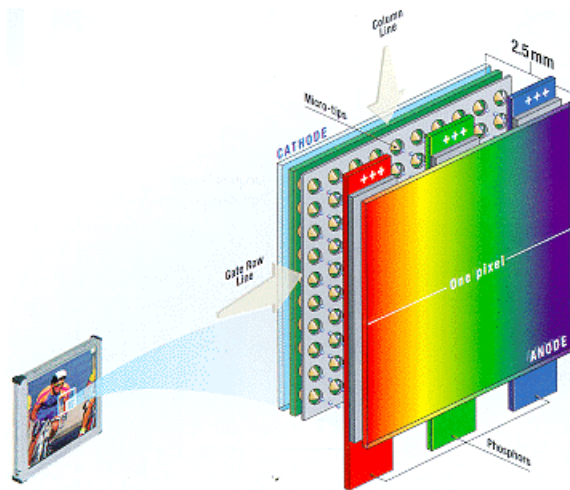


Symbol	Name	Formula	Value
a_{C-C}	carbon-carbon distance		1.421 Å (graphite)
a	length of unit vector	$\sqrt{3}a_{C-C}$	2.46 Å
a_1, a_2	unit vectors	$\left(\frac{\sqrt{3}}{2}, \frac{1}{2}\right)a, \left(\frac{\sqrt{3}}{2}, -\frac{1}{2}\right)a$	in (x, y) coordinates
b_1, b_2	reciprocal lattice vectors	$\left(\frac{1}{\sqrt{3}}, 1\right)\frac{2\pi}{a}, \left(\frac{1}{\sqrt{3}}, -1\right)\frac{2\pi}{a}$	in (x, y) coordinates
C_h	chiral vector	$C_h = na_1 + ma_2 \equiv (n, m)$	n, m : integers
L	circumference of nanotube	$L = C_h = a\sqrt{n^2 + m^2 + nm}$	$0 \leq m \leq n$
d_t	diameter of nanotube	$d_t = \frac{L}{\pi} = \frac{\sqrt{n^2 + m^2 + nm}}{\pi}a$	
θ	chiral angle	$\sin \theta = \frac{\sqrt{3}m}{2\sqrt{n^2 + m^2 + nm}}$ $\cos \theta = \frac{2n + m}{2\sqrt{n^2 + m^2 + nm}}$ $\tan \theta = \frac{\sqrt{3}m}{2n + m}$	$0 \leq \theta \leq 30^\circ$
d	the highest common divisor of (n, m)		
d_R	the highest common divisor of $(2n+m, 2m+n)$	$d_R = \begin{cases} d & \text{if } n-m \text{ not a multiple of } 3d \\ 3d & \text{if } n-m \text{ a multiple of } 3d \end{cases}$	
T	translation vector of 1D unit cell	$T = t_1 a_1 + t_2 a_2$ $t_1 = \frac{2m+n}{d_R}$ $t_2 = -\frac{2n+m}{d_R}$	t_1, t_2 : integers
T	length of T	$T = \frac{\sqrt{3}L}{d_R}$	
N	number of hexagons per 1D unit cell	$N = \frac{2(n^2 + m^2 + nm)}{d_R}$	$2N \equiv n_C/\text{unit cell}$
R	symmetry vector [†]	$R = pa_1 + qa_2 \equiv (p, q)$ $d = mp - nq, 0 \leq p \leq n/d, 0 \leq q \leq m/d$	p, q : integers [†]
M	number of 2π revolutions	$M = [(n2 + m)p + (2m + n)q]/d_R$	M : integer
R	Basic symmetry operation [‡]	$NR = MC_h + dT$	
ψ	rotation operation	$R = \frac{\psi}{\tau}$	ψ : radians
τ	translation operation	$\tau = \frac{dT}{N}$	τ, χ : length

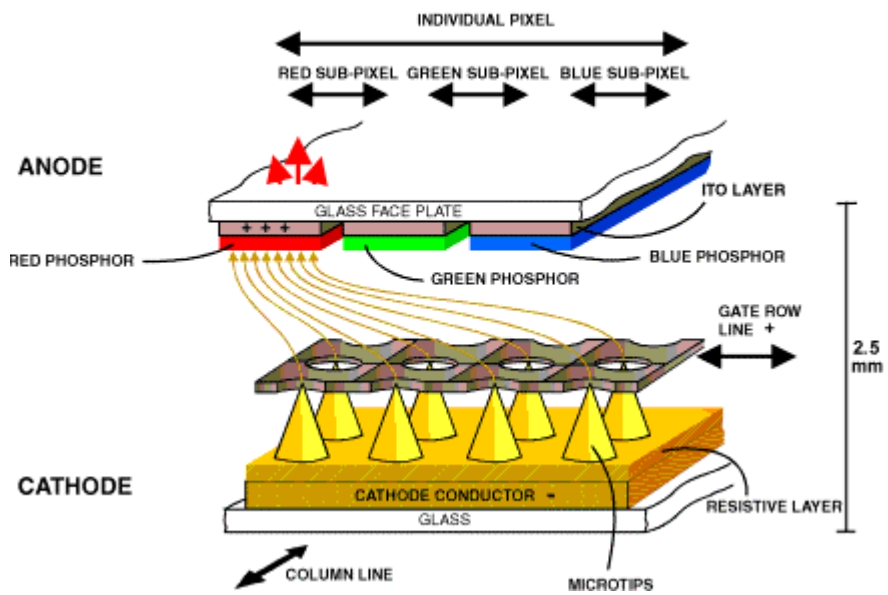
[†] (p, q) are uniquely determined by $d = mp - nq$, subject to conditions stated in table, except for zigzag tubes which $C_h = (n, 0)$, and we define $p = 1, q = -1$, which gives $M = 1$.

[‡] R and R refer to the same symmetry operation.

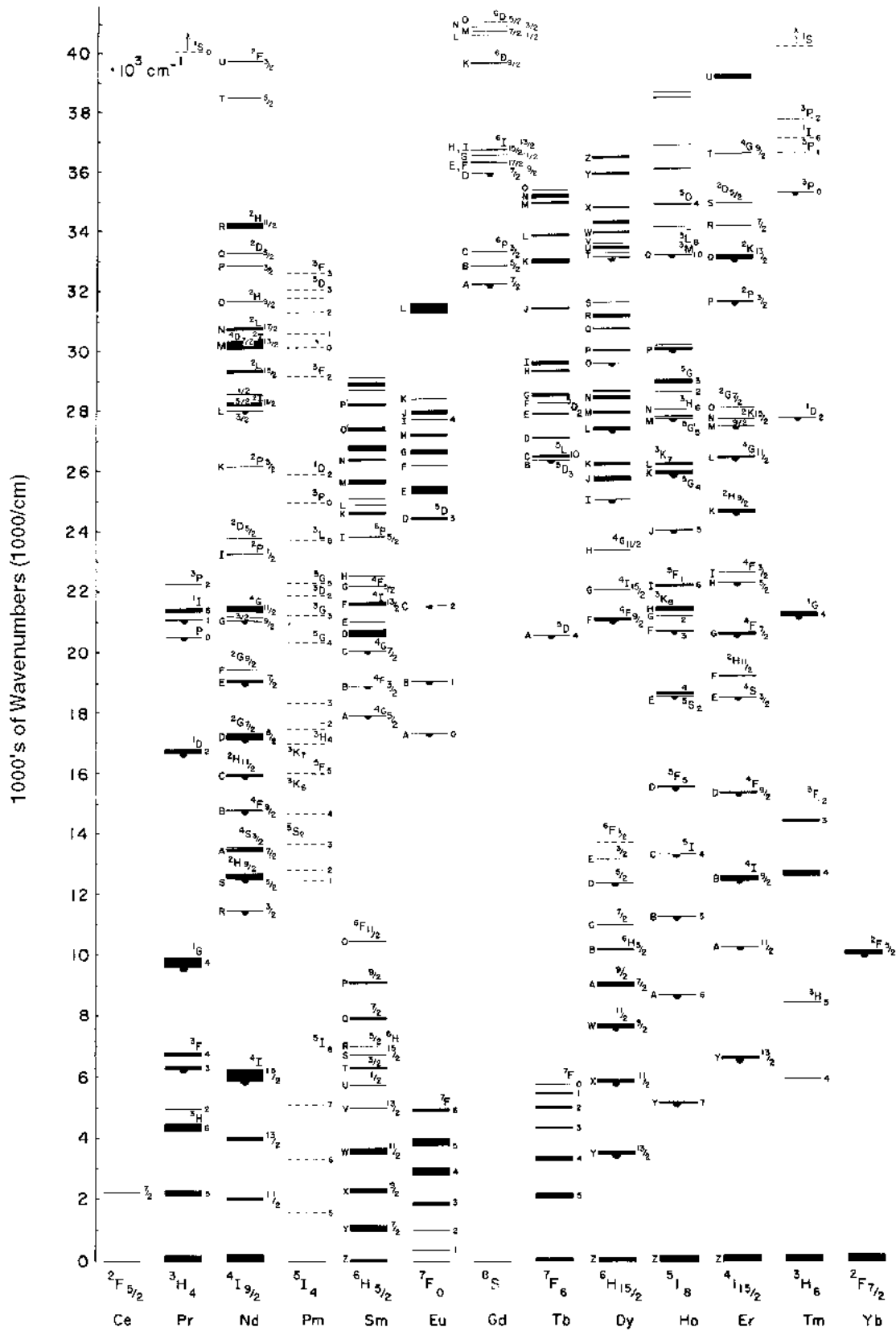
Tab. 2



Field Emission Display



Classic 'Dieke Diagram' for Rare Earth Ions



G. H. Dieke. Spectra and energy levels of rare earth ions in crystals (Interscience Publishers, New York, 1968)

Fig. 17

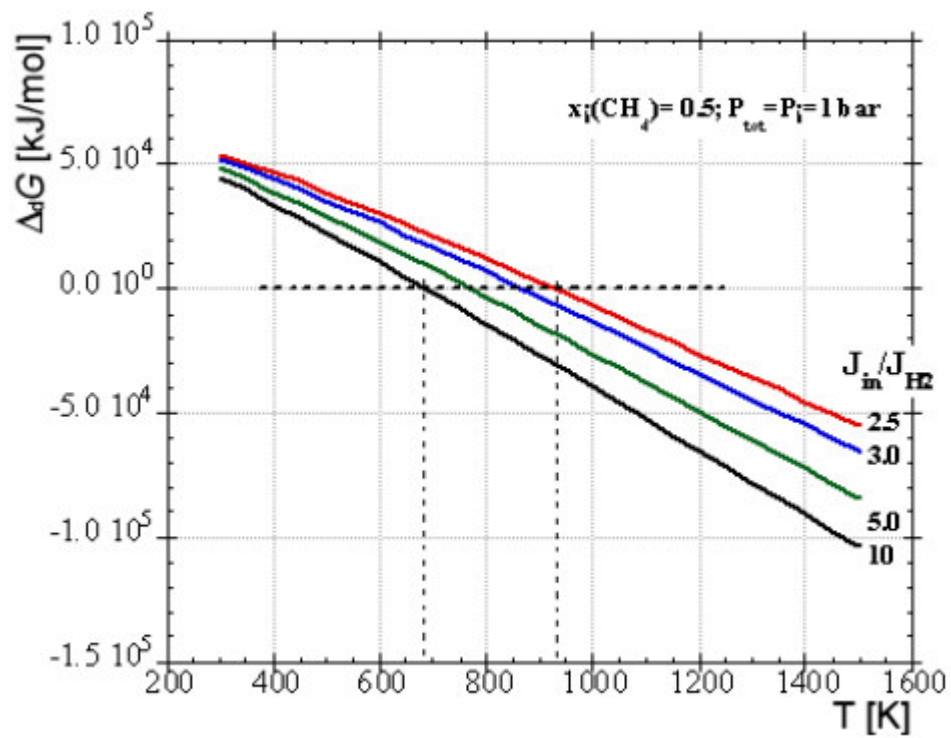
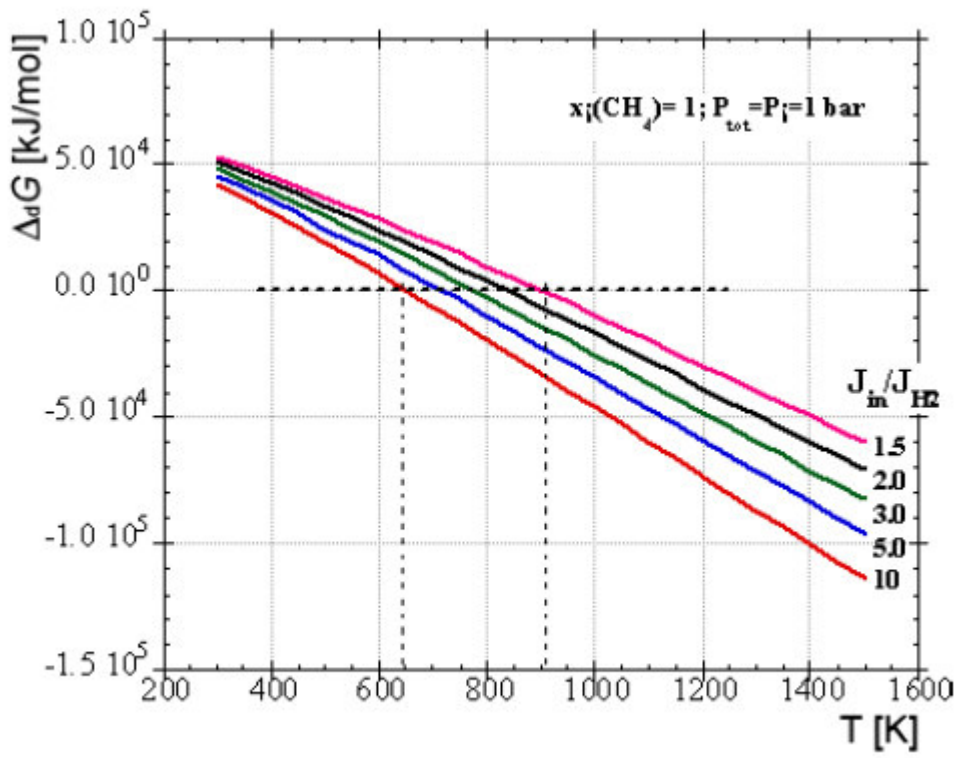


Fig. 24-25

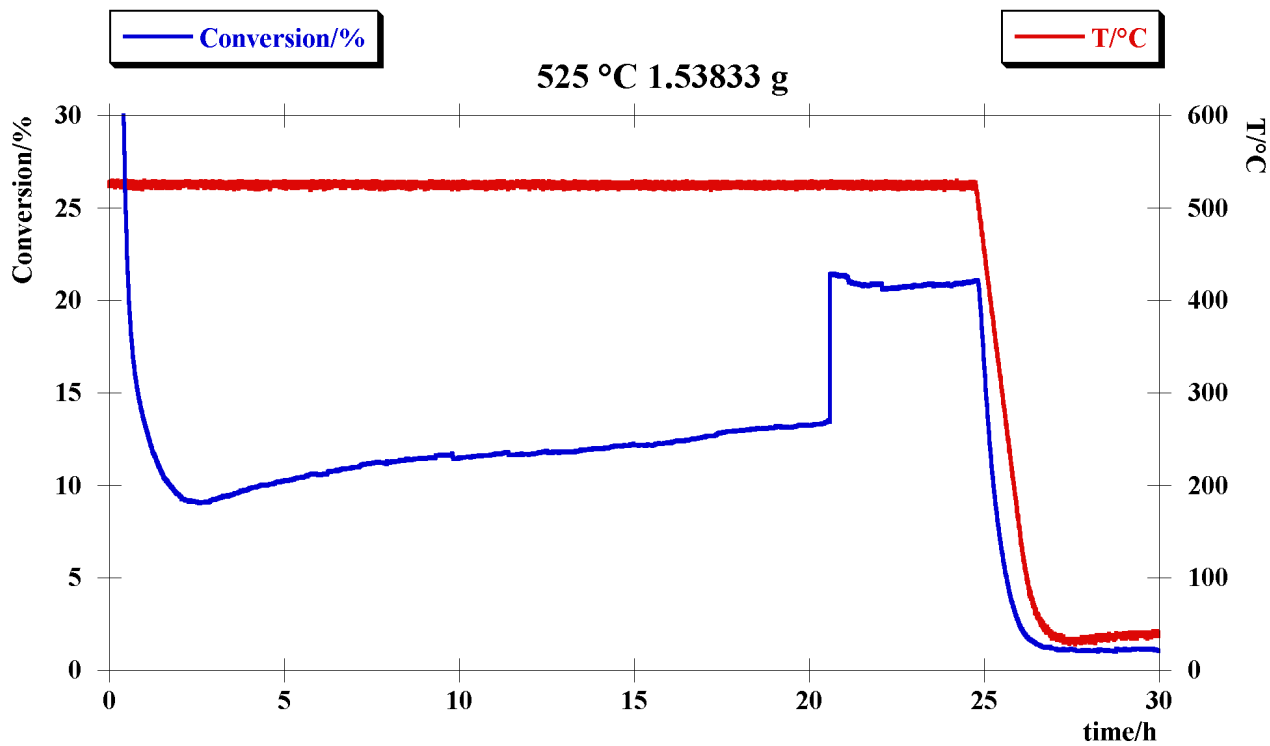


Fig. 41

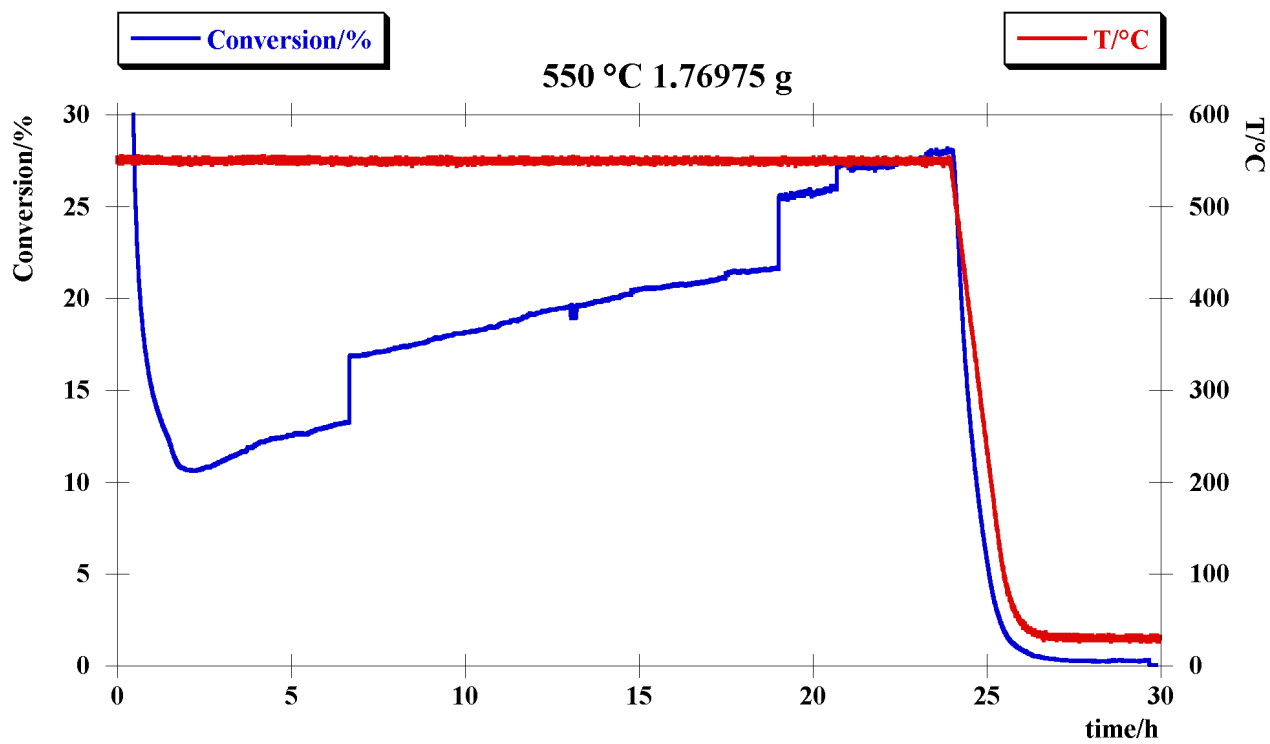


Fig. 42

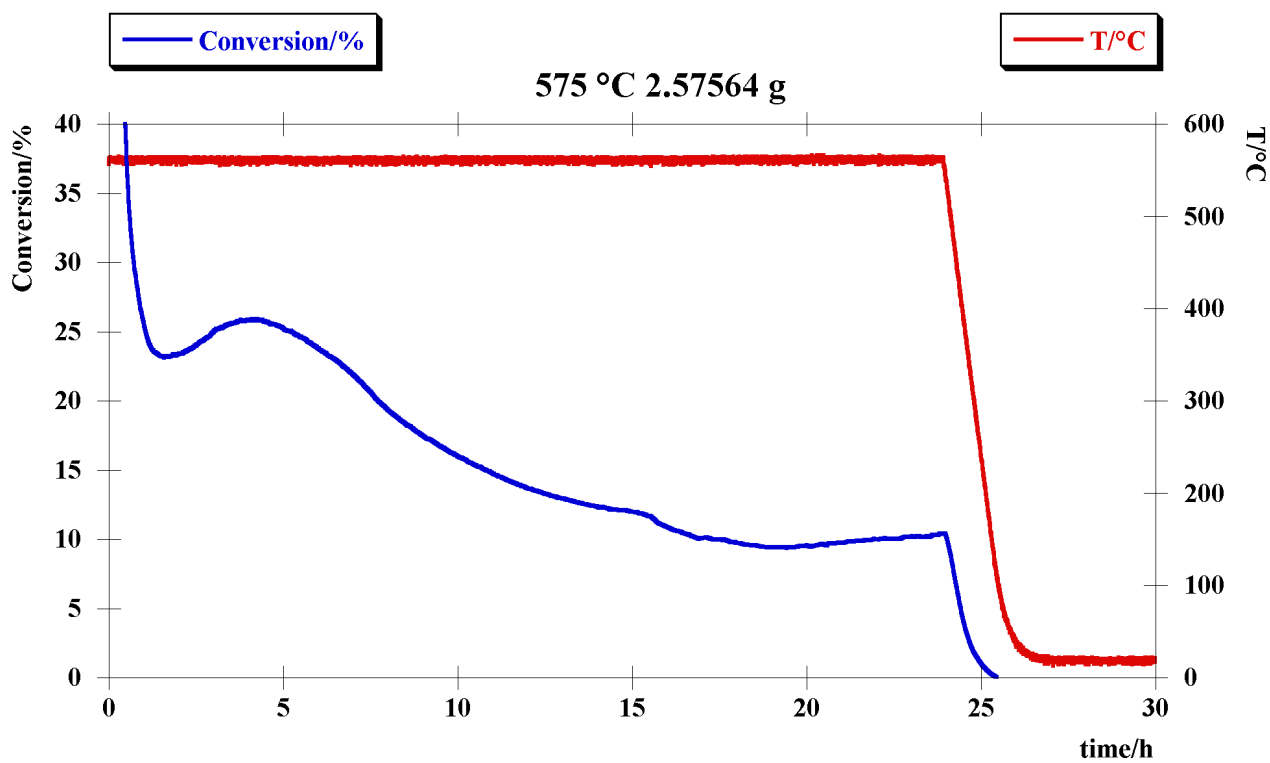


Fig. 43

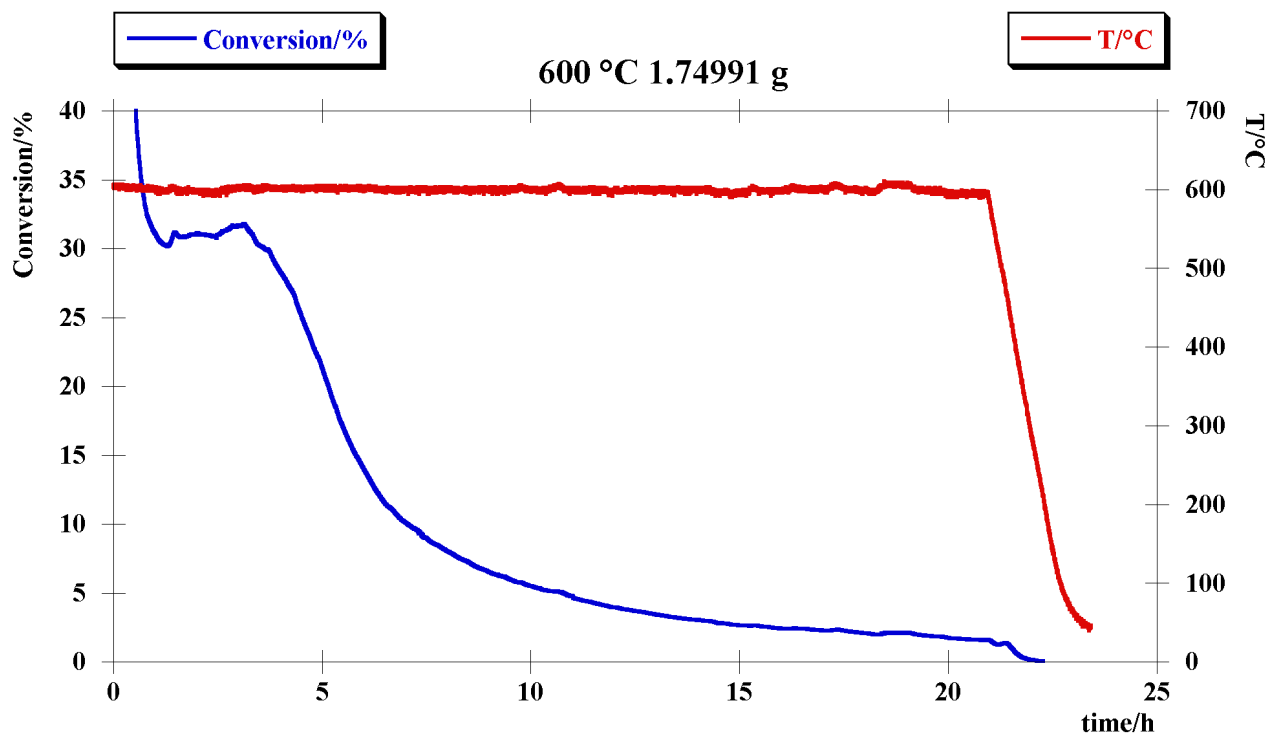


Fig. 44

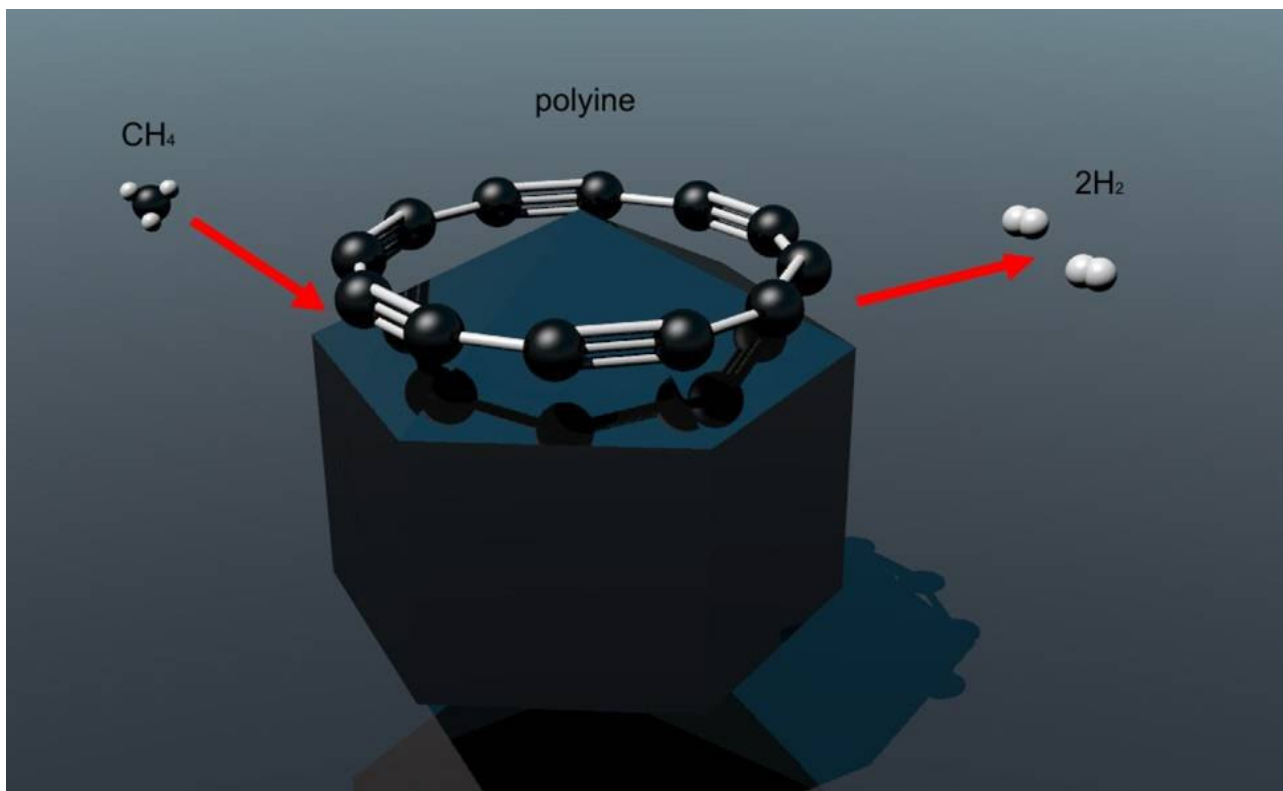


Fig. 65

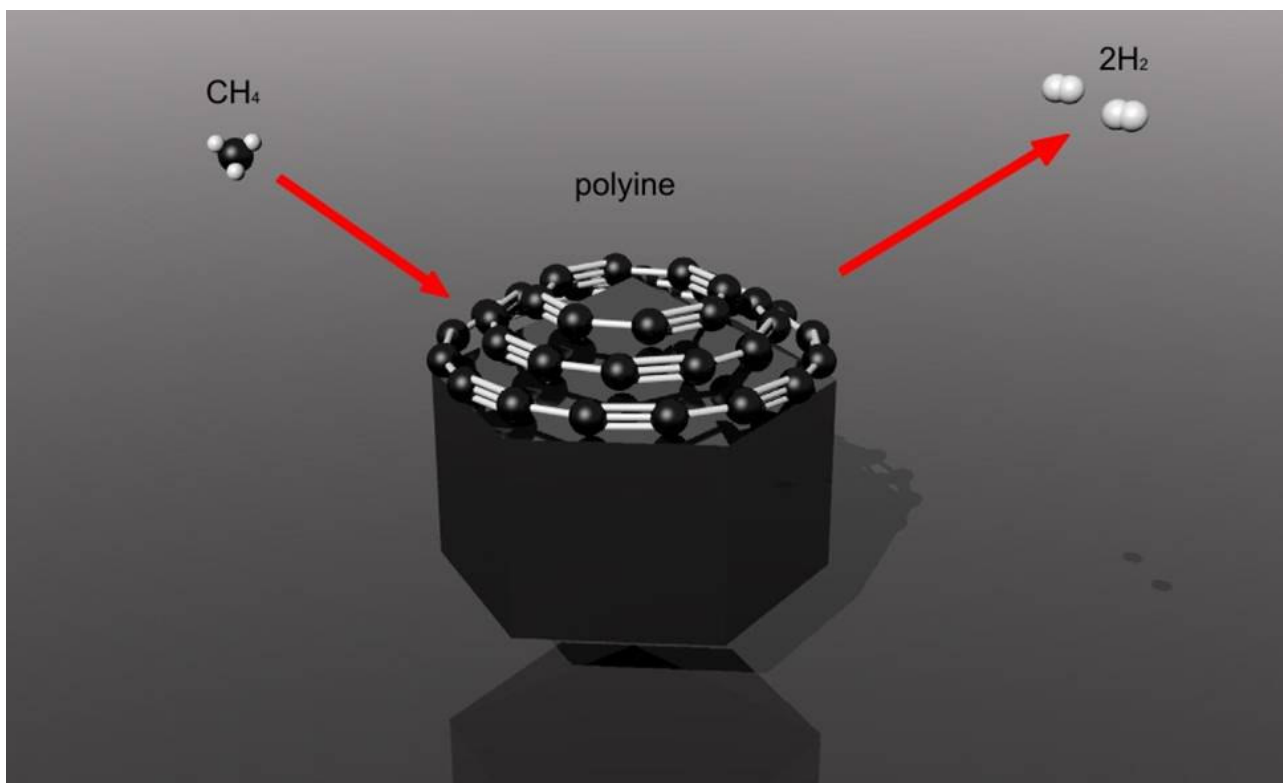


Fig. 66

475 °C:

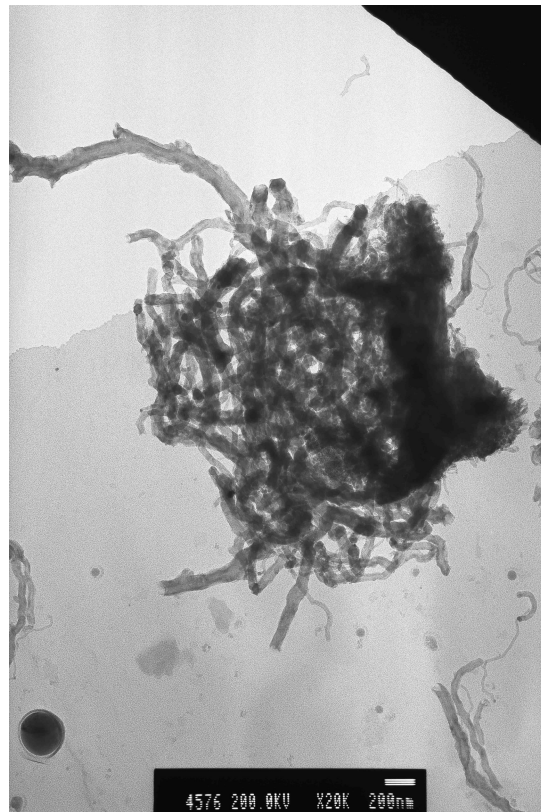


Fig. 68

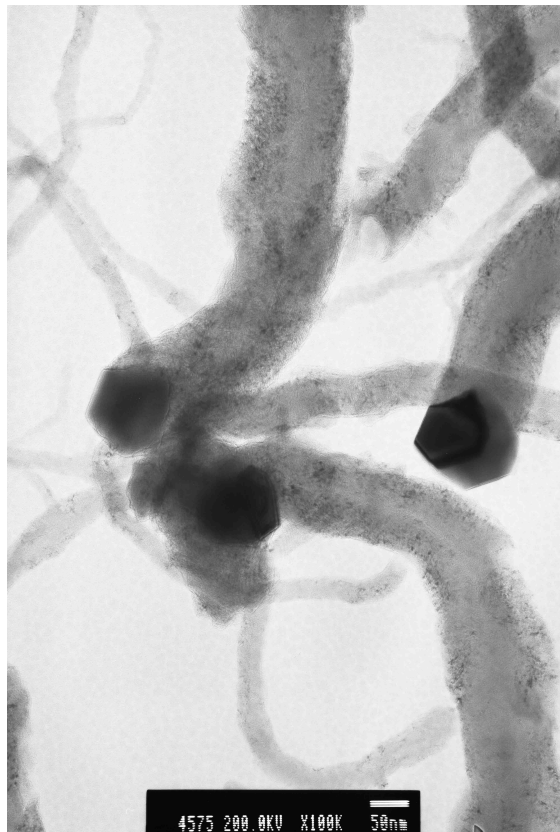


Fig. 69

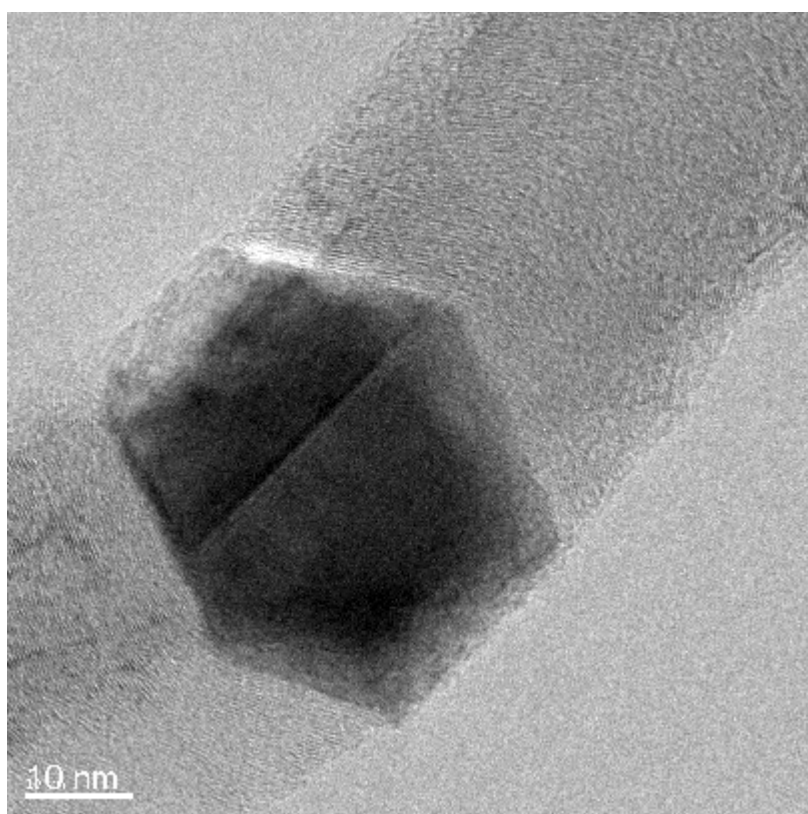


Fig. 70

Metal nanoparticle with two attached CNTs. A stacking fault in the nanoparticle is clearly visible.

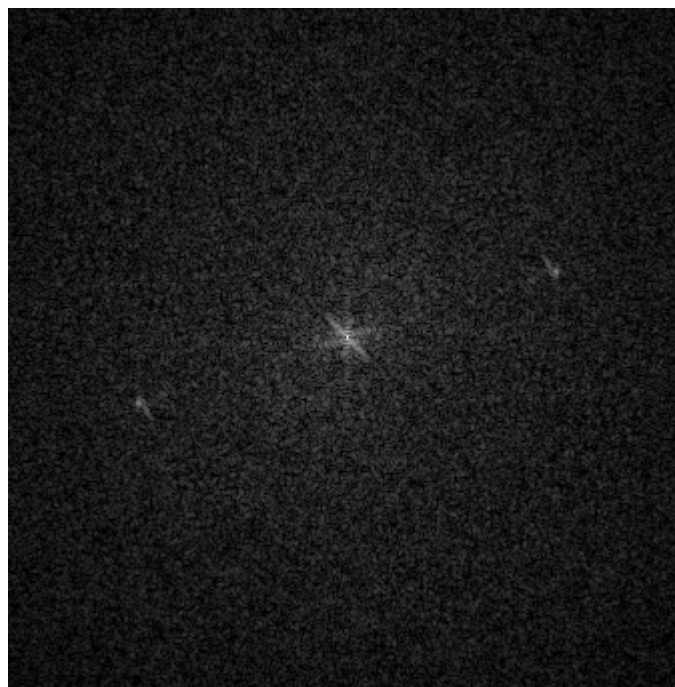


Fig. 71

Fourier transform of the image of the metal nanoparticle in fig. 70.

500 °C:

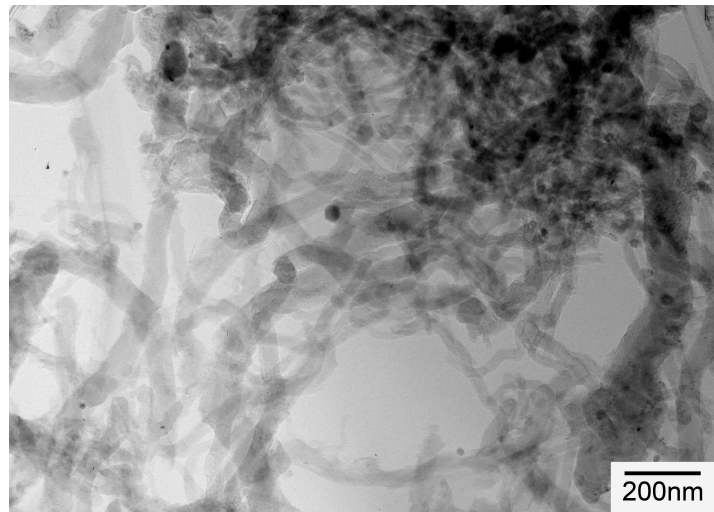


Fig. 72

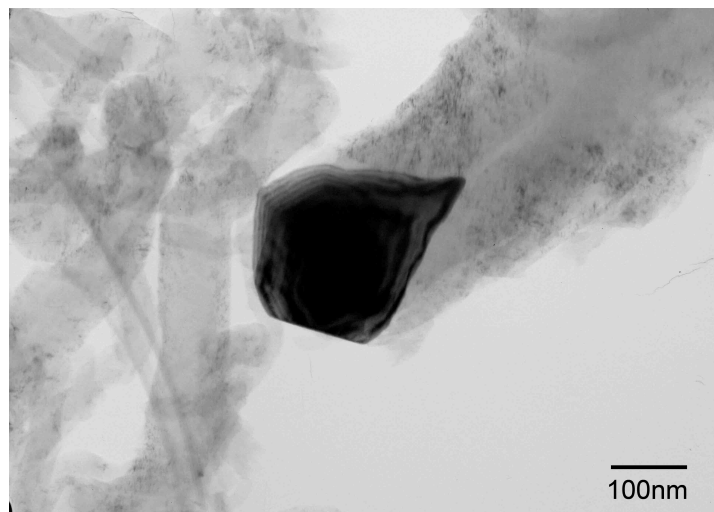


Fig. 73

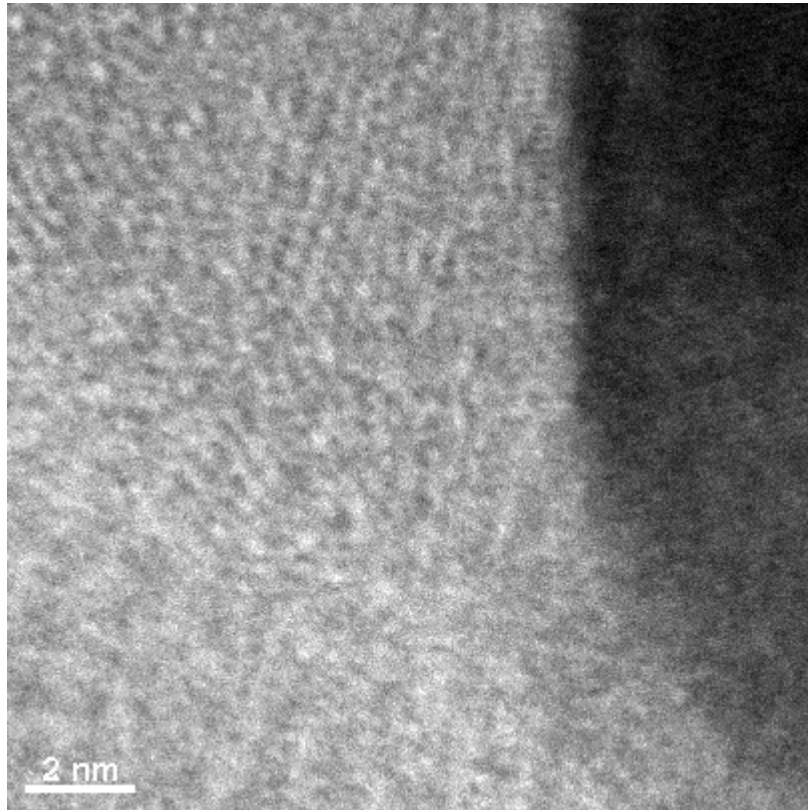


Fig. 74

Fig. 73 metal nanoparticle-CNT interface.

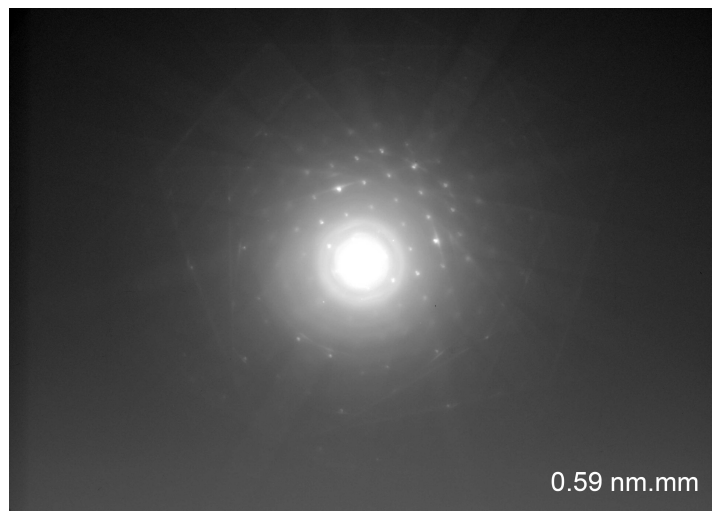


Fig. 75

Electron diffraction pattern of the metal nanoparticle shown in fig. 73.

525 °C:

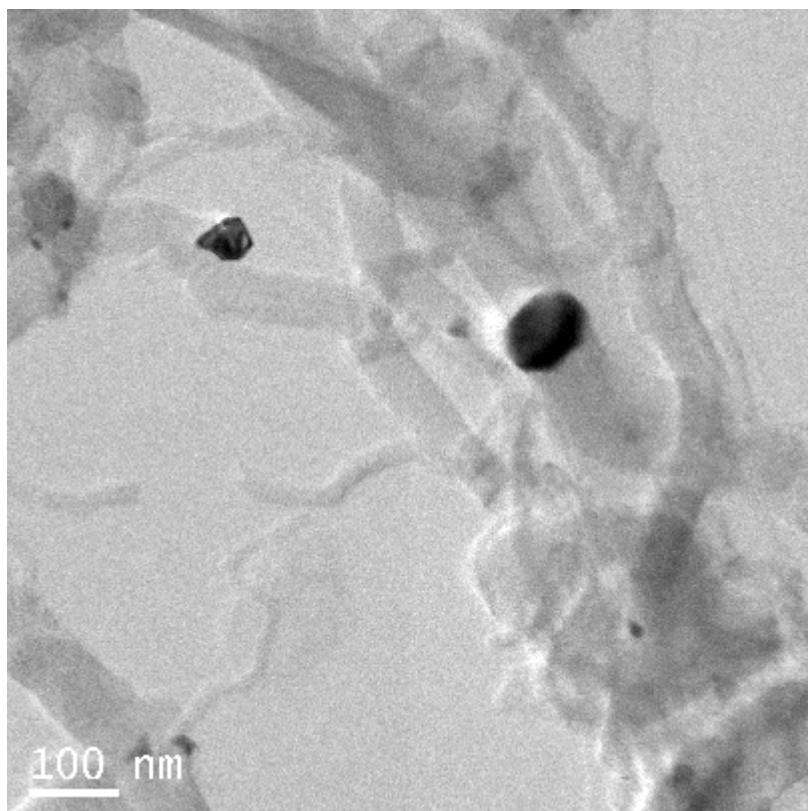


Fig. 76

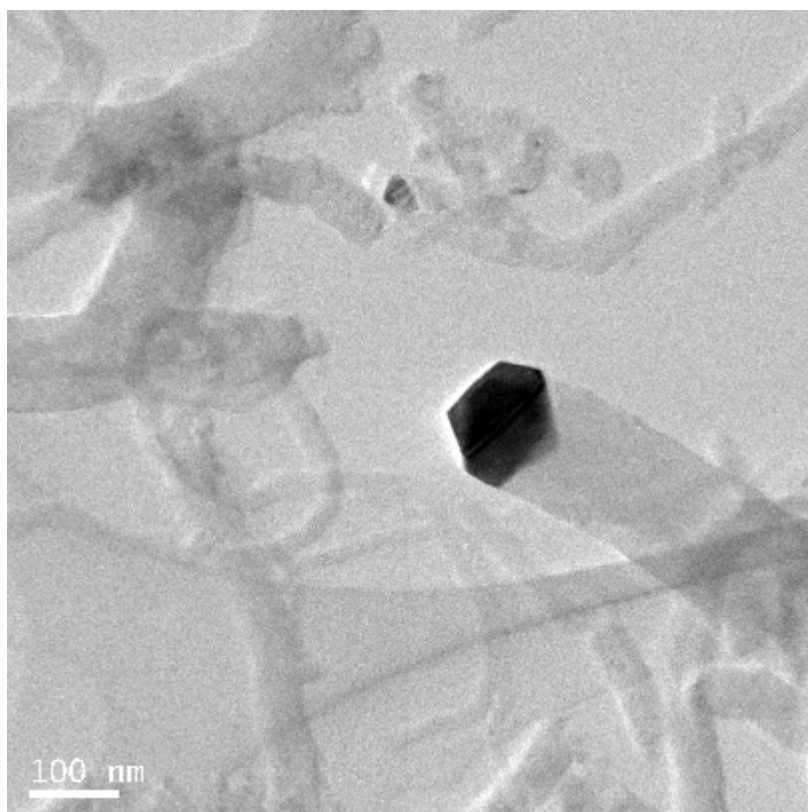


Fig. 77

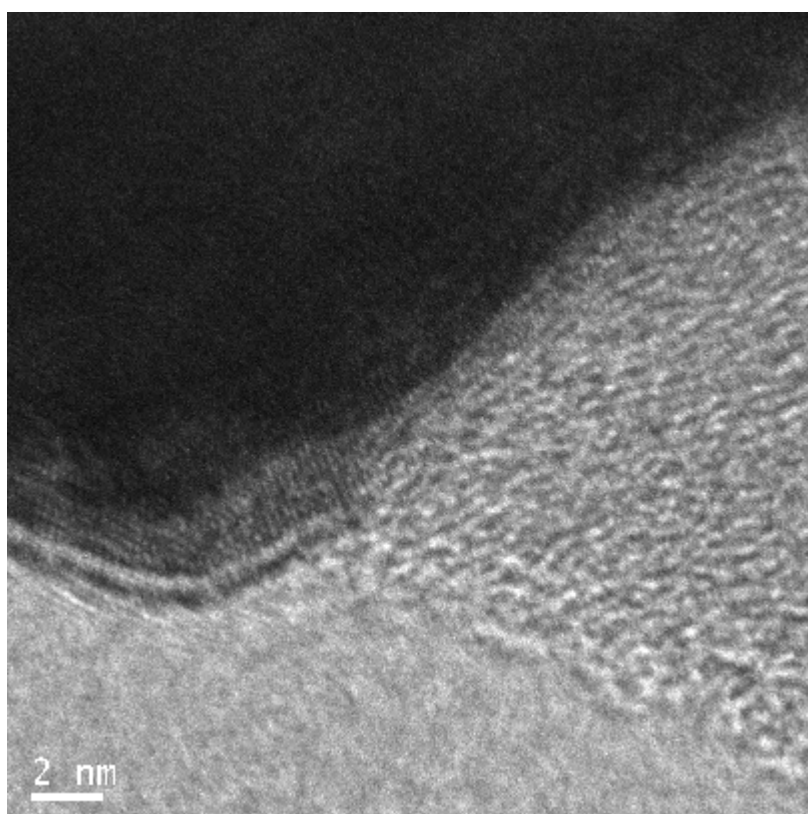


Fig. 78

Fig. 77 metal nanoparticle-CNT interface.

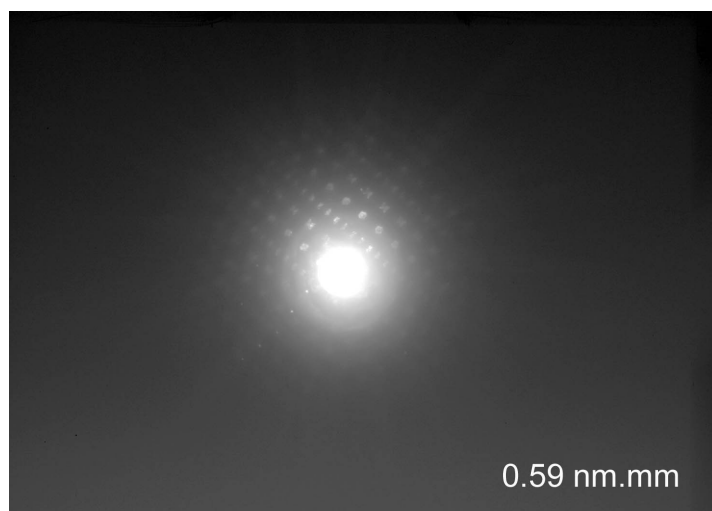


Fig. 79

Electron diffraction pattern of the metal nanoparticle shown in fig. 77.

550 °C:

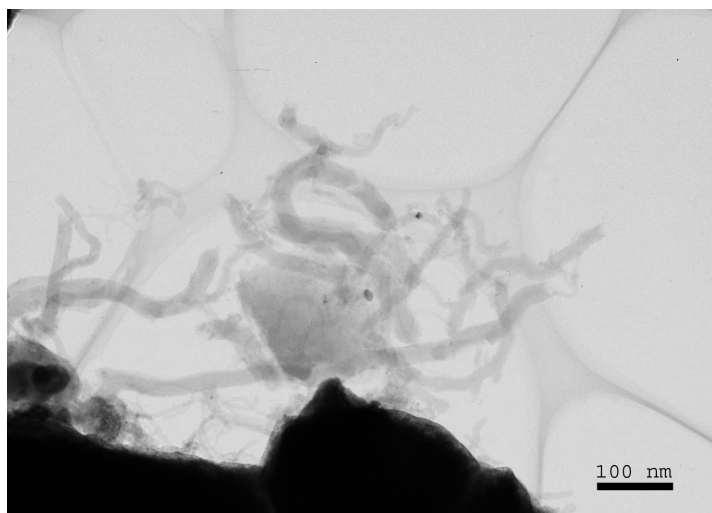


Fig. 80

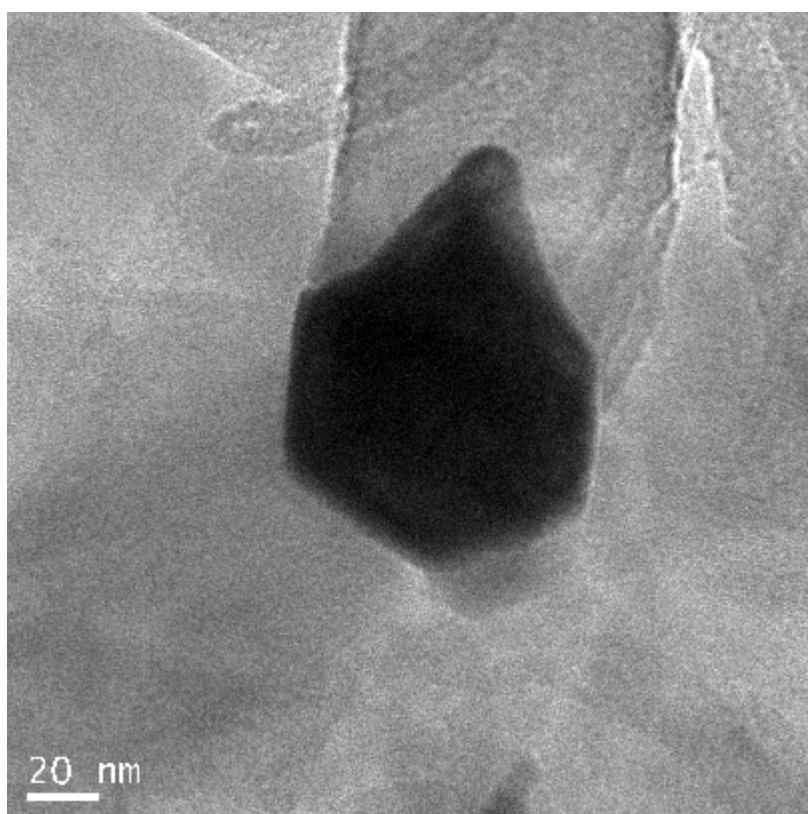


Fig. 81

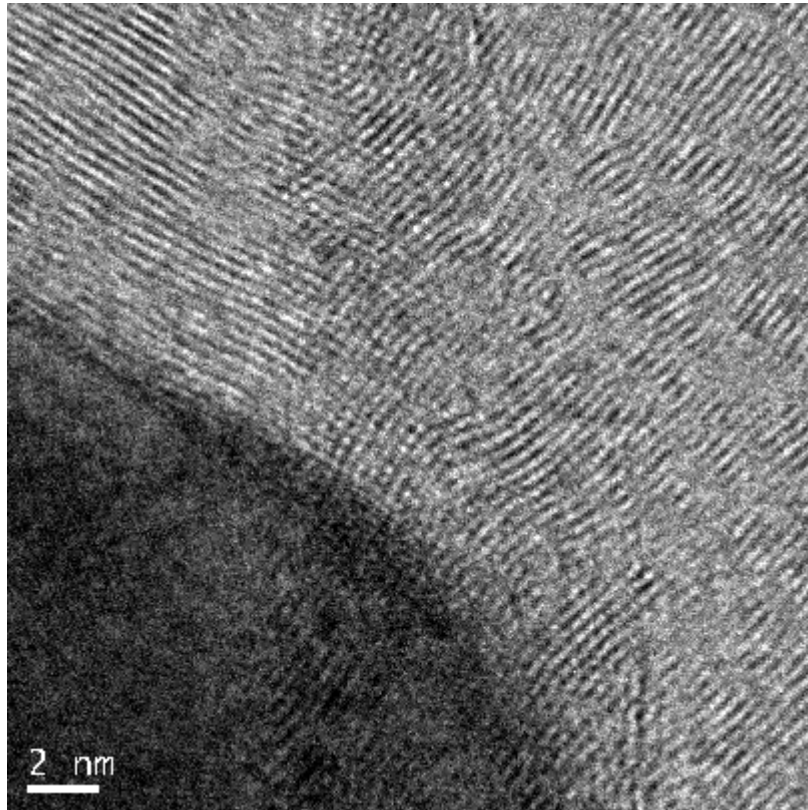


Fig. 82

Fig. 81 metal nanoparticle-CNT interface.

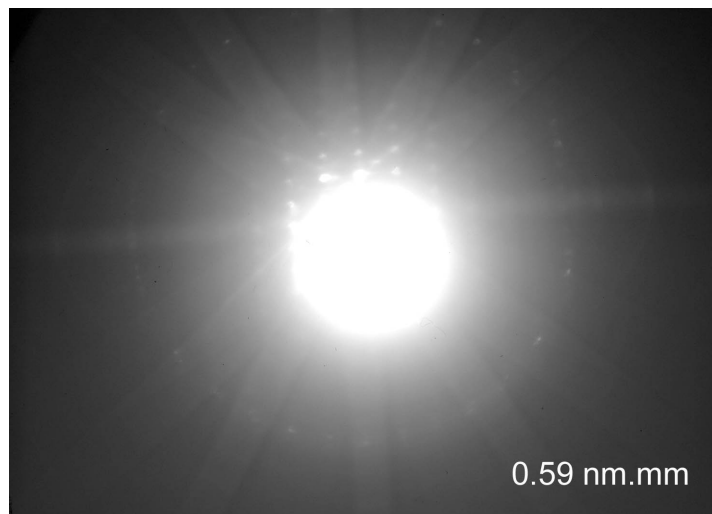


Fig. 83

Electron diffraction pattern of the metal nanoparticle shown in fig. 77.

575 °C:

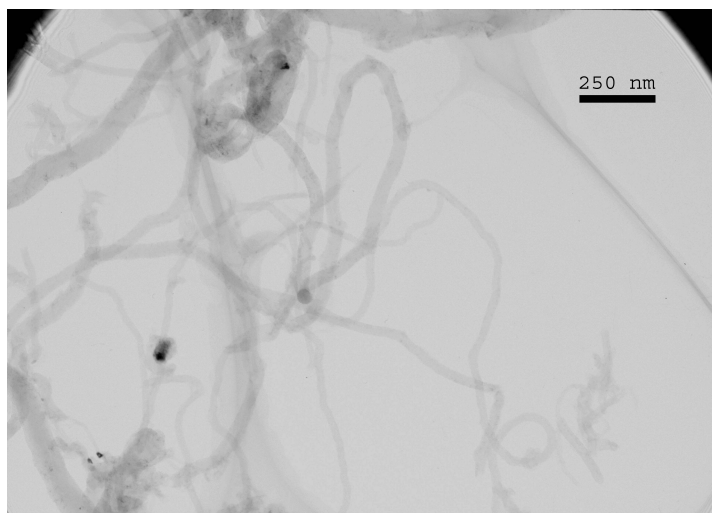


Fig. 84

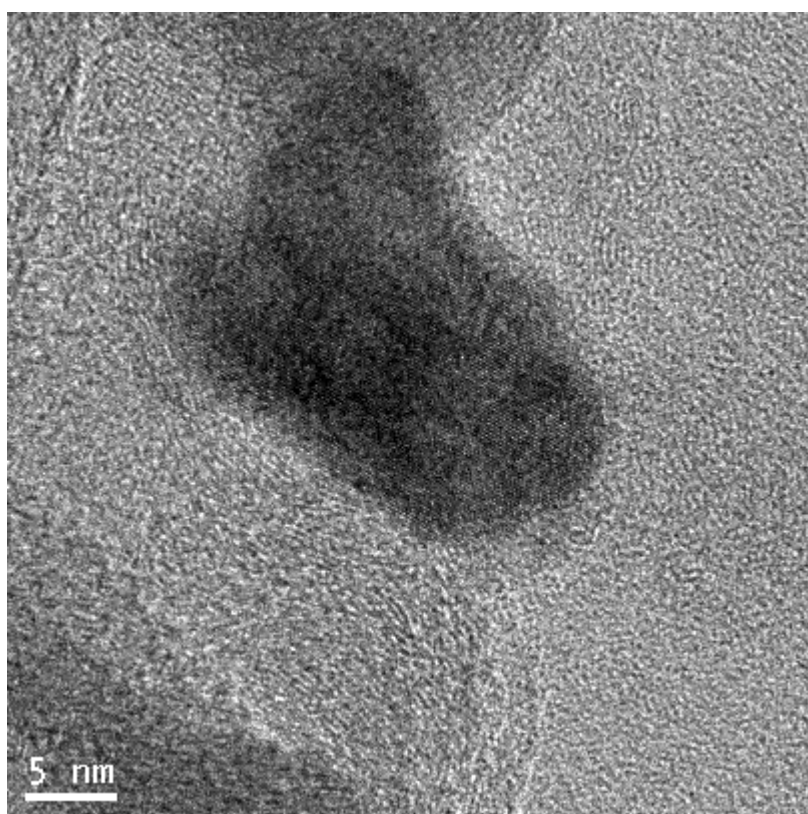


Fig. 85