.00	70.00	70.00	70.00	70.00	70.00	70.00
MEAN	99.16	100.07	100.15	109.79	99.68	101.90	98.30	101.14
STD DEV.	8.77	10.22	8.65	24.73	10.68	18.66	9.50	8.27
CONFIDENCE 99%	2.70	3.15	2.66	7.61	3.29	5.75	2.93	2.55
L 1	96.99	97.72	98.19	104.04	97.09	97.64	96.03	99.23
L 2	101.33	102.42	102.11	115.55	102.26	106.16	100.58	103.04
LCL	96.02	96.40	97.05	100.93	95.85	95.21	94.90	98.17
UCL	102.31	103.73	103.25	118.66	103.51	108.59	101.71	104.10

	Group 17	Group 18	Group 19	Group 20	Group 21	Group 22	Group 23	Group 24
	TOLUENE	BUTHYL ACETATE	DIMETHYLFORM AMIDE	2-BUTANOL	1-BUTANOL	PYRIDINE	DIISOPROPYL ETHER	OCTANE
DIMENSION	70.00	70.00	70.00	70.00	70.00	70.00	70.00	70.00
MEAN	101.34	99.99	99.87	100.08	99.02	98.78	96.31	98.11
STD DEV.	8.45	9.36	9.70	11.76	12.16	9.81	10.10	12.52
CONFIDENCE 99%	2.60	2.88	2.99	3.62	3.74	3.02	3.11	3.85
L 1	99.37	97.79	97.67	97.41	95.32	95.79	93.31	94.83
L 2	103.32	102.20	102.07	102.74	102.71	101.76	99.32	101.39
LCL	98.31	96.64	96.39	95.86	94.65	95.26	92.69	93.62
UCL	104.37	103.35	103.35	104.30	103.38	102.29	99.94	102.60

	GENERAL
DIMENSION	1680
MEAN	100.53
STD DEV.	3.67
GROUPS	24

A6. Reproducibility Tests

GC-FID_Control	Group 1	Group 2	Group 3	Group 4	Group 5	Group 6
EXPERIMENTS	METHANOL	ETHANOL	ISOPROPANOL	ACETONE	ACETONITRILE	METHYL ACETATE
1	100.95	99.73	99.51	104.22	100.71	101.72
2	88.71	90.58	89.19	91.70	90.61	88.25
3	105.04	108.51	109.15	108.76	109.09	104.19
4	108.29	113.13	114.19	111.52	112.89	105.94
5	114.52	119.49	119.46	118.03	118.45	112.70
6	85.44	82.66	83.10	88.30	83.04	87.02
7	88.73	84.71	85.20	91.84	86.57	90.74

GC-FID_Control	METHANOL	ETHANOL	ISOPROPANOL	ACETONE	ACETONITRILE	METHYL ACETATE
DIMENSION	7	7	7	7	7	7
MEAN	98.81	99.83	99.97	102.05	100.19	98.65
STANDARD DEVIATION	11.27	14.42	14.64	11.51	13.82	9.97
MEAN_all	98.57	101.55	98.07	100.83	99.39	101.93
STD DEV_all	9.85	10.43	10.14	10.00	9.30	10.34
TEST Z (Z=2.2326)	0.01	-0.06	0.07	0.05	0.03	-0.12

GC-FID_Control	Group 7	Group 8	Group 9	Group 10	Group 11	Group 12
EXPERIMENTS	DICHLOROMETANE	HEXANE	2-BUTANONE	ETHYL ACETATE	TETRAHYDROFURAN	CYCLOHEXANE
1	107.21	119.62	101.18	101.64	103.73	121.29
2	92.17	85.33	88.87	88.28	86.59	85.80
3	108.40	92.17	107.61	106.33	101.67	93.94
4	110.07	84.20	110.91	108.92	101.78	86.03
5	117.15	93.33	116.61	114.74	108.60	94.06
6	90.75	105.08	72.15	88.25	53.49	106.83
7	94.76	111.89	73.90	90.63	55.34	113.59

GC-FID_Control	DICHLOROMETANE	HEXANE	2-BUTANONE	ETHYL ACETATE	TETRAHYDROFURAN	CYCLOHEXANE
DIMENSION	7	7	7	7	7	7
MEAN	102.93	98.80	95.89	99.83	87.32	100.22
STANDARD DEVIATION	10.26	13.62	17.87	10.82	23.47	13.86
MEAN_all	100.98	107.81	99.16	100.07	100.15	109.79
STD DEV_all	8.76	14.57	8.77	10.22	8.65	24.73
TEST Z (Z=2.2326)	0.08	-0.23	-0.14	-0.01	-0.56	-0.15

GC-FID_Control	Group 13	Group 14	Group 15	Group 16	Group 17	Group 18
EXPERIMENTS	ISOPROPYL ACETATE	HEPTANE	1,4-DIOXAN	CYCLOPENTYL METHYL ETHER	TOLUENE	BUTYL ACETATE
1	103.34	118.98	101.52	108.29	106.94	101.36
2	89.78	85.31	87.90	87.62	87.41	88.00
3	109.00	96.49	106.54	104.77	104.72	107.75
4	112.08	89.94	109.93	105.21	105.77	112.59
5	117.80	95.74	115.58	109.90	110.83	116.97
6	88.70	107.93	88.32	95.37	94.28	88.30
7	91.02	114.06	90.82	99.06	97.83	89.89

GC-FID_Control	ISOPROPYL ACETATE	HEPTANE	1,4-DIOXAN	CYCLOPENTYL METHYL ETHER	TOLUENE	BUTYL ACETATE
DIMENSION	7	7	7	7	7	7
MEAN	101.67	101.21	100.09	101.46	101.11	100.69
STANDARD DEVIATION	11.89	12.64	11.21	7.93	8.25	12.17
MEAN_all	99.68	101.90	98.30	101.14	101.34	99.99
STD DEV_all	10.68	18.66	9.50	8.27	8.45	9.36
TEST Z (Z=2.2326)	0.07	-0.01	0.07	0.01	-0.01	0.03

GC-FID_Control	Group 19	Group 20	Group 21	Group 22	Group 23	Group 24
EXPERIMENTS	DIMETHYLFORMAMIDE	2-BUTANOL	1-BUTANOL	PYRIDINE	DIISOPROPYL ETHER	OCTANE
1	100.37	77.90	77.51	88.84	90.99	91.34
2	88.64	93.59	93.69	91.63	102.44	102.70
3	108.76	102.82	103.90	95.67	104.23	105.24
4	114.77	112.83	113.62	101.90	116.66	117.41
5	119.36	135.95	137.41	110.50	120.12	120.97
6	87.45	72.87	72.90	86.58	88.67	90.21
7	88.80	82.98	83.74	88.35	88.67	90.21

GC-FID_Control	DIMETHYLFORMAMIDE	2-BUTANOL	1-BUTANOL	PYRIDINE	DIISOPROPYL ETHER	OCTANE
DIMENSION	7	7	7	7	7	7
MEAN	101.16	96.99	97.54	94.78	101.68	102.58
STANDARD DEVIATION	13.37	22.20	22.74	8.68	13.07	12.89
MEAN_all	99.87	100.08	99.02	98.78	96.31	98.11
STD DEV_all	9.70	11.76	12.16	9.81	10.10	12.52
TEST Z (Z=2.2326)	0.05	-0.10	-0.05	-0.15	0.20	0.14

GC-FID_Principal	Group 1	Group 2	Group 3	Group 4	Group 5	Group 6
EXPERIMENTS	METHANOL	ETHANOL	ISOPROPANOL	ACETONE	ACETONITRILE	METHYL LACETATE
1	91.22	95.85	91.71	95.78	91.21	94.81
2	98.62	112.90	96.33	102.46	98.60	99.70
3	100.83	104.68	98.46	104.50	100.93	101.03
4	104.72	107.51	102.99	108.82	105.24	104.40
5	105.62	108.10	102.18	109.10	105.48	102.73
6	86.25	89.65	81.71	87.04	84.05	88.71
7	102.61	115.12	94.53	103.54	98.73	105.01

GC-FID_Principal	METHANOL	ETHANOL	ISOPROPANOL	ACETONE	ACETONITRILE	METHYL LACETATE
DIMENSION	7	7	7	7	7	7
MEAN	98.55	104.83	95.42	101.61	97.75	99.49
STANDRD DEVIATION	7.24	9.13	7.25	7.82	7.72	5.86
MEAN_all	98.57	101.55	98.07	100.83	99.39	101.93
STD DEV_all	9.85	10.43	10.14	10.00	9.30	10.34
TEST Z (Z=2.2326)	0.00	0.12	-0.10	0.03	-0.07	-0.09

GC-FID_Principal	Group 7	Group 8	Group 9	Group 10	Group 11	Group 12
EXPERIMENTS	DICHLOROMETANE	HEXANE	2-BUTANONE	ETHYL LACETATE	TETRAHYDROFURAN	CYCLOHEXANE
1	99.30	118.76	93.22	92.93	96.32	120.22
2	103.64	99.66	96.71	95.01	95.04	98.64
3	104.49	91.67	97.82	96.17	94.07	91.30
4	108.11	96.23	101.92	99.89	97.87	94.96
5	108.32	89.13	101.14	98.95	95.93	87.39
6	89.20	113.97	85.95	84.85	93.70	114.10
7	105.85	138.66	99.66	98.89	109.72	137.87

GC-FID_Principal	DICHLOROMETANE	HEXANE	2-BUTANONE	ETHYL LACETATE	TETRAHYDROFURAN	CYCLOHEXANE
DIMENSION	7	7	7	7	7	7
MEAN	102.70	106.87	96.63	95.24	97.52	106.35
STANDRD DEVIATION	6.69	17.86	5.55	5.21	5.56	18.35
MEAN_all	100.98	107.81	99.16	100.07	100.15	109.79
STD DEV_all	8.76	14.57	8.77	10.22	8.65	24.73
TEST Z (Z=2.2326)	0.07	-0.02	-0.11	-0.18	-0.11	-0.05

GC-FID_Principal	Group 13	Group 14	Group 15	Group 16	Group 17	Group 18
EXPERIMENTS	ISOPROPYL ACETATE	HEPTANE	1,4-DIOXAN	CYCLOPENTYL METHYL ETHER	TOLUENE	BUTHYL LACETATE
1	92.78	116.95	93.67	100.53	99.19	93.81
2	96.84	94.57	61.93	93.90	93.57	93.23
3	96.33	90.22	94.04	92.05	92.49	93.22
4	100.61	92.46	98.83	95.83	96.19	98.11
5	100.66	83.64	96.94	91.57	92.58	95.43
6	82.66	109.11	85.04	91.42	90.49	83.96
7	97.19	129.48	98.02	106.67	105.13	96.17

GC-FID_Principal	ISOPROPHYL ACETATE	HEPTANE	1,4-DIOXAN	CYCLOPENTYL METHYL ETHER	TOLUENE	BUTHY LACETATE
DIMENSION	7	7	7	7	7	7
MEAN	95.30	102.35	89.78	96.00	95.66	93.42
STANDRD DEVIATION	6.19	16.58	13.12	5.70	5.05	4.53
MEAN_all	99.68	101.90	98.30	101.14	101.34	99.99
STD DEV_all	10.68	18.66	9.50	8.27	8.45	9.36
TEST Z (Z=2.2326)	-0.16	0.01	-0.34	-0.23	-0.25	-0.27

GC-FID_Principal	Group 19	Group 20	Group 21	Group 22	Group 23	Group 24
EXPERIMENTS	DIMETHYLFORMAMIDE	2-BUTANOL	1-BUTANOL	PYRIDINE	DIISOPROPYL ETHER	OCTANE
1	93.33	88.28	87.50	102.94	106.72	99.45
2	94.84	87.89	85.73	87.32	104.95	96.84
3	94.72	108.21	105.21	100.97	89.29	80.54
4	101.34	90.95	89.29	82.53	93.05	83.87
5	98.72	106.33	104.12	86.66	108.05	95.94
6	83.16	81.14	79.60	100.49	83.22	74.67
7	93.91	92.89	91.37	103.37	89.83	82.45

GC-FID_Principal	DIMETHYLFORMAMIDE	2-BUTANOL	1-BUTANOL	PYRIDINE	DIISOPROPYL ETHER	OCTANE
DIMENSION	7	7	7	7	7	7
MEAN	94.29	93.67	91.83	94.90	96.45	87.68
STANDRD DEVIATION	5.70	9.99	9.51	8.97	9.95	9.60
MEAN_all	99.87	100.08	99.02	98.78	96.31	98.11
STD DEV_all	9.70	11.76	12.16	9.81	10.10	12.52
TEST Z (Z=2.2326)	-0.22	-0.21	-0.22	-0.15	0.00	-0.31

A7. Experiments on Laboratory Control Sample, Matrix Spike (MS) and Matrix Spike Duplicate (MSD)

- Laboratory Control Sample

EXPERIMENTS	METHANOL	ETHANOL	ISOPROPANOL	ACETONITRILE	ETHYL ACETATE	TOLUENE
1	116.37	108.65	110.50	108.01	113.17	109.84
2	99.88	122.82	96.86	94.12	98.79	96.77
3	111.35	130.43	103.24	100.92	105.21	101.14
4	112.37	93.82	100.74	96.41	102.75	100.89
5	109.88	117.37	94.27	89.54	97.58	96.03
6	119.42	114.38	108.38	99.49	110.54	104.15
7	111.37	112.56	88.41	83.26	93.35	90.07
8	105.82	96.69	92.59	88.59	93.50	94.19
9	96.94	96.43	92.33	87.28	91.05	93.58
10	104.16	102.62	93.76	87.81	93.27	96.59
11	116.75	113.14	108.34	106.62	107.97	101.28
12	120.97	125.23	121.23	119.06	111.86	108.84
13	118.48	121.57	107.23	109.26	106.54	103.39
14	120.09	117.25	121.85	109.42	104.68	103.57
15	102.16	119.75	109.86	113.48	115.79	99.46
16	102.46	117.39	111.63	116.08	115.16	105.59
17	107.20	120.21	110.81	101.01	111.86	139.48
18	104.65	109.32	116.50	96.25	106.85	117.77
19	90.56	120.24	113.32	103.01	113.77	111.94
20	75.90	116.17	110.45	101.54	111.91	109.68

	GROUPS					
	METHANOL	ETHANOL	ISOPROPANOL	ACETONITRILE	ETHYL ACETATE	TOLUENE
DIMENSION	20.00	20.00	20.00	20.00	20.00	20.00
AVERAGE VALUE	107.34	113.80	105.61	100.56	105.28	104.21
STANDRD DEVIATION	11.14	9.94	9.86	10.22	8.10	10.81
RSD%	10.37	8.73	9.34	10.16	7.70	10.37

- Matrix Spike (MS)

EXPERIMENTS	METHANOL	ETHANOL	ISOPROPANOL	ACETONITRILE	ETHYL ACETATE	TOLUENE
1	121.40	134.87	111.68	119.51	84.81	118.70
2	118.57	139.47	110.62	119.78	85.42	111.26
3	124.29	141.76	111.82	125.30	89.55	117.13
4	129.61	148.47	109.73	129.47	91.91	119.21
5	121.69	124.68	112.32	121.47	87.20	113.48
6	123.56	129.75	120.97	123.78	88.41	114.57
7	136.21	133.71	135.46	137.32	98.28	126.04
8	204.58	138.65	212.94	204.46	150.47	210.42
9	97.61	154.22	100.83	97.51	70.69	99.23
10	106.35	104.94	108.12	106.49	76.80	105.79
11	105.40	112.75	108.52	105.49	74.26	103.73
12	113.95	114.67	116.54	114.55	80.99	109.55
13	118.51	126.63	122.24	119.47	83.88	112.38
14	70.16	73.62	68.70	69.00	47.30	61.98
15	104.83	111.22	110.62	107.55	74.85	95.11
16	103.56	95.58	116.07	106.45	80.31	105.66
17	107.52	94.67	110.12	97.44	78.07	107.38
18	115.96	104.68	120.64	107.05	84.87	117.97
19	106.42	107.78	111.37	97.31	78.19	108.30
20	115.39	125.36	122.16	106.13	85.51	117.92

ANOVA	GROUPS					
	METHANOL	ETHANOL	ISOPROPANOL	ACETONITRILE	ETHYL ACETATE	TOLUENE
DIMENSION	20.00	20.00	20.00	20.00	20.00	20.00
AVERAGE VALUE	117.28	120.87	117.07	115.78	84.59	113.79
STANDRD DEVIATION	24.82	20.42	25.82	25.61	18.62	26.32
RSD %	21.16	16.89	22.05	22.12	22.02	23.13

- Matrix Spike Duplicate (MSD)

ANOVA	GROUPS - MATRIX SPIKE DUPLICATE					
EXPERIMENTS	METHANOL	ETHANOL	ISOPROPANOL	ACETONITRILE	ETHYL ACETATE	TOLUENE
1	129.59	147.66	116.83	125.46	89.07	126.21
2	123.28	140.53	114.93	122.68	87.31	114.50
3	112.36	127.24	99.13	112.19	79.87	104.03
4	128.21	132.80	108.84	130.29	91.55	120.29
5	121.26	128.78	112.57	121.69	85.99	112.96
6	126.77	133.63	123.25	126.18	89.66	116.08
7	130.85	121.71	129.50	131.68	94.17	120.13
8	196.91	123.96	205.60	197.78	145.89	204.28
9	100.36	159.08	103.87	91.13	72.74	102.70
10	105.84	111.15	108.37	96.35	76.98	107.87
11	108.08	108.48	110.42	98.19	78.06	104.27
12	75.92	127.41	116.79	114.22	81.57	99.47
13	121.15	136.83	122.35	121.27	85.07	112.49
14	75.92	75.01	73.40	74.45	51.18	66.67
15	115.39	113.88	122.16	117.42	82.99	105.72
16	120.36	117.43	133.94	123.97	92.62	121.29
17	101.66	110.19	106.91	92.81	76.03	104.09
18	116.47	106.92	118.97	105.67	84.33	116.59
19	114.74	110.28	122.16	104.58	83.30	114.91
20	113.15	119.19	118.95	103.81	83.64	114.45

ANOVA	GROUPS					
	METHANOL	ETHANOL	ISOPROPANOL	ACETONITRILE	ETHYL ACETATE	TOLUENE
DIMENSION	20.00	20.00	20.00	20.00	20.00	20.00
AVERAGE VALUE	116.91	122.61	118.45	115.59	85.60	114.45
STANDRD DEVIATION	24.26	17.85	24.15	24.59	16.96	24.52
RSD %	20.75	14.56	20.39	21.27	19.81	21.42

BIBLIOGRAPHY

1. Decreto Legislativo 3 aprile 2006, n. 152
2. Directive 2008/1/EC on Integrated Pollution Prevention and Control (IPPC Directive)
3. Regulation (EC) No 1272/2008 of the parliament and of the council of 16 December 2008 on classification, labelling and packaging of substances and mixtures, amending and repealing Directives 67/548/EEC and 1999/45/EC, and amending Regulation (EC) No 1907/2006, Official Journal of the European Union L 353/1
4. Pooya Azadi, Robert Malina, Steven R.H. Barrett, Markus Kraft, The evolution of the biofuel science, *Renewable and Sustainable Energy Reviews* 76 (2017) 1479–1484
5. Augustine Quek , Rajasekhar Balasubramanian, Life Cycle Assessment of Energy and Energy Carriers from Waste Matter e A Review, *Journal of Cleaner Production* 79 (2014) 18 - 31
6. Adeel Tariq, Yuosre F. Badi, Waqas Tariq, Umair Saeed Bhutta, Drivers and consequences of green product and process innovation: A systematic review, conceptual framework, and future outlook, *Technology in Society* 51 (2017) 8-23
7. M.A.Curran, Life Cycle Assessment: Principles and Practices, United States Environmental Protection Agency, National Risk Management Research Laboratory, Cincinnati, Ohio, 2006
8. Michael J. Raymond, C. Stewart Slater and Mariano J. Savelski, LCA approach to the analysis of solvent waste issues in the pharmaceutical Industry, *GreenChem.*, 2010,12, 1826–1834
9. Analytical procedures for quality control of pharmaceuticals in terms of residual solvents content: Challenges and recent developments, Maciej Tankiewicz, Jacek Namies'nik, Wiesław Sawicki, *Trends in Analytical chemistry* 80 (2016) 328–344
10. ICH guideline Q3C (R6) on impurities: guideline for residual solvents EMA/CHMP/ICH/82260/2006
11. Eduardo J., Cavanagh, Mariano J.Savelski, C.Stewart Slater, Optimization of environmental impact reduction and economic feasibility of solventwasterecovery using a new software tool, *Chemical Engineering Research and Design*, 9 2 (2 0 1 4) 1942–1954
12. Abejon R., Gareia A., Irabien A., 2015, Organic solvent recovery and reuse in pharmaceutical purification processes by nanofiltration membrane cascades, *Chemical Engineering Transactions*, 43, 1057-1062 DOI: 10.3303/CET1543177
13. Verónica García, Eva Pongrácz, Paul S. Phillips, Riitta L. Keiski, From waste treatment to resource efficiency in the chemical industry: recovery of organic solvents from waters containing electrolytes by pervaporation, *Journal of Cleaner Production* 39 (2013) 146e153
14. Ruth Barro, Jorge Regueiro, María Llompart, Carmen Garcia-Jares, Analysis of industrial contaminants in indoor air: Part 1. Volatile organic compounds, carbonyl compounds, polycyclic aromatic hydrocarbons and polychlorinated biphenyls, *Journal of Chromatography A*, 1216 (2009) 540–566
15. Jo Dewulf, Herman Van Langenhove, Anthropogenic volatile organic compounds in ambient air and natural waters: a review on recent developments of analytical methodology, performance and interpretation of field measurements, *Journal of Chromatography A*, 843 (1999) 163–177

16. Kristof Demeestere, Jo Dewulf, Bavo De Witte, Herman Van Langenhove, Sample preparation for the analysis of volatile organic compounds in air and water matrices, *Journal of Chromatography A*, 1153 (2007) 130–144
17. Martin Harper, Sorbent trapping of volatile organic compounds from air, *Journal of Chromatography A*, 885 (2000) 129–151
18. Anastasia D. Nikolaou, Spyros K. Golfinopoulos, Maria N. Kostopoulou, George A. Kolokythas, Themistokles D. Lekkas, Determination of volatile organic compounds in surface waters and treated wastewater in Greece, *Water Research* 36 (2002) 2883–2890
19. Claudia Witschi, Erik Doelker, Residual solvent in pharmaceutical products: acceptable limits, influences on physicochemical properties, analytical methods and documented values, *European Journal of Pharmaceutics and Biopharmaceutics* 43 (1997) 215–242
20. Jo Dewulf, Herman Van Langenhove, Anthropogenic volatile organic compounds in ambient air and natural waters: a review on recent developments of analytical methodology, performance and interpretation of field measurements, *Journal of Chromatography A*, 843 (1999) 163–177
21. Method EPA 8015D, Non-halogenated organics using GC-FID, *Revision 4 June 2003*
22. Method 8260C, Volatile Organic Compounds by gas chromatography / Mass Spectrometry

ACNOWLEDGMENTS

Thanks to NEOSIS srl

Thanks to Dr Gianni Maticchione

Thanks to Prof Roberta Curini

Thanks to my colleagues Dr Giuseppe De Martino and Mrs Daniela Steffani

A special thanks to my Family

Enrico Mattocci